

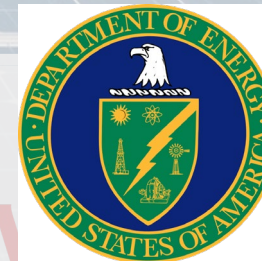
# The Sun Is Ready to Make Your Electricity Greener and Cheaper in Wisconsin

**Michael S. Arnold**  
Dept. of Materials Science  
& Engineering  
[michael.arnold@wisc.edu](mailto:michael.arnold@wisc.edu)

Arnold group research on  
solar & electronic  
materials funded by:



National  
Science  
Foundation



Department  
of Energy



Air Force  
Office of  
Scientific  
Research



Army  
Research  
Office

Wed. Nite @ the Lab, January 13, 2010

## Advanced Materials for Energy and Electronics Group

@ UW-Madison

**Why doesn't my electricity  
come from the sun?  
Future materials for  
harnessing solar energy**

**Michael S. Arnold**

**Michael S. Arnold**, *Assistant Professor*  
msarnold@wisc.edu; (608) 262-3863

Materials Science and Engineering, UW-Madison  
248 MS&E Building, 1509 University Ave.



# The Sun Is Ready to Make Your Electricity Greener and Cheaper in Wisconsin

**Prof. Michael S. Arnold**  
Dept. of Materials Science  
& Engineering  
[michael.arnold@wisc.edu](mailto:michael.arnold@wisc.edu)

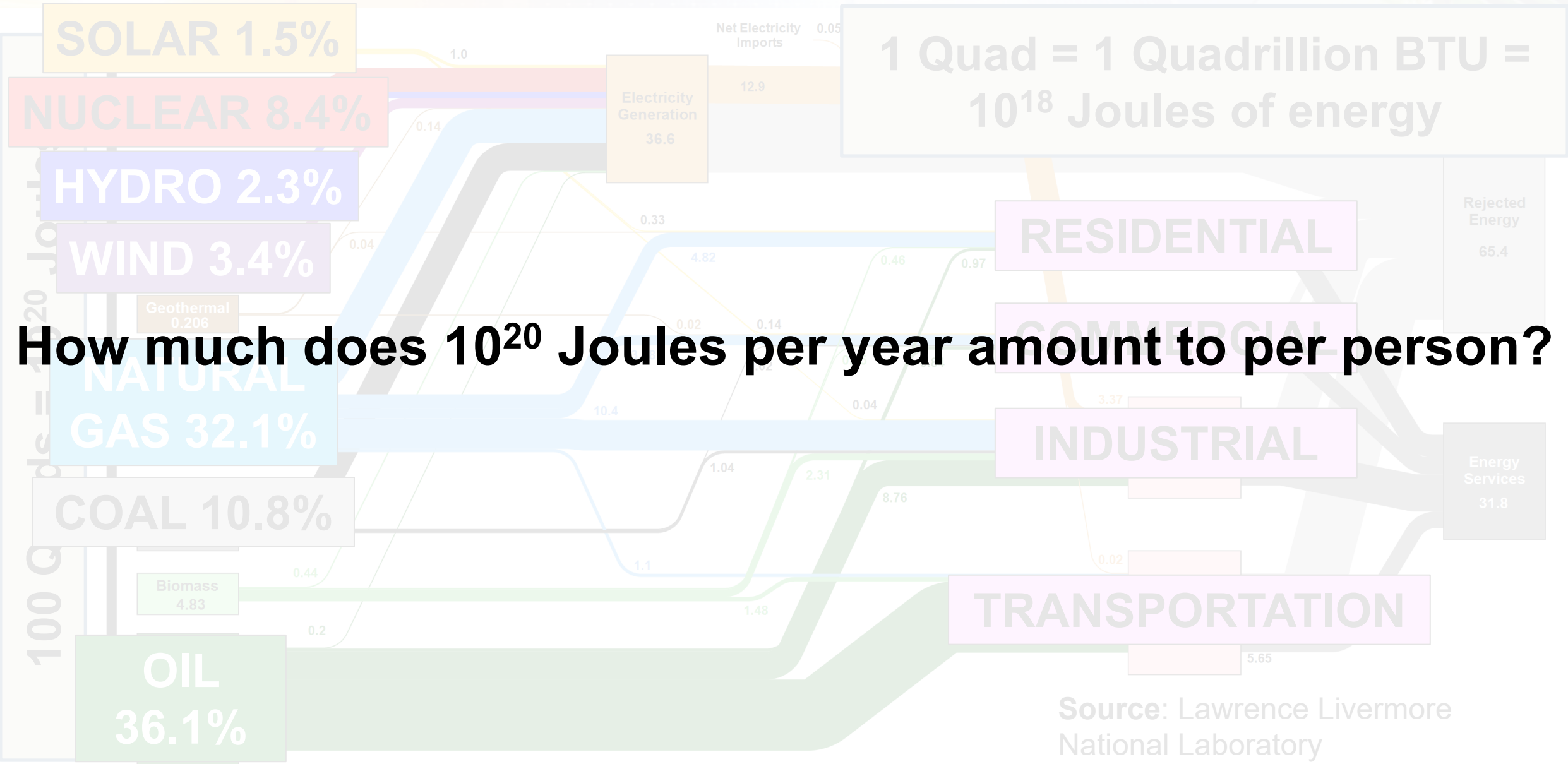


**WISCONSIN**  
UNIVERSITY OF WISCONSIN-MADISON

- ➔ **The sun as a source of energy**
- How does a photovoltaic solar cell work?
- Types of solar cells and their efficiencies
- Economics of solar cells then and now
- Solar in Wisconsin
- The future



# 2019 U.S. energy production (Quads)



Source: Lawrence Livermore National Laboratory

# Energy consumption per person

■ Per person =  $3 \times 10^{11}$  **Joules per year per person**

**QUESTION:** When you curl a 20 lb. dumb-bell and lift it about 3 feet, how much energy does it gain?

**ANSWER:** About 100 Joules.

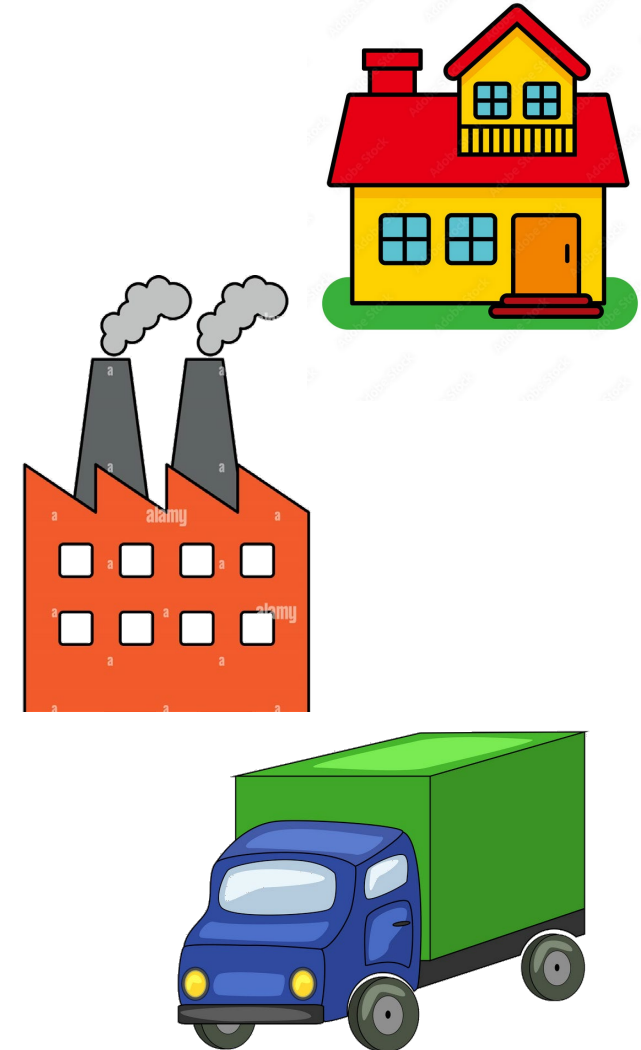


# Your energy consumption



100 Joules

300,000,000,000 Joules



# Your energy consumption

- You would need to lift this 20 lb. dumb-bell **3 billion times** a year in order to supply your own energy (assuming your efforts could be harnessed with 100% efficiency).



**340,000 times  
an hour**



Power = energy per time

■ U.S. Energy consumption

=  $10^{20}$  Joules per year

=  $10^{20}$  Joules per 31.6 million seconds

=  $3 \times 10^{12}$  Joules per second

= 3 Trillion Watts

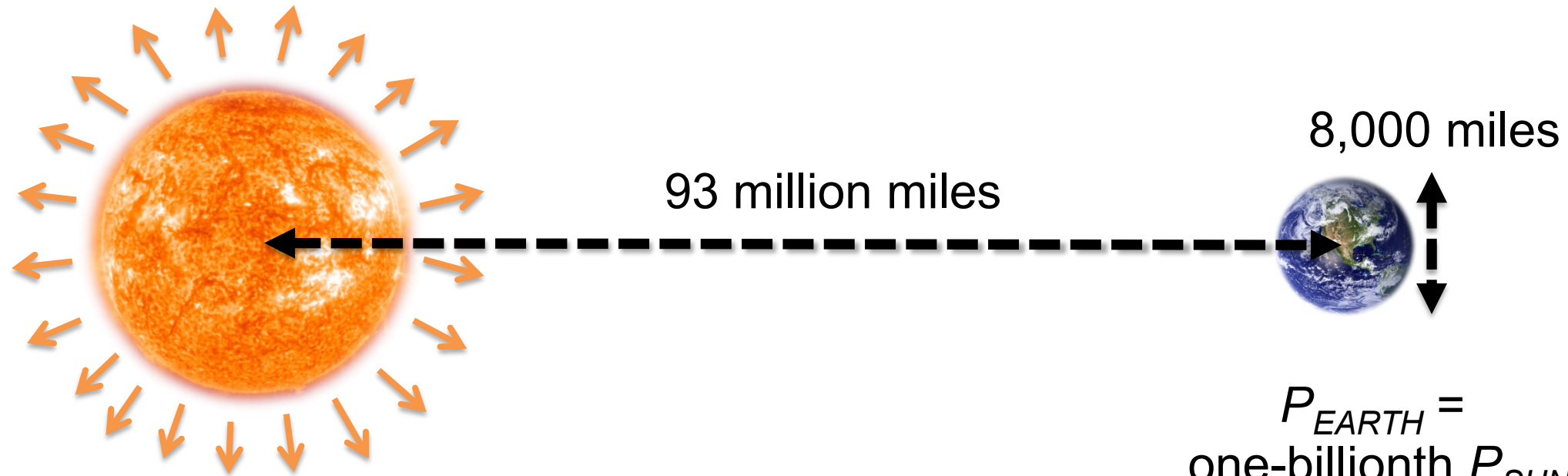
= **3 TW** (3,000,000,000,000 Watts)

■ Worldwide

= **18 TW** (18,000,000,000,000 Watts)

**Is the sun a viable source  
for supplying humankind with  
18 TW of power?**

Yes. The energy from the sun is enormous.



$$T = 6000 \text{ K}$$

$$P_{SUN} = 3.9 \times 10^{26} \text{ Watts}$$

$$P_{EARTH} =$$

$$\text{one-billionth } P_{SUN}$$

$$= 3 \times 10^{17} \text{ Watts}$$

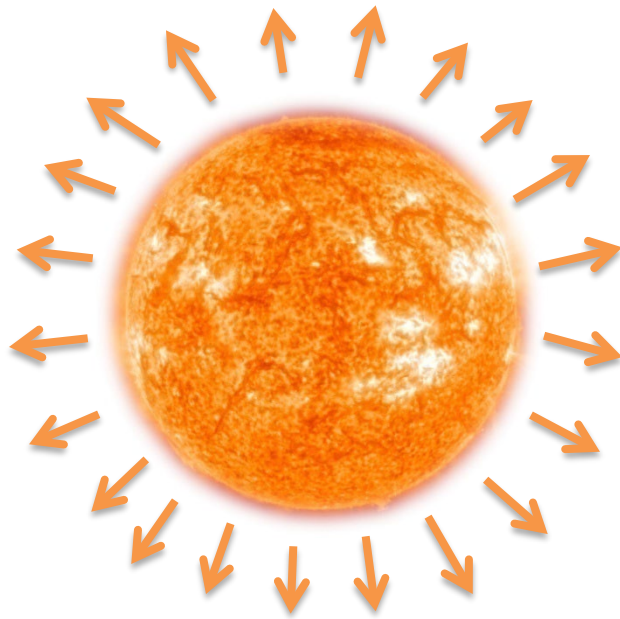
$$= 300,000,000,000,000,000 \text{ Watts}$$

**Global needs = 18,000,000,000,000 Watts**

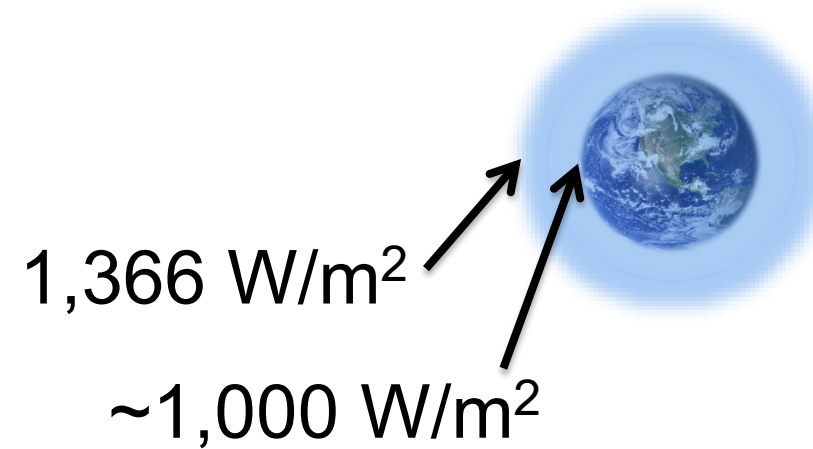
Image of sun:

<http://nssdc.gsfc.nasa.gov/planetary/factsheet/sunfact.html>

*~1000 Watts per meter-by-meter area on sunny day*



Peak intensities:



# Averaging over the seasons and day and night?

- **Geography and climate-dependent:** considering clouds, four seasons, day and night

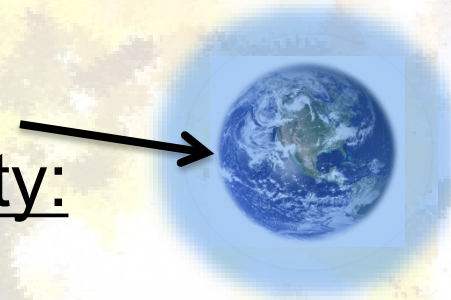


Peak intensity:

$\sim 1,000 \text{ W/m}^2$

Average intensity:

$\sim 100\text{-}300 \text{ W/m}^2$



# Map of average intensity in U.S.



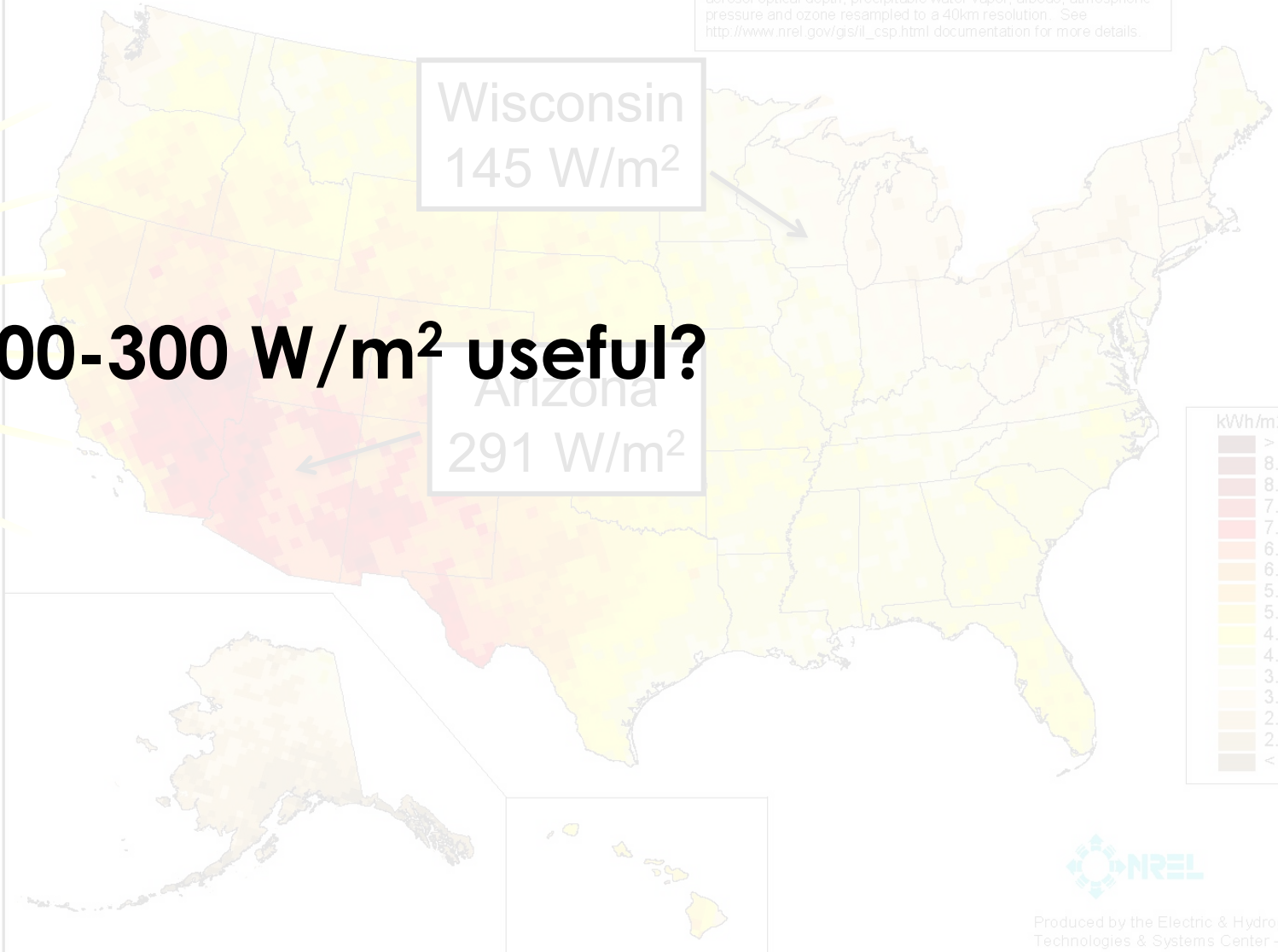
$P_{SUN} = 3.9 \times 10^{26}$   
Watts

**Is 100-300 W/m<sup>2</sup> useful?**

Direct Normal Solar Radiation  
(Two-Axis Tracking Concentrator)

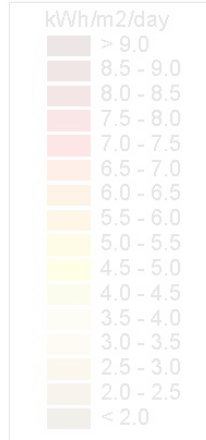
Annual

Model estimates of monthly average daily total radiation using inputs derived from satellite and/or surface observations of cloud cover, aerosol optical depth, precipitable water vapor, albedo, atmospheric pressure and ozone resampled to a 40km resolution. See [http://www.nrel.gov/gis/il\\_csp.html](http://www.nrel.gov/gis/il_csp.html) documentation for more details.



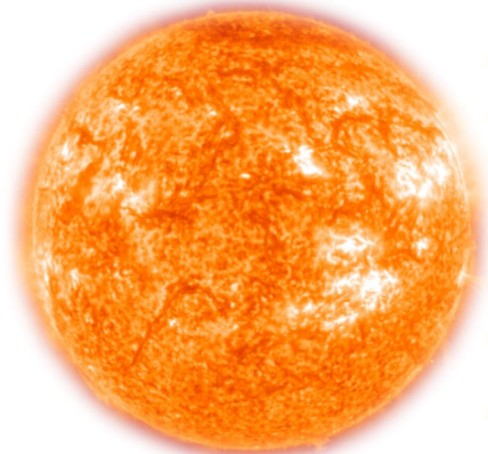
Wisconsin  
145 W/m<sup>2</sup>

Arizona  
291 W/m<sup>2</sup>

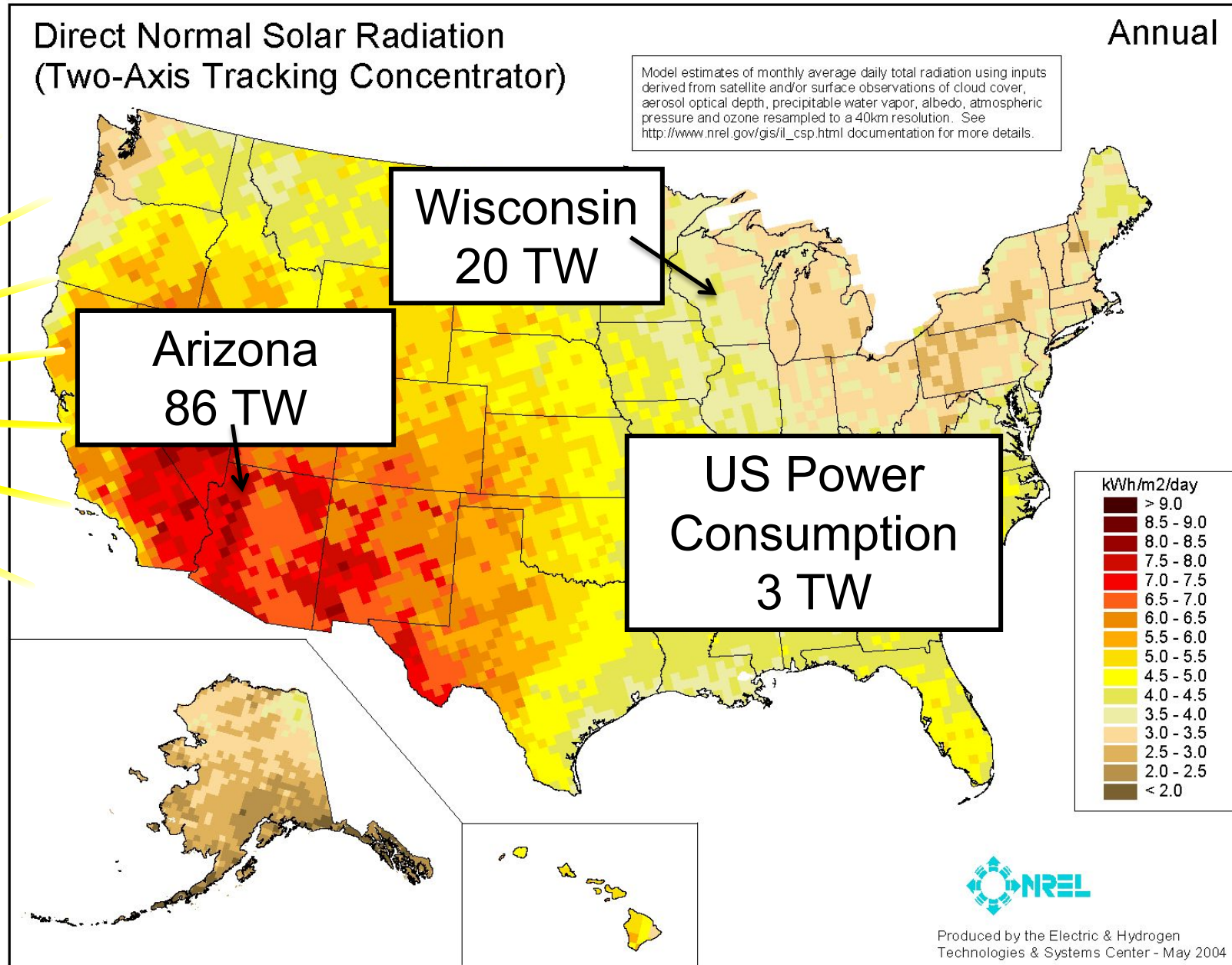


Produced by the Electric & Hydrogen Technologies & Systems Center - May 2004

# Yes, when summed over the entire state.



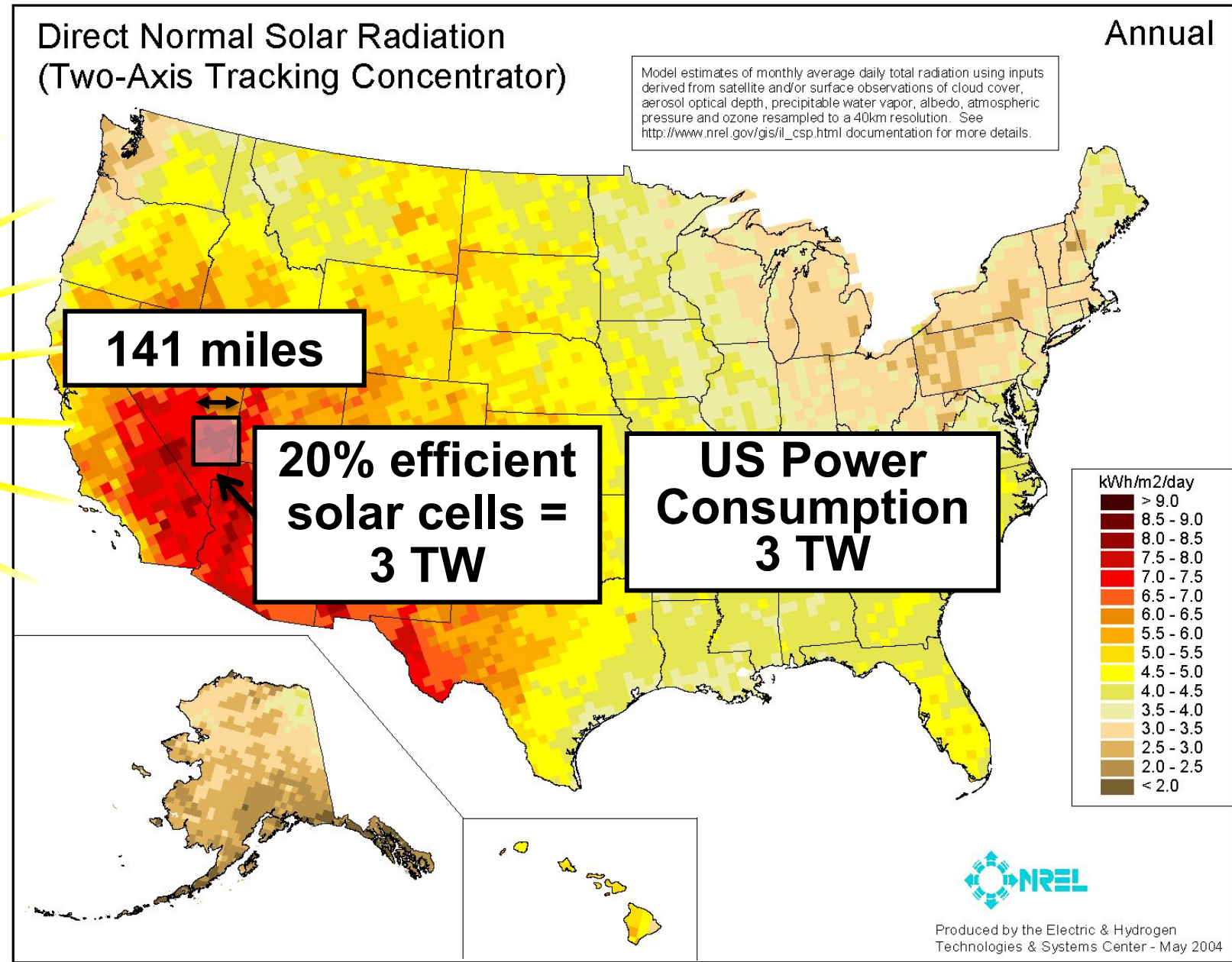
$P_{SUN} = 3.9 \times 10^{26}$   
Watts



# Small fraction of land-area needed to cover energy needs.



$P_{SUN} = 3.9 \times 10^{26}$   
Watts



# Conclusions about solar resources

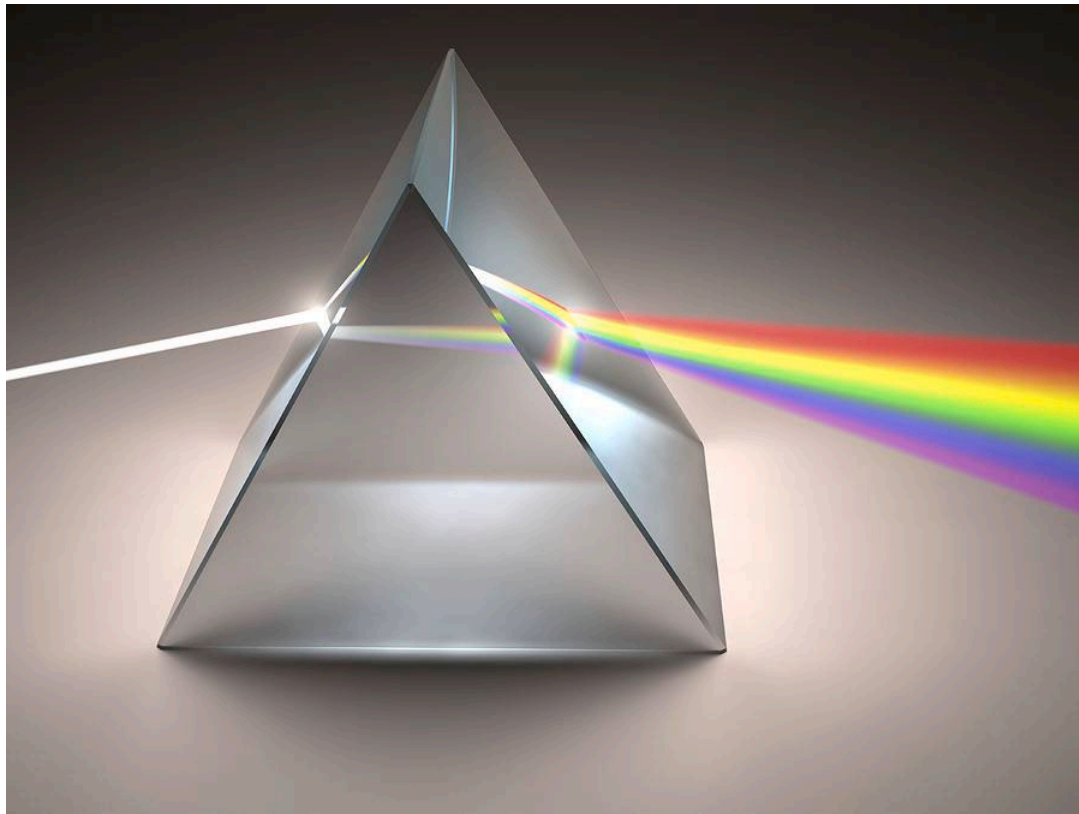
- Humankind's energy needs are enormous.
- The energy we receive from the sun is more enormous.
- Sunlight could power our society if the proper technology were available and affordable.



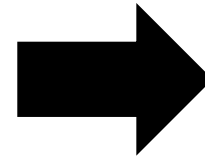
- The sun as a source of energy
- ➔ ■ **How does a photovoltaic solar cell work?**
- Types of solar cells and their efficiencies
- Economics of solar cells then and now
- Solar in Wisconsin
- The future



# Photovoltaic solar cells are the technology we need to convert sunlight into electricity.



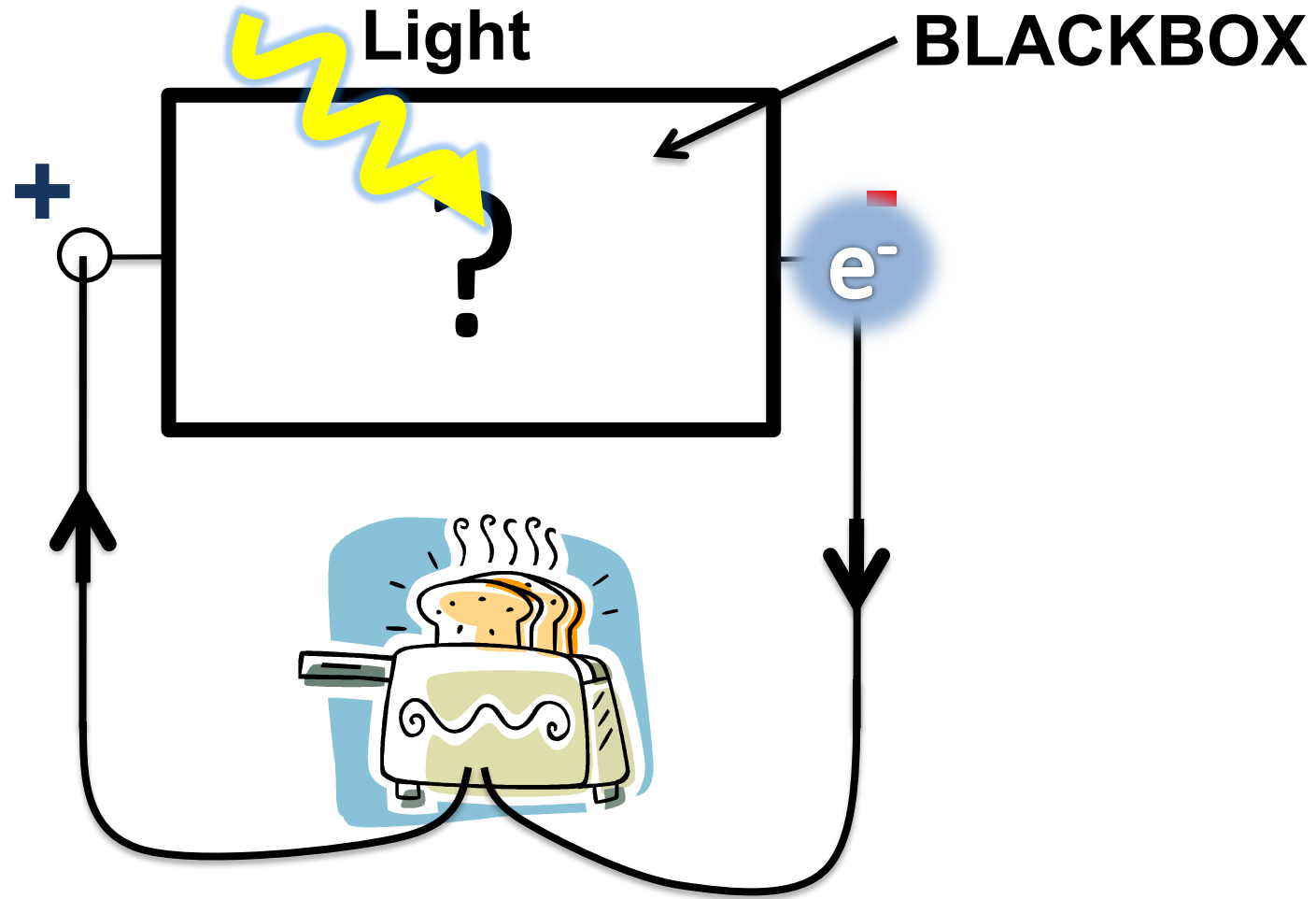
Light



Electricity

[www.freefoto.com](http://www.freefoto.com)

# What is a solar cell made from?



# What is a solar cell made from?

## ■ At heart: a semiconductor

- Silicon is most common

## ■ Also: inactive components such as metal, glass, structural materials



**A computer chip and a solar cell are both made from semiconductors.**

Ample silicon on earth



# Other semiconductors useful for solar cells

- GaAs
- CdTe
- CuInGaSe (CIGS)
- Organic molecules (C)
- PbI<sub>2</sub> + Organic molecules (Organic-inorganic hybrid perovskite)

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>

Ptable.com

Many elements and compounds are semiconductors.

# What makes a semiconductor a semiconductor?

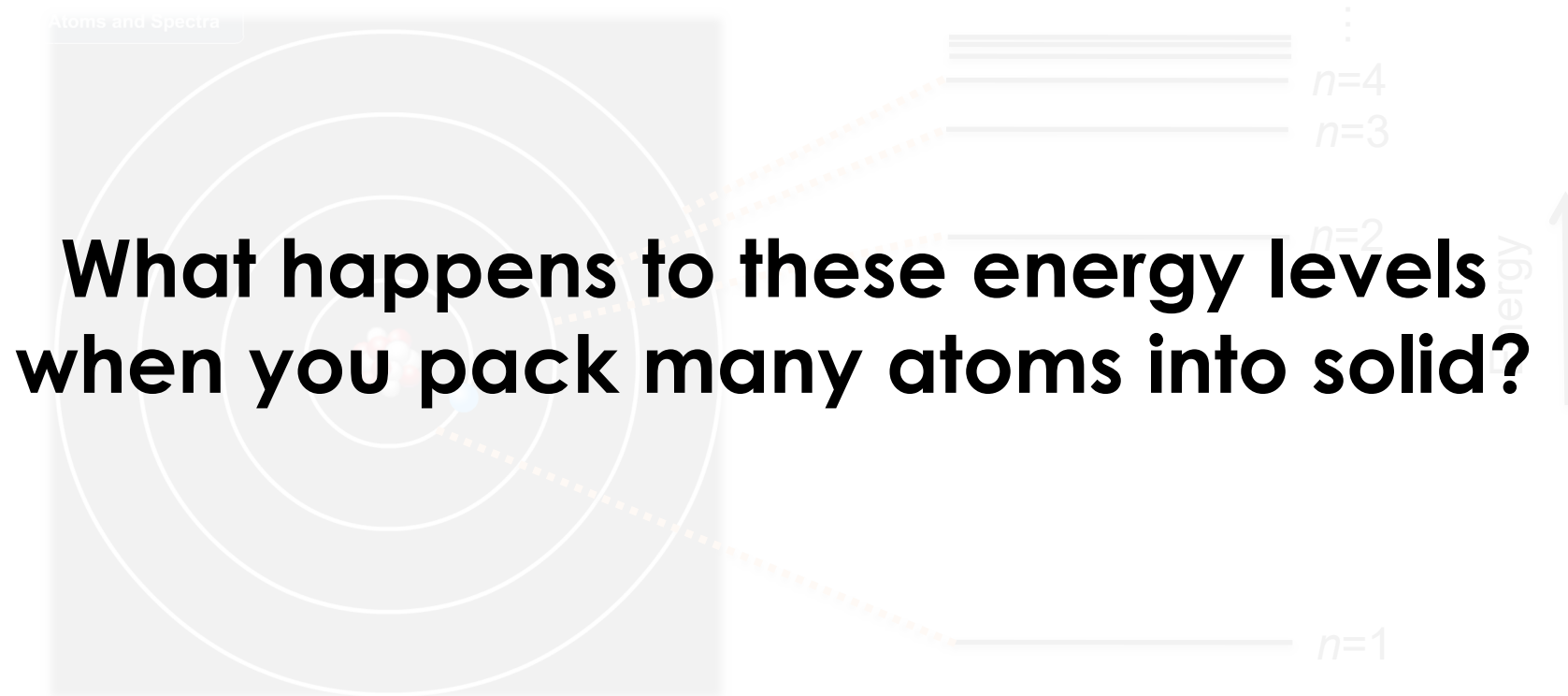
- It can act like both a conductor or an insulator.
- It can be doped to conduct negative charge (electrons) or positive charge (holes).
- It has an “energy gap” or “band gap”.



Conductor



Insulator



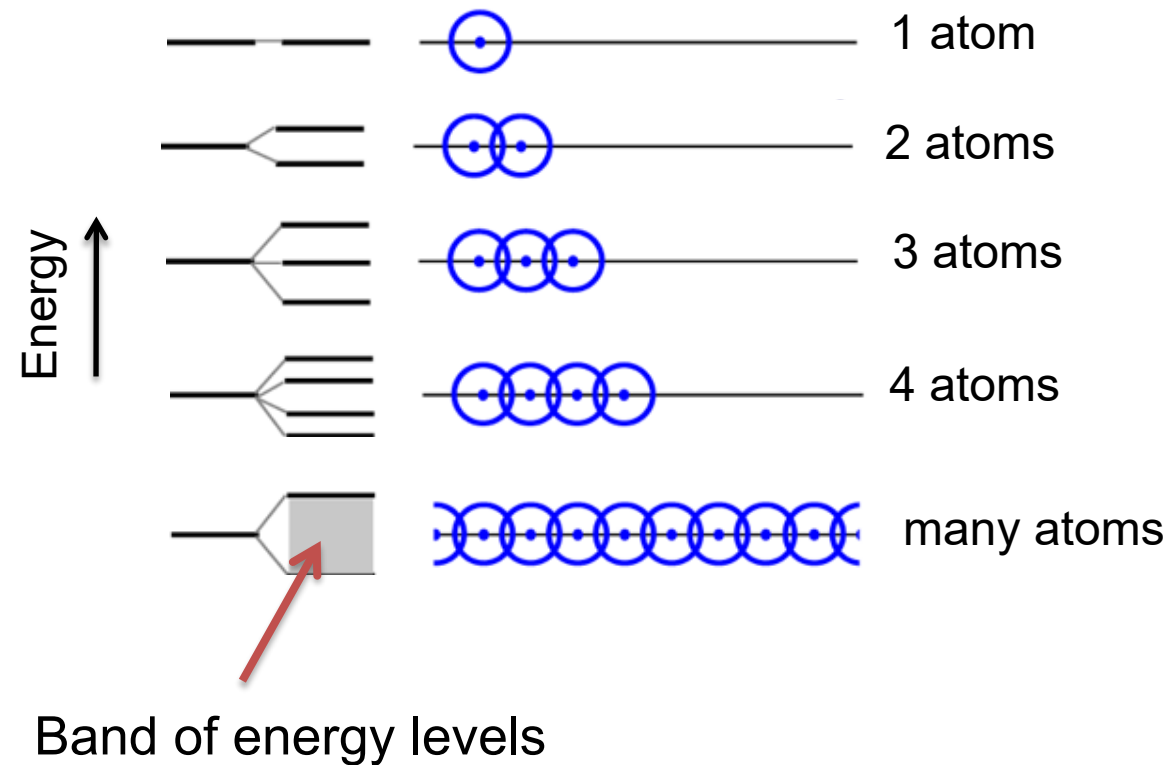
**What happens to these energy levels when you pack many atoms into solid?**

Image from:  
[http://spiff.rit.edu/classes/phys301/lectures/spec\\_lines/Atoms\\_Nav.swf](http://spiff.rit.edu/classes/phys301/lectures/spec_lines/Atoms_Nav.swf)

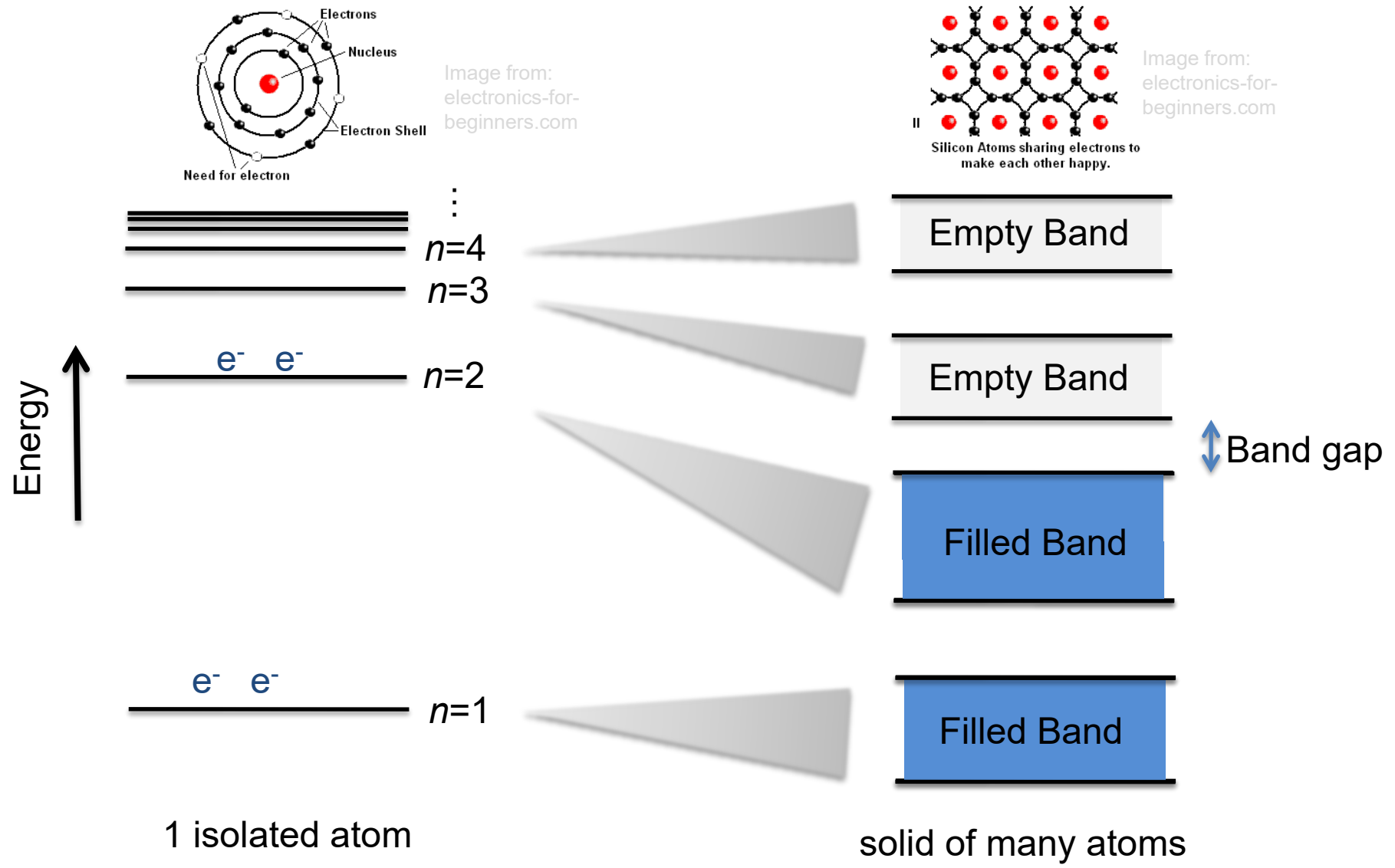


# Energy levels split in a solid to form bands

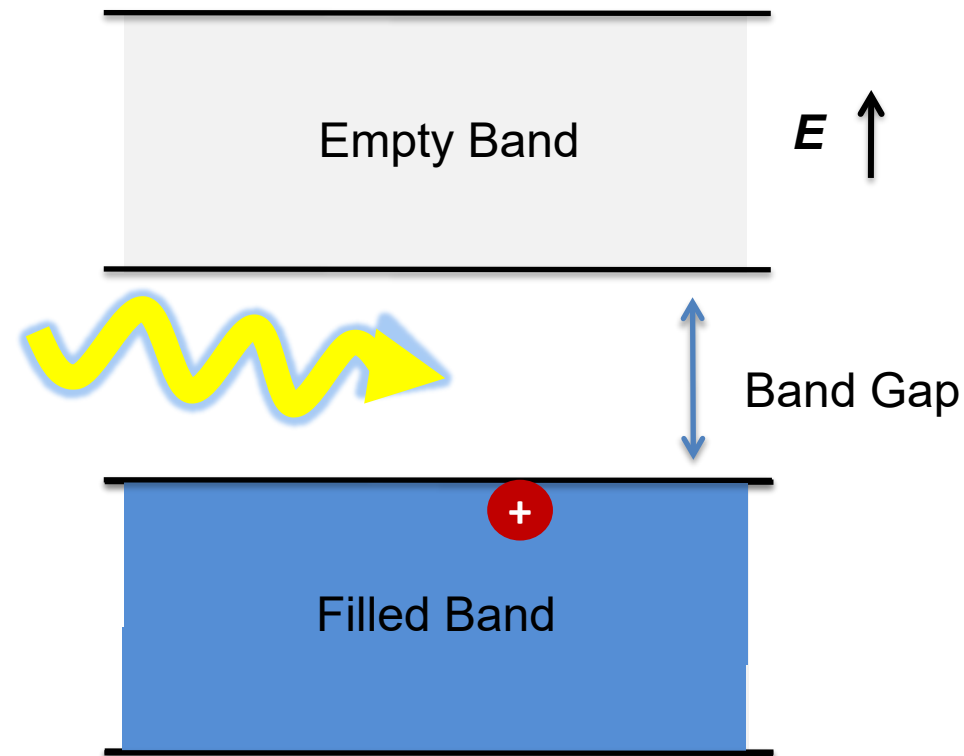
Look at effect on lowest  $n=1$  orbital.



# Semiconductor has gap between bands

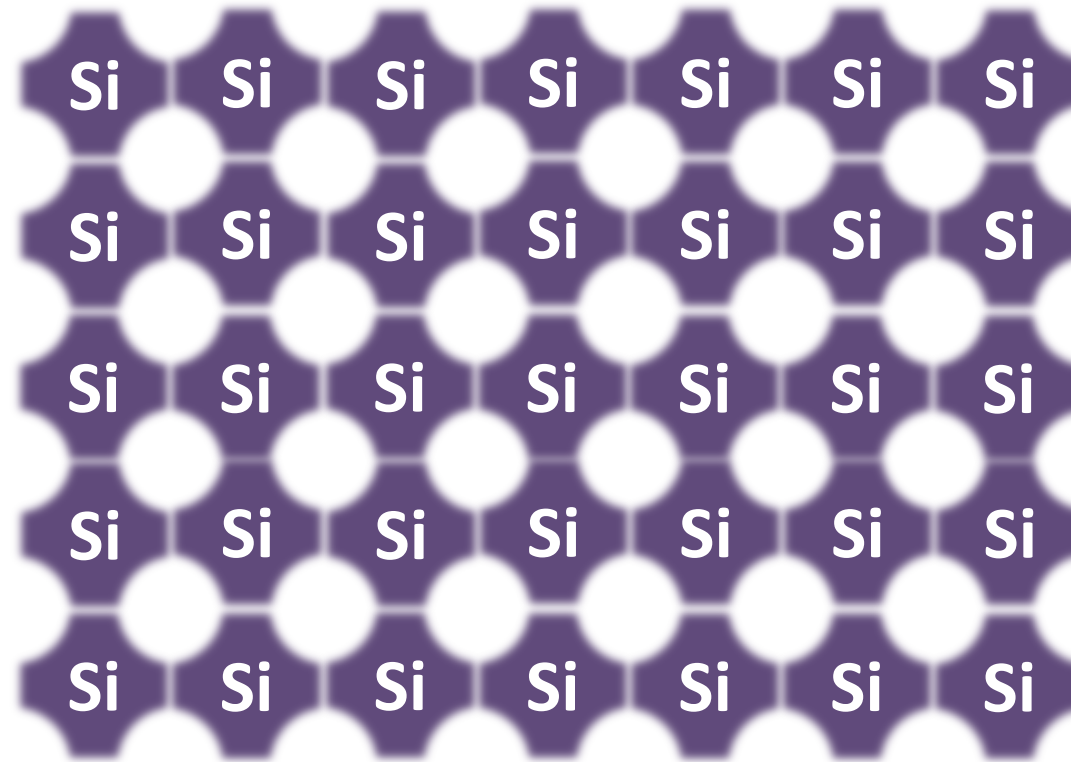


# Light excites electrons across band gap



Challenge is to next separate negative and positive charge to opposite sides of a solar cell before the electron falls back down to the filled band and turns into heat!

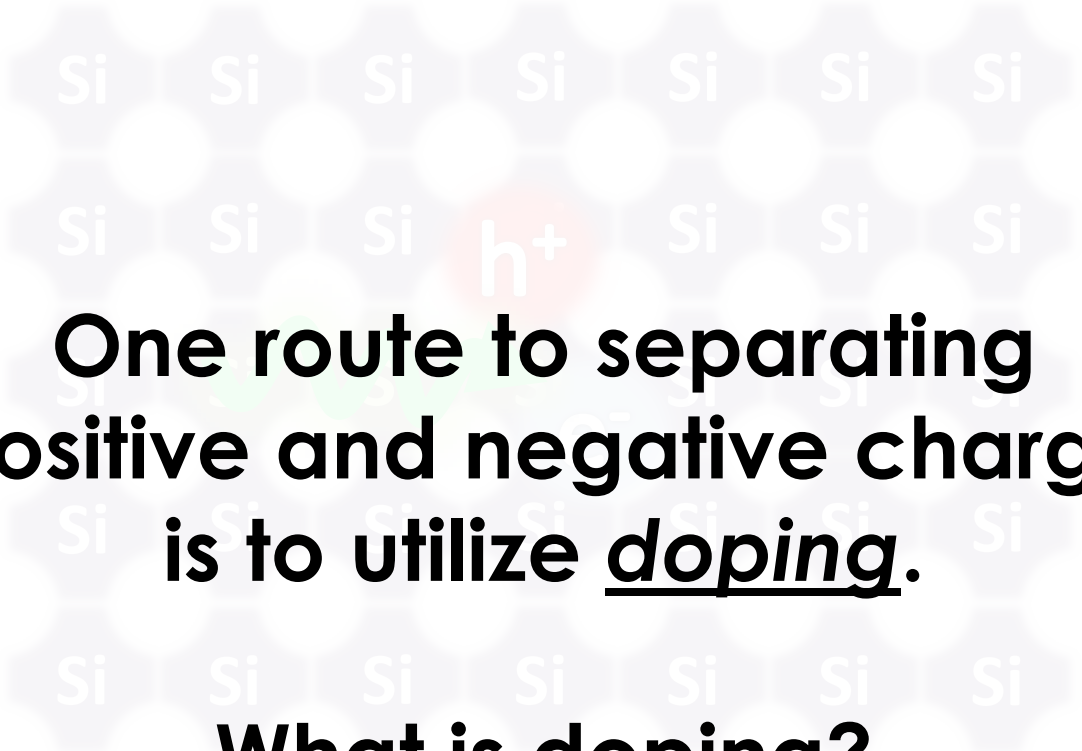
# How can charge be spatially separated?



## *Example 2D Representation of Si:*

- Each Si atom is covalently bonded to 4 other Si atoms.
- All 4 electrons in the outer-shell are in covalent bonds.
- **No free charges to move around – low conductivity.**

# Absorption of light



**One route to separating positive and negative charge is to utilize doping.**

## What is doping?

- When a semiconductor absorbs light  $\rightarrow$  positive (holes) and negative (electrons) charges are created.
- But, how can you induce their spontaneous separation?

# Not this kind of doping!!!



*www.independent.co.uk: Was London 2012's 1500m Olympic final the dirtiest race in history?*



# What is doping?

- Doping results from the ***intentional*** or ***unintentional*** addition of impurities to a semiconductor.

# Periodic Table of Elements

Boron and Phosphorous are dopants

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18		
1	<b>H</b> Hydrogen 1.00794																		<b>He</b> Helium 4.002602	
2	<b>Li</b> Lithium 6.941	<b>Be</b> Beryllium 9.012182																		<b>Ne</b> Neon 20.1797
3	<b>Na</b> Sodium 22.98976928	<b>Mg</b> Magnesium 24.3050																		<b>Ar</b> Argon 39.948
4	<b>K</b> Potassium 39.0983	<b>Ca</b> Calcium 40.078	<b>Sc</b> Scandium 44.955912	<b>Ti</b> Titanium 47.887	<b>V</b> Vanadium 50.9415	<b>Cr</b> Chromium 51.9961	<b>Mn</b> Manganese 54.938045	<b>Fe</b> Iron 55.845	<b>Co</b> Cobalt 58.933195	<b>Ni</b> Nickel 58.6934	<b>Cu</b> Copper 63.546	<b>Zn</b> Zinc 65.38	<b>Ga</b> Gallium 69.723	<b>Ge</b> Germanium 72.64	<b>As</b> Arsenic 74.92160	<b>Se</b> Selenium 78.96	<b>Br</b> Bromine 79.904	<b>Kr</b> Krypton 83.798		
5	<b>Rb</b> Rubidium 85.4678	<b>Sr</b> Strontium 87.62	<b>Y</b> Yttrium 88.90585	<b>Zr</b> Zirconium 91.224	<b>Nb</b> Niobium 92.90638	<b>Mo</b> Molybdenum 95.96	<b>Tc</b> Technetium (97.9072)	<b>Ru</b> Ruthenium 101.07	<b>Rh</b> Rhodium 102.90550	<b>Pd</b> Palladium 106.42	<b>Ag</b> Silver 107.8682	<b>Cd</b> Cadmium 112.411	<b>In</b> Indium 114.818	<b>Sn</b> Tin 118.710	<b>Sb</b> Antimony 121.757	<b>Te</b> Tellurium 127.603	<b>I</b> Iodine 126.905	<b>Xe</b> Xenon 131.293		
6	<b>Cs</b> Caesium 132.9054519	<b>Ba</b> Barium 137.327	57-71	<b>Hf</b> Hafnium 178.49	<b>Ta</b> Tantalum 180.94788	<b>W</b> Tungsten 183.84	<b>Re</b> Rhenium 186.207	<b>Os</b> Osmium 190.23	<b>Ir</b> Iridium 192.217	<b>Pt</b> Platinum 195.084	<b>Au</b> Gold 196.966569	<b>Hg</b> Mercury 200.59	<b>Tl</b> Thallium 204.3833	<b>Pb</b> Lead 207.2	<b>Bi</b> Bismuth 208.9804	<b>Po</b> Polonium (209)	<b>At</b> Astatine (210)	<b>Rn</b> Radon (222.0176)		
7	<b>Fr</b> Francium (223)	<b>Ra</b> Radium (226)	89-103	<b>Rf</b> Rutherfordium (261)	<b>Db</b> Dubnium (262)	<b>Sg</b> Seaborgium (266)	<b>Bh</b> Bohrium (264)	<b>Hs</b> Hassium (277)	<b>Mt</b> Meitnerium (268)	<b>Ds</b> Darmstadtium (271)	<b>Rg</b> Roentgenium (272)	<b>Cn</b> Copernicium (285)	<b>Nh</b> Nihonium (286)	<b>Fl</b> Flerovium (289)	<b>Mc</b> Moscovium (288)	<b>Lv</b> Livermorium (293)	<b>Ts</b> Tennessine (294)	<b>Og</b> Oganesson (294)		

**C** Solid  
**Hg** Liquid  
**H** Gas  
**Rf** Unknown

**Metals**  
 Alkali metals  
 Alkaline earth metals  
 Lanthanoids  
 Actinoids  
 Transition metals  
 Poor metals

**Nonmetals**  
 Other nonmetals  
 Noble gases

Silicon is a group IV element → 4 electrons short of a complete "shell" of 8 electrons

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

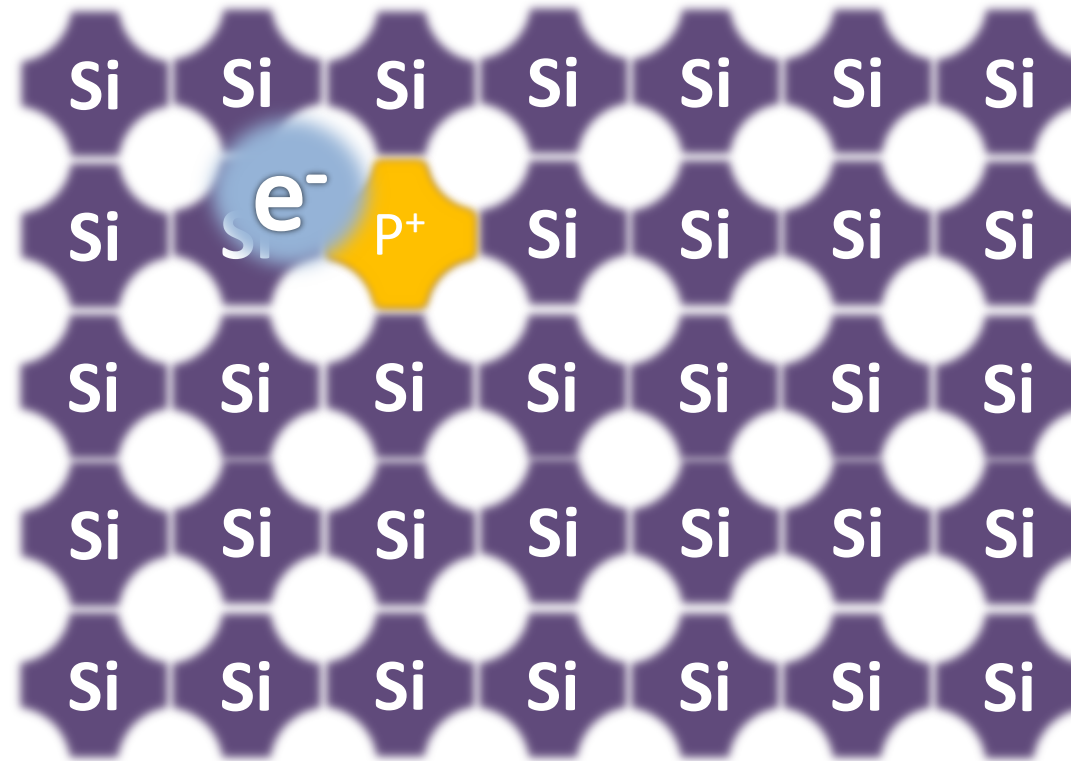
Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>



57 <b>La</b> Lanthanum 138.90547	58 <b>Ce</b> Cerium 140.116	59 <b>Pr</b> Praseodymium 140.90765	60 <b>Nd</b> Neodymium 144.242	61 <b>Pm</b> Promethium (145)	62 <b>Sm</b> Samarium 150.38	63 <b>Eu</b> Europium 151.964	64 <b>Gd</b> Gadolinium 157.25	65 <b>Tb</b> Terbium 158.92535	66 <b>Dy</b> Dysprosium 162.500	67 <b>Ho</b> Holmium 164.93032	68 <b>Er</b> Erbium 167.259	69 <b>Tm</b> Thulium 168.93421	70 <b>Yb</b> Ytterbium 173.054	71 <b>Lu</b> Lutetium 174.9688
89 <b>Ac</b> Actinium (227)	90 <b>Th</b> Thorium 232.03806	91 <b>Pa</b> Protactinium 231.03688	92 <b>U</b> Uranium 238.02891	93 <b>Np</b> Neptunium (237)	94 <b>Pu</b> Plutonium (244)	95 <b>Am</b> Americium (243)	96 <b>Cm</b> Curium (247)	97 <b>Bk</b> Berkelium (247)	98 <b>Cf</b> Californium (251)	99 <b>Es</b> Einsteinium (252)	100 <b>Fm</b> Fermium (257)	101 <b>Md</b> Mendelevium (258)	102 <b>No</b> Nobelium (259)	103 <b>Lr</b> Lawrencium (262)

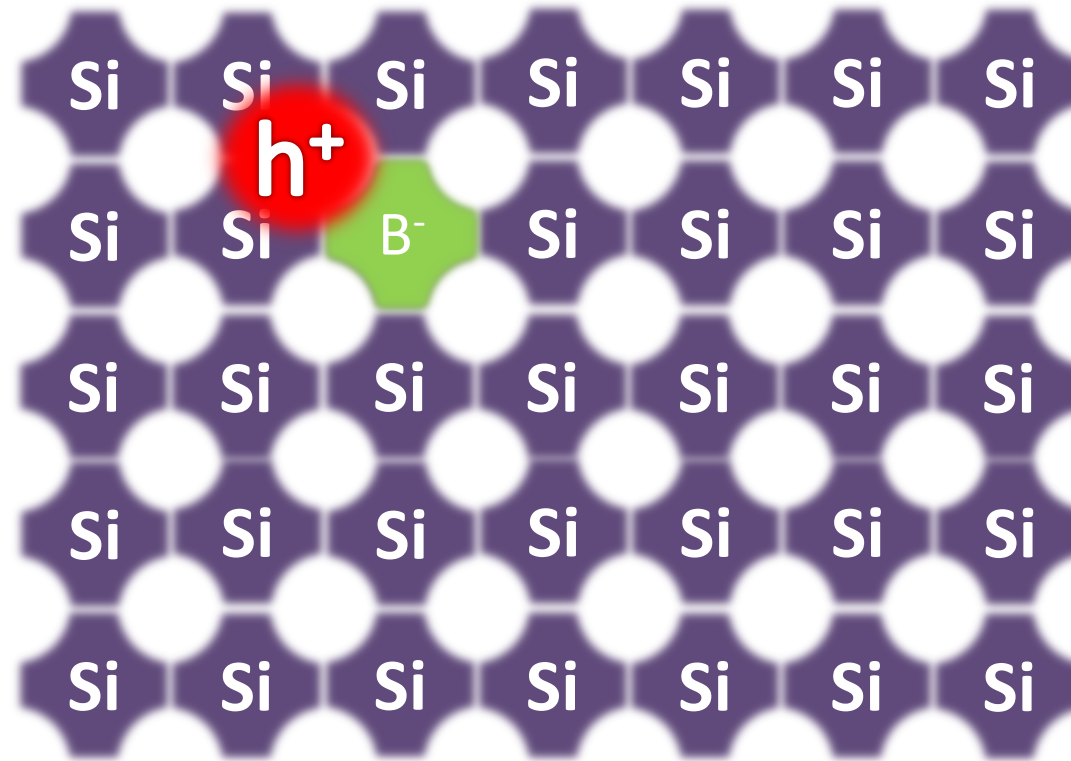


# N-type (electron doped) silicon (Si)



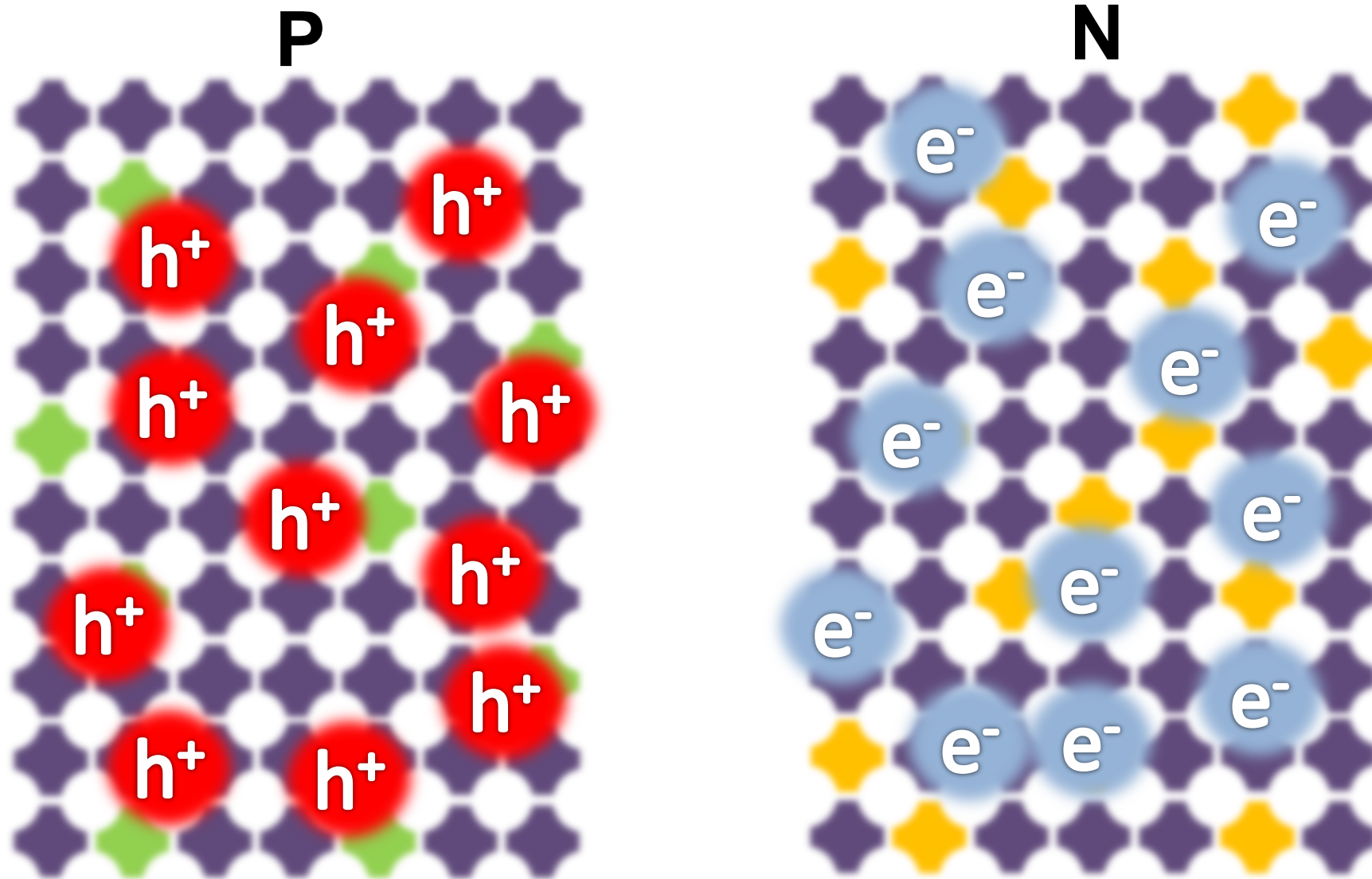
- Replacement of group IV Si atom with group V P atom
- P atom becomes positively ionized, releasing free electron that can move and conduct.

# P-type (“hole” doped) silicon (Si)

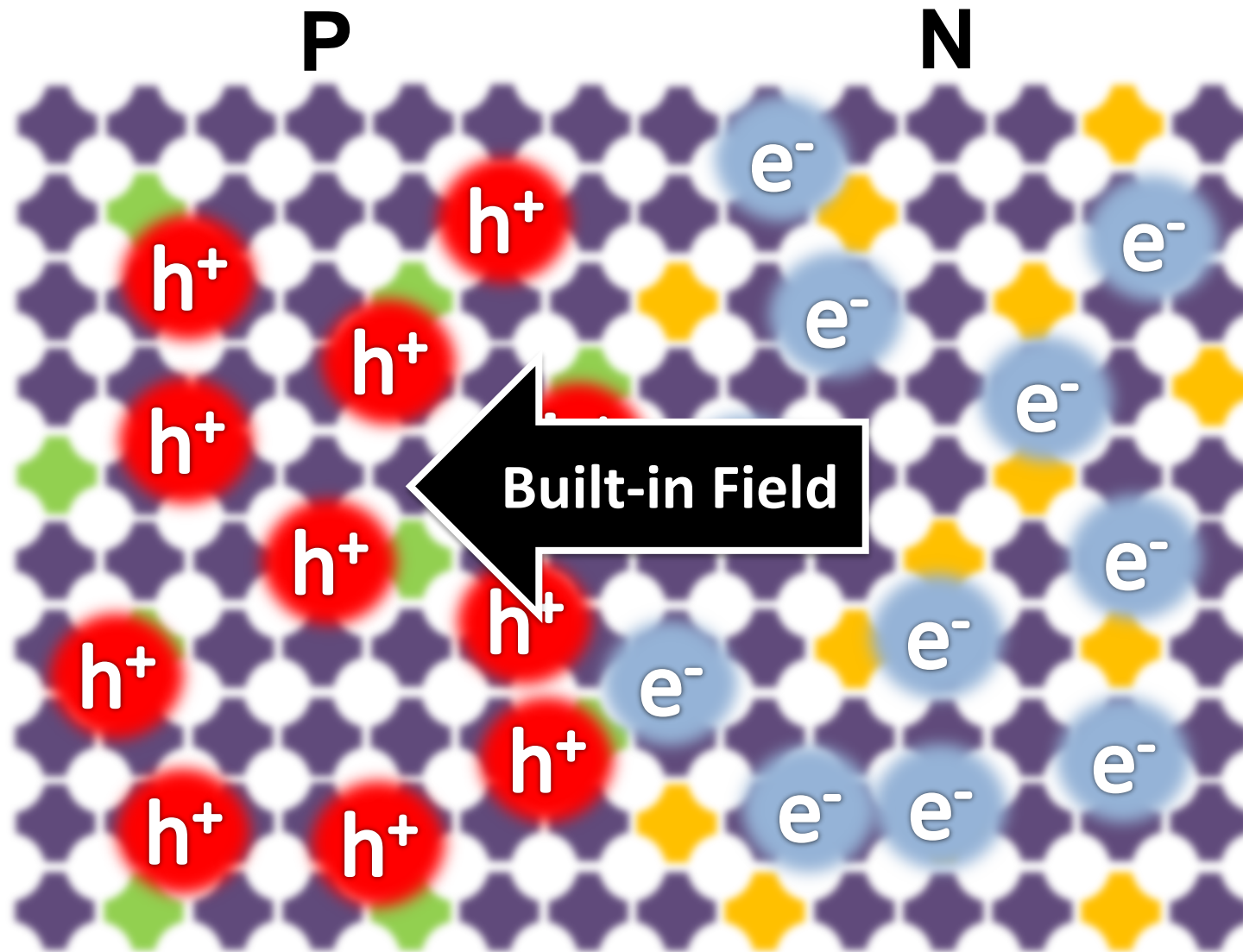


- Replacement of group IV Si atom with group III B atom
- B atom becomes negatively ionized, effectively releasing positive charge called a hole that can move and conduct.

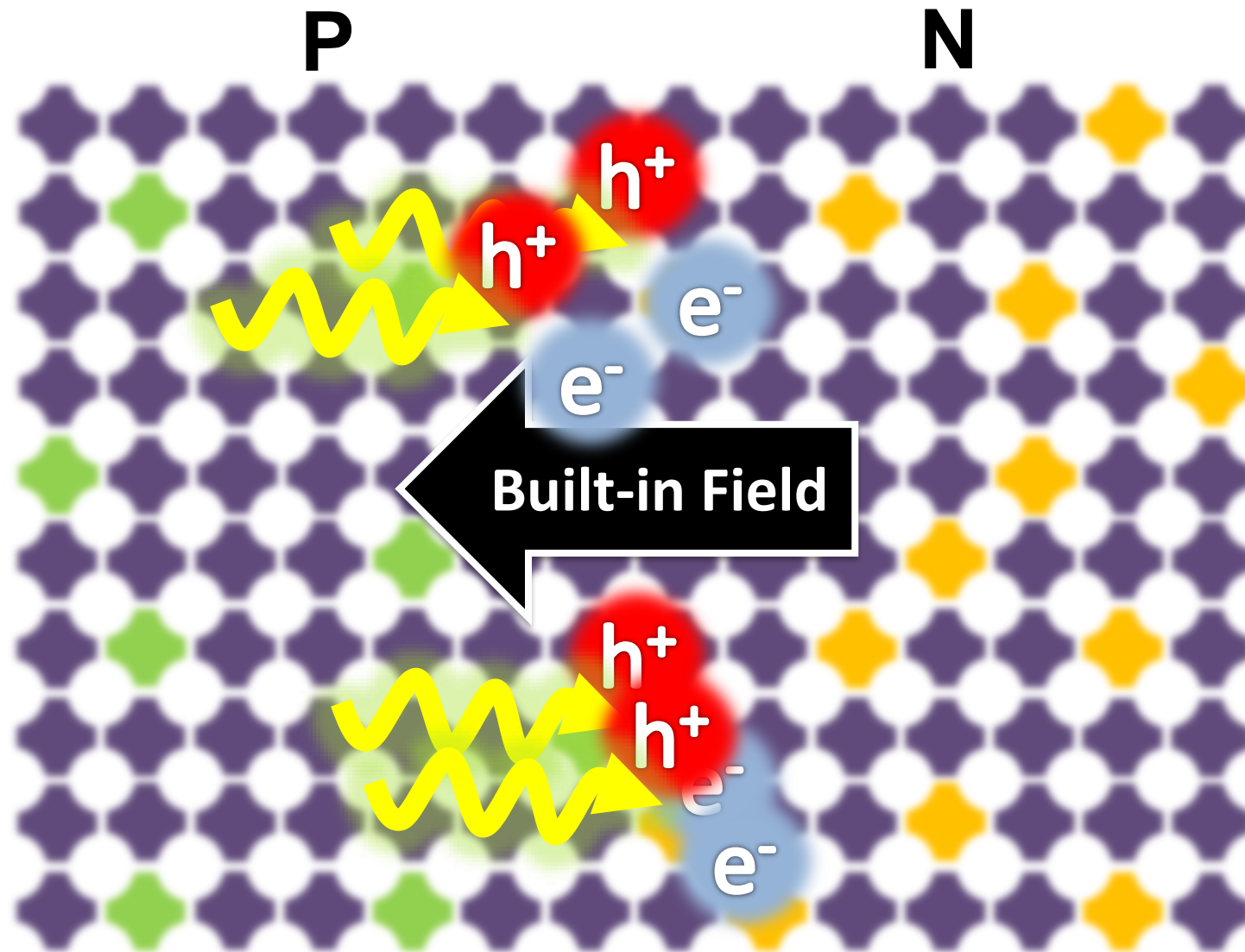
# A common device is: P-N junction



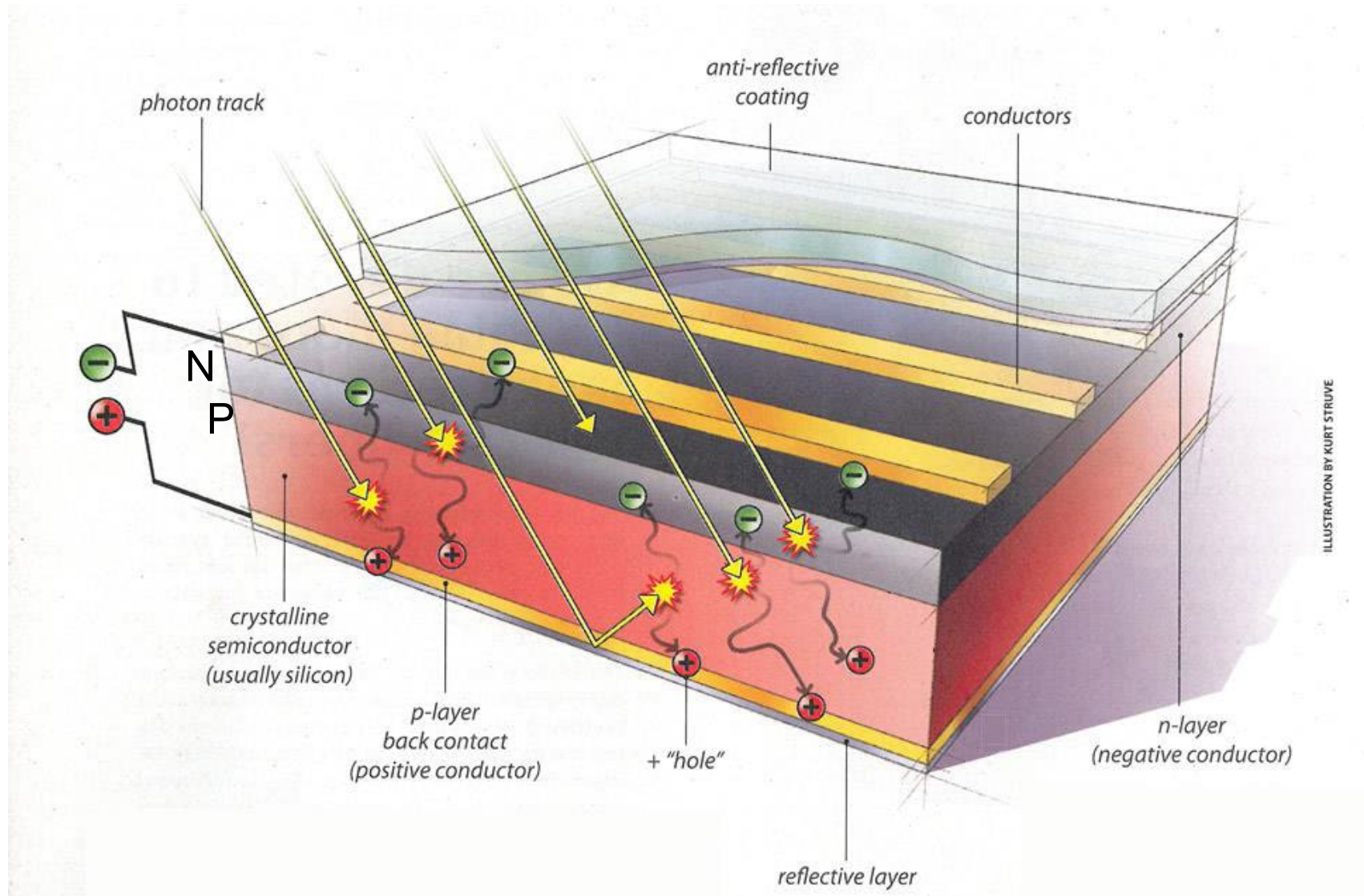
# P-N junction



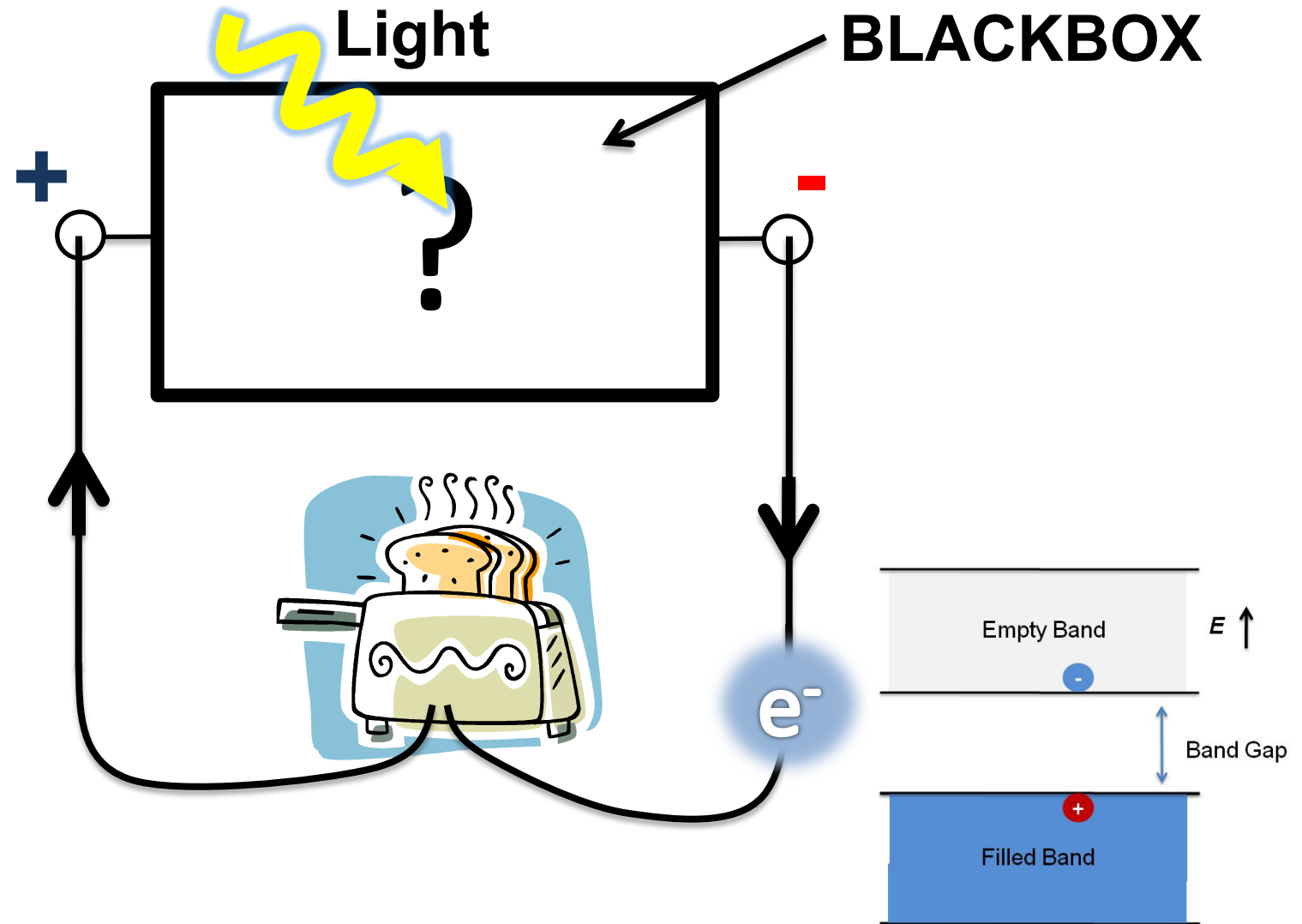
# P-N junction



# Other components: electrodes, coatings



# Our blackbox



# Conclusions about how a solar cell works on the level of electrons and atoms

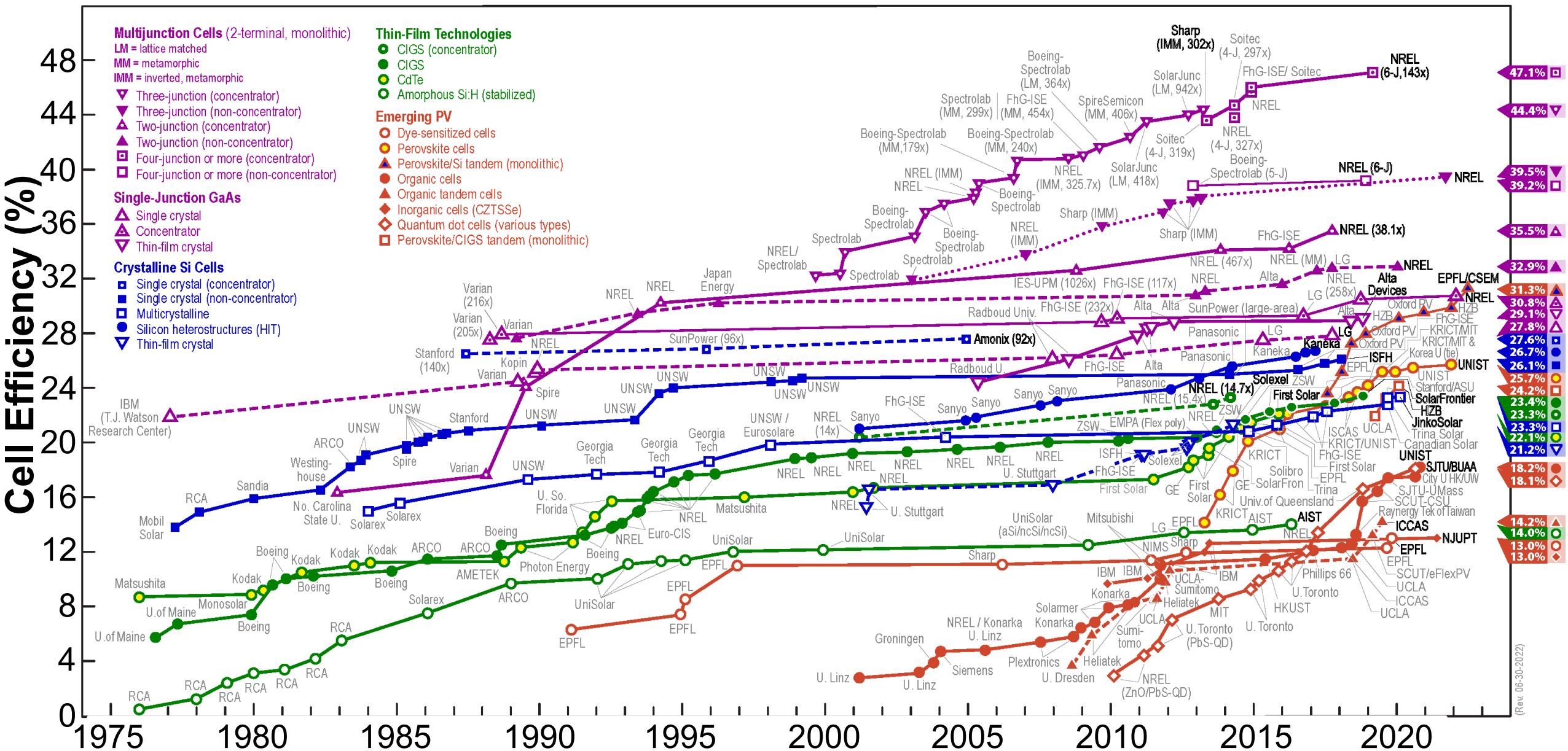
- The active materials in a photovoltaic solar cell are semiconductors.
- Light excites negative charge (electrons) higher in energy across the band gap.
- A junction between P-doped and N-doped semiconductors (PN junction) is used to separate the positive and negative charges → electricity.



- The sun as a source of energy
- How does a photovoltaic solar cell work?
- ➔ ■ **Types of solar cells and their efficiencies**
- Economics of solar cells then and now
- Solar in Wisconsin
- The future

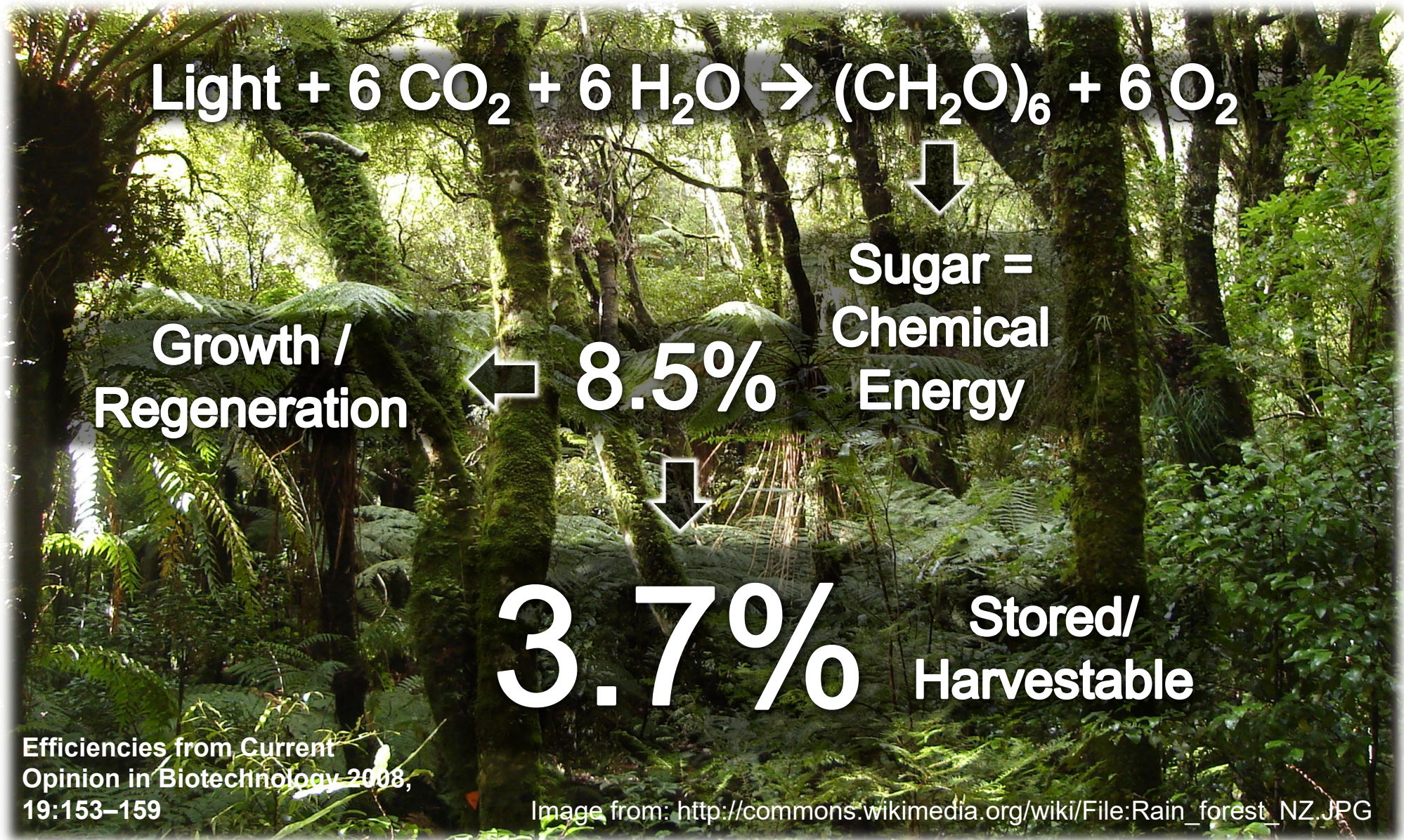


# Best Research-Cell Efficiencies



Observation #1: Efficiency up with time! Ranges from 10-47%!!!

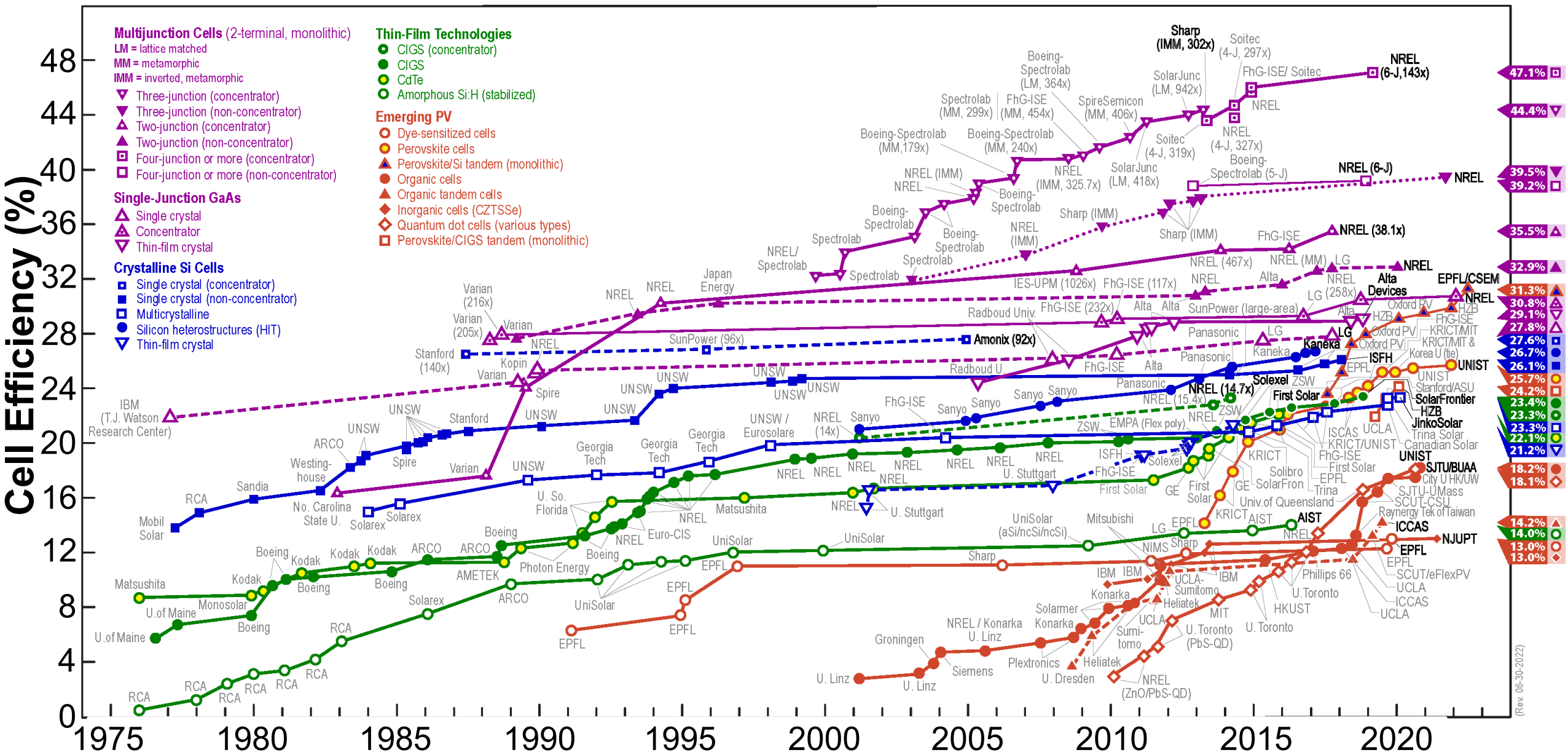
# Aside: How efficient is photosynthesis?



Efficiencies from Current Opinion in Biotechnology 2008, 19:153–159

Image from: [http://commons.wikimedia.org/wiki/File:Rain\\_forest\\_NZ.JPG](http://commons.wikimedia.org/wiki/File:Rain_forest_NZ.JPG)

# Best Research-Cell Efficiencies



(Rev. 06-30-2022)

Observation #2: Lots of different cell-types!

# Cell-type classifications

## Crystallinity

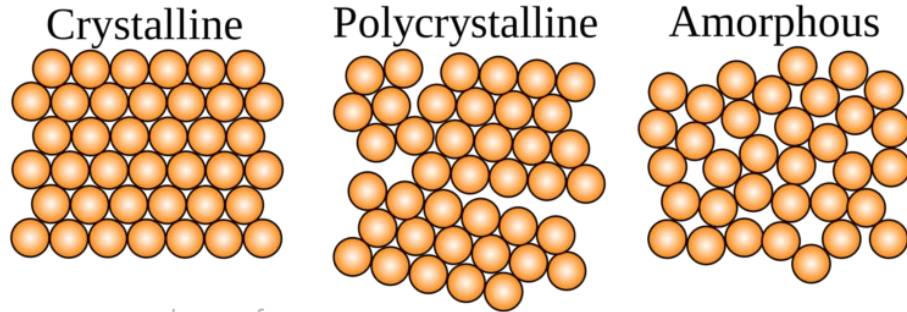


Image from:  
[https://commons.wikimedia.org/wiki/File:Crystalline\\_polycrystalline\\_amorphous.svg](https://commons.wikimedia.org/wiki/File:Crystalline_polycrystalline_amorphous.svg)

## Thick Crystal or Thin Film

*Thin-Film*



*Thick Crystal*

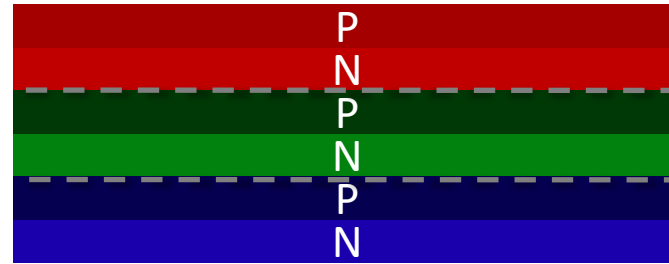


## Number of Junctions

*Single-Junction:*



*Multi-Junction:*



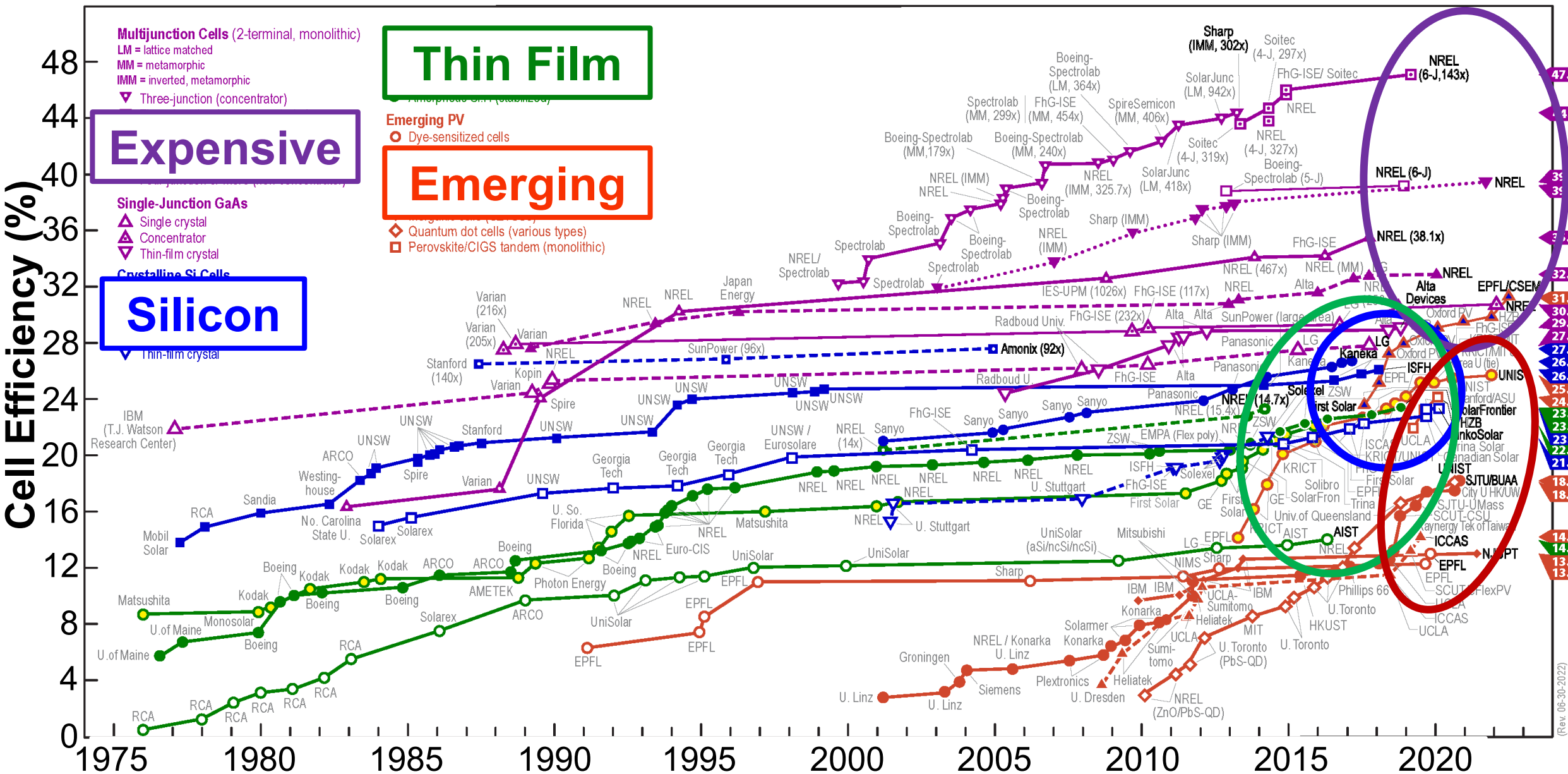
**Small Band gap**  
**Medium Band gap**  
**Large Band gap**

## Composition

**Periodic Table of Elements**

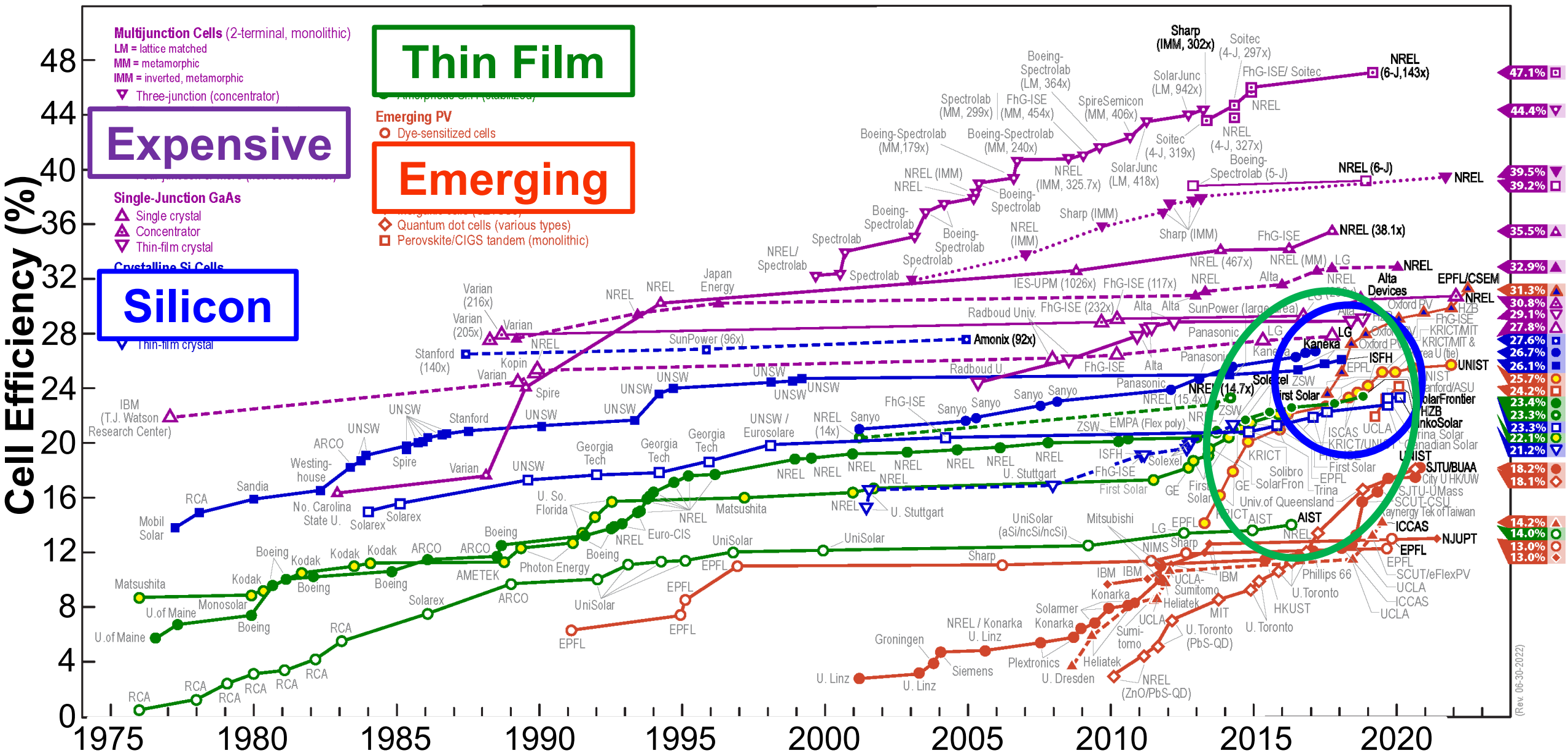
Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>

# Best Research-Cell Efficiencies



Observation #2: Lots of different cell-types!

# Best Research-Cell Efficiencies



(Rev. 06-30-2022)

Observation #2: Lots of different cell-types!

# Single crystal silicon

## Crystallinity

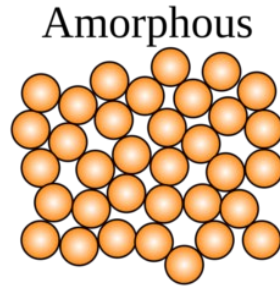
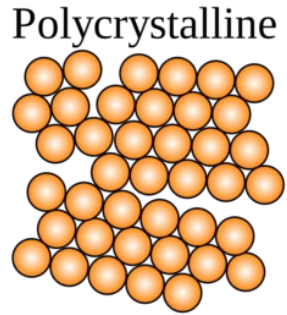
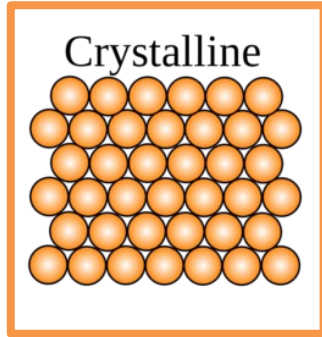


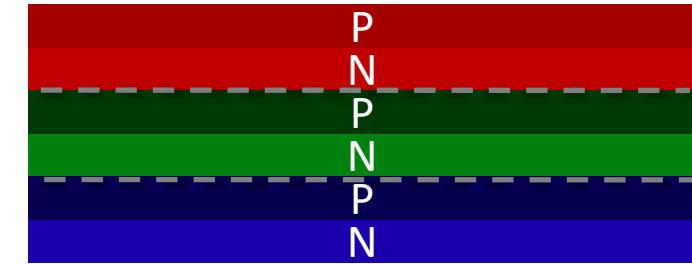
Image from:  
[https://commons.wikimedia.org/wiki/File:Crystalline\\_polycrystalline\\_amorphous.svg](https://commons.wikimedia.org/wiki/File:Crystalline_polycrystalline_amorphous.svg)

## Number of Junctions

Single-Junction:



Multi-Junction:



**Small  
Band gap**  
**Medium  
Band gap**  
**Large  
Band gap**

## Thick Crystal or Thin Film

Thin-Film



Thick Crystal



## Composition

**Periodic Table of Elements**

Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>



# Single crystal silicon manufacturing



**Polycrystalline Si**  
> 99.9999%

→  
**1400 °C**



<http://www.quora.com/Semiconductor-Fabrication/How-do-silicon-boules-not-break-off-during-semiconductor-fabrication>

**Single crystalline Si**



<http://www.products.cvdequipment.com/applications/polysilicon/1/>

<http://www.youtube.com/watch?v=aWVyhzuHnQ&NR=1>  
(start @ 1:48)

# Polycrystalline silicon

## Crystallinity

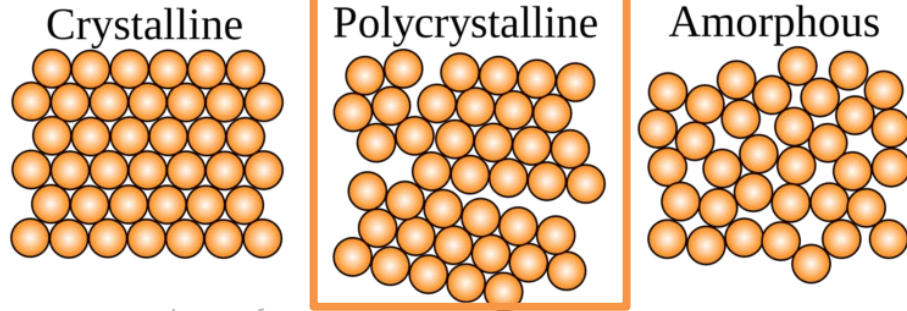


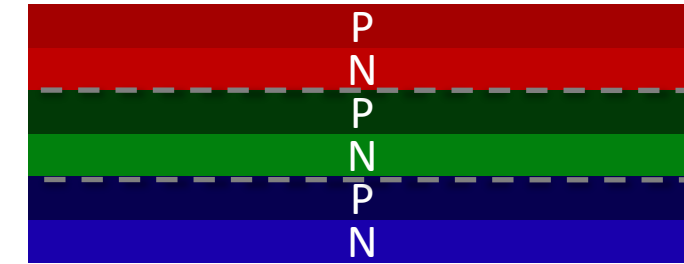
Image from:  
[https://commons.wikimedia.org/wiki/File:Crystalline\\_polycrystalline\\_amorphous.svg](https://commons.wikimedia.org/wiki/File:Crystalline_polycrystalline_amorphous.svg)

## Number of Junctions

Single-Junction:



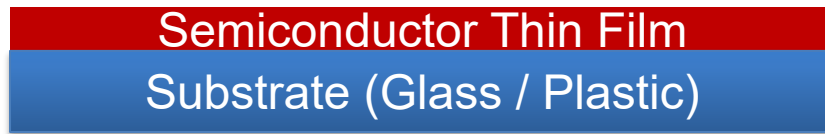
Multi-Junction:



**Small Band gap**  
**Medium Band gap**  
**Large Band gap**

## Thick Crystal or Thin Film

Thin-Film



Thick Crystal



## Composition

**Periodic Table of Elements**

Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>

# Polycrystalline silicon manufacturing

Deposition by chemical vapor deposition:  $\text{SiH}_4 (\text{g}) \rightarrow \text{Si} (\text{s}) + 2\text{H}_2 (\text{g})$ , 650 °C



■ Lower cost due to lower temperature processing but lower efficiency due to loss at grain boundaries.

Image from: [http://upload.wikimedia.org/wikipedia/commons/1/15/Polycrystalline-silicon-wafer\\_20060626\\_568.jpg](http://upload.wikimedia.org/wikipedia/commons/1/15/Polycrystalline-silicon-wafer_20060626_568.jpg)

# Thin film

## Crystallinity

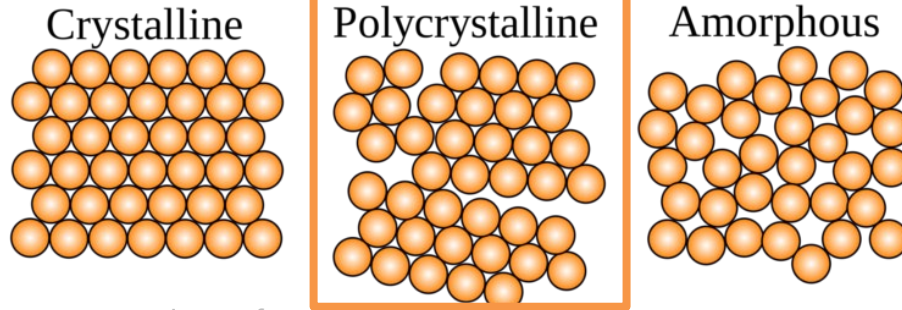


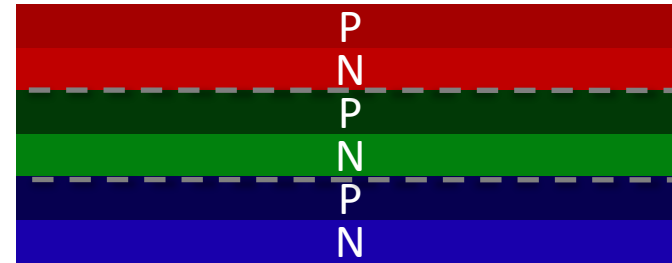
Image from:  
[https://commons.wikimedia.org/wiki/File:Crystalline\\_polycrystalline\\_amorphous.svg](https://commons.wikimedia.org/wiki/File:Crystalline_polycrystalline_amorphous.svg)

## Number of Junctions

Single-Junction:



Multi-Junction:



**Small Band gap**  
**Medium Band gap**  
**Large Band gap**

## Thick Crystal or Thin Film

Thin-Film



Thick Crystal



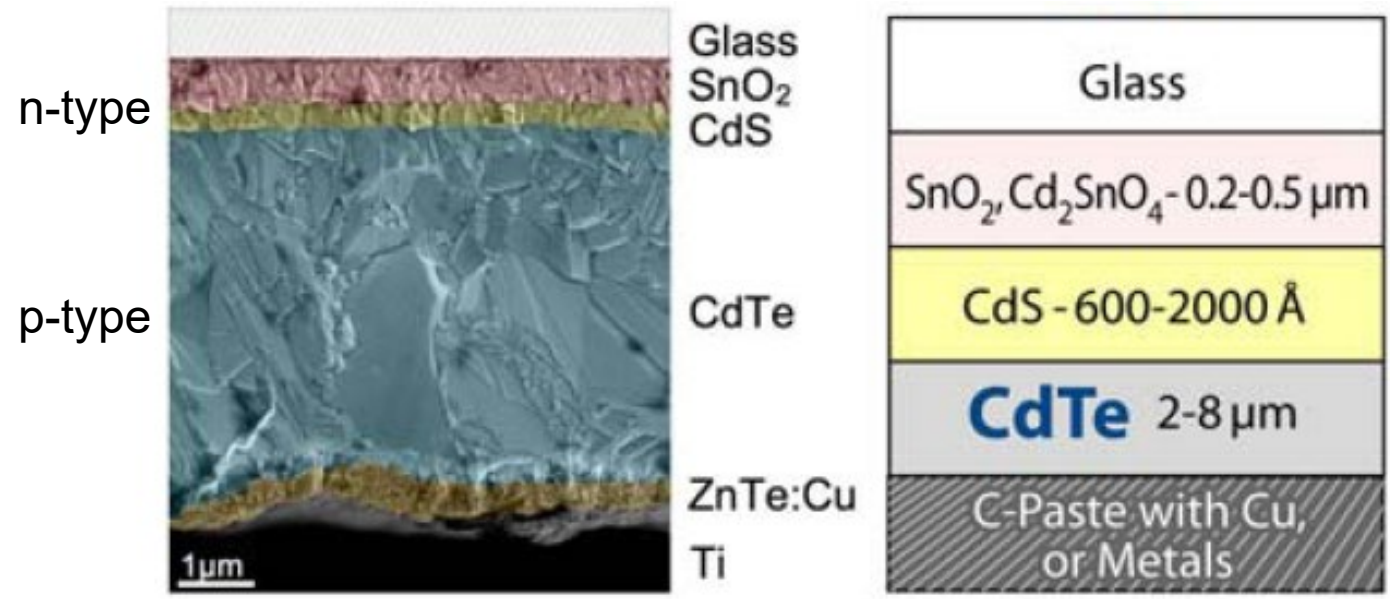
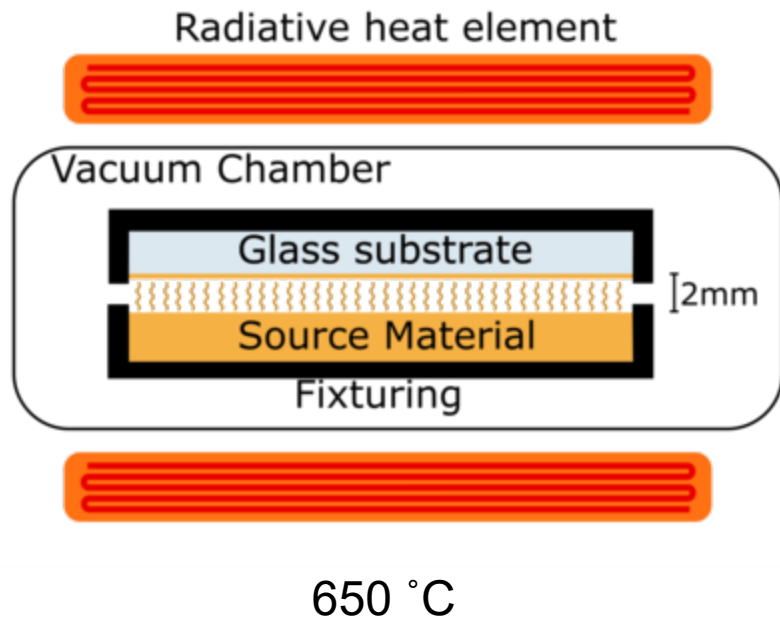
## Composition

**Periodic Table of Elements**

Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>

# Thin film manufacturing (CdTe)

## Sublimation

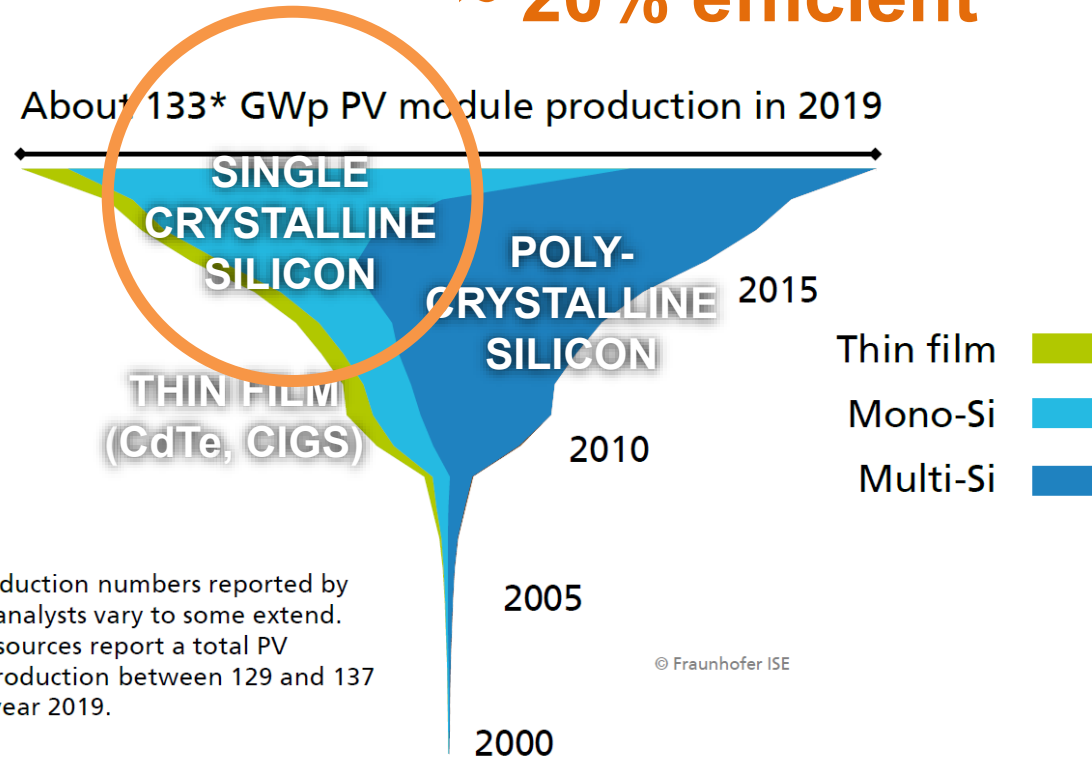


from: Noufi and Zweibel, IEEE 4<sup>th</sup> WCPEC-4 (2006)

# Single crystal silicon is most prevalent.

## Annual PV Production by Technology Worldwide (in GWp)

~ 20% efficient



\*2019 production numbers reported by different analysts vary to some extent. Different sources report a total PV module production between 129 and 137 GWp for year 2019.

© Fraunhofer ISE

Data: from 2000 to 2009: Navigant; from 2010: IHS Markit. Graph: PSE Projects GmbH 2020

- The sun as a source of energy
- How does a photovoltaic solar cell work?
- Types of solar cells and their efficiencies
- ➔ ■ **Economics of solar cells then and now**
- Solar in Wisconsin
- The future



# Economics on both residential and utility scales

## Residential

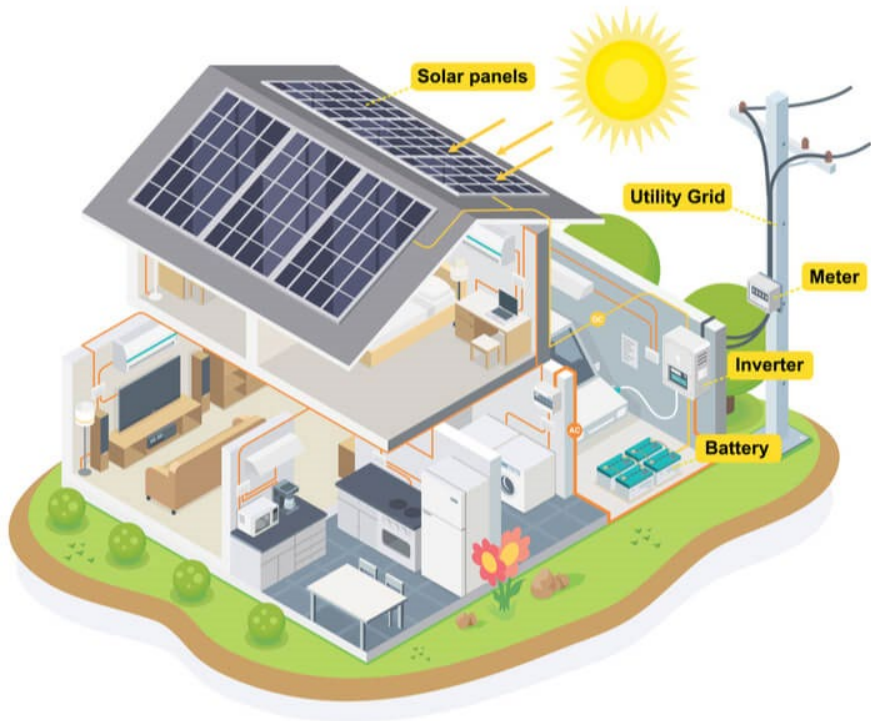


Image from: <https://www.saveonenergy.com/solar-energy/solar-panels-for-home/>

## Utility-Scale (Solar Farms)

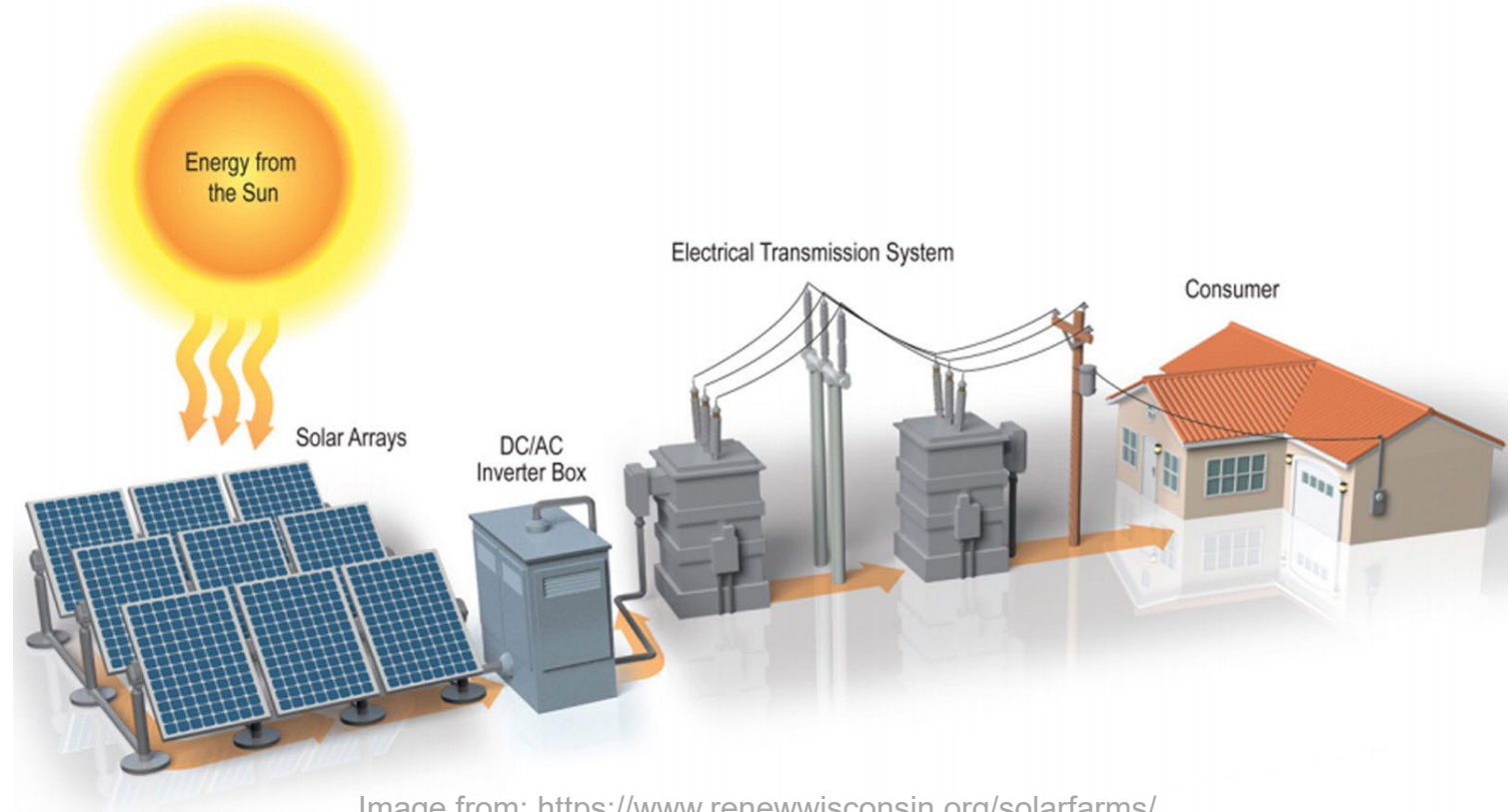
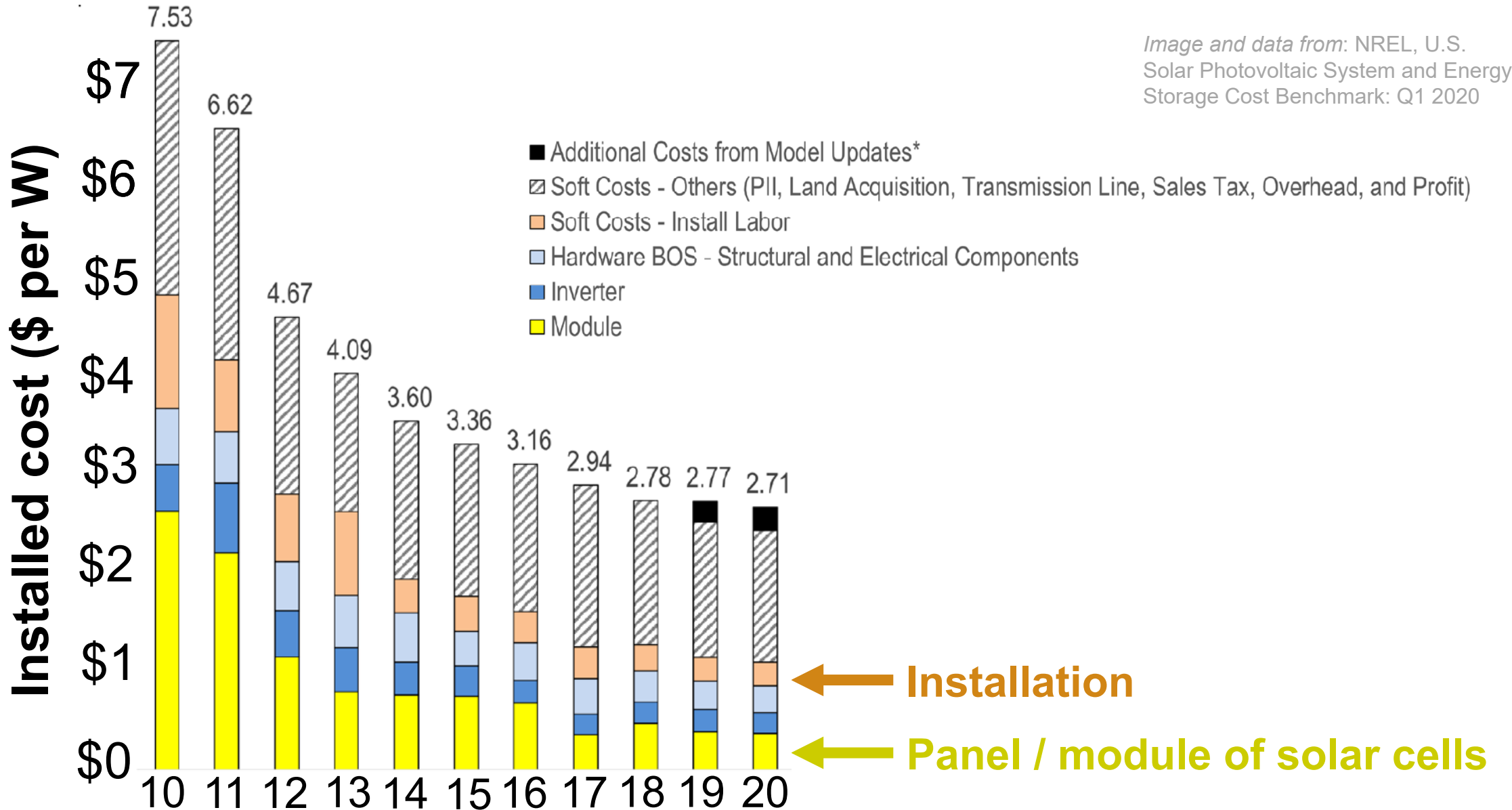


Image from: <https://www.renewwisconsin.org/solarfarms/>



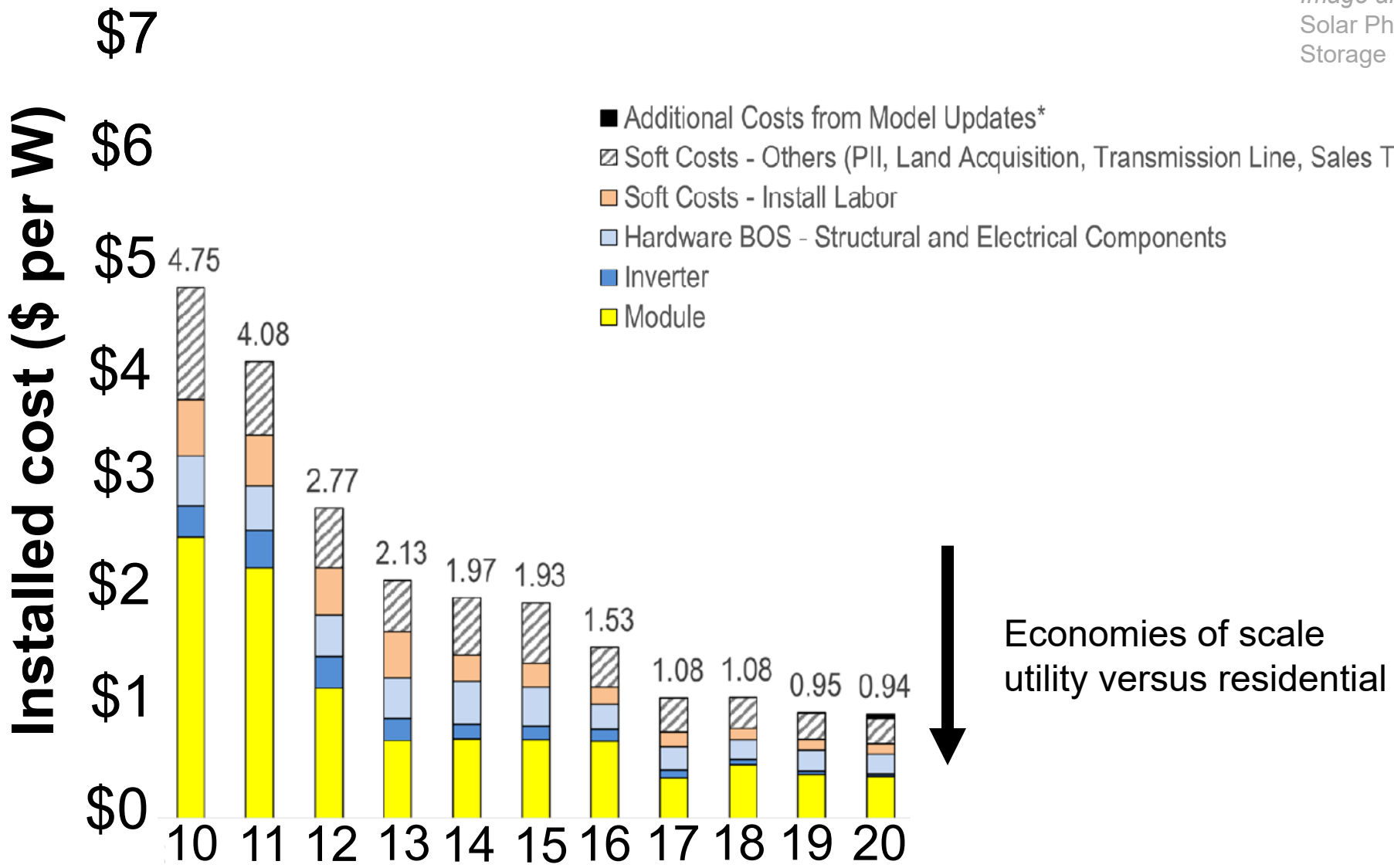
# Installed cost: *residential*

Image and data from: NREL, U.S. Solar Photovoltaic System and Energy Storage Cost Benchmark: Q1 2020

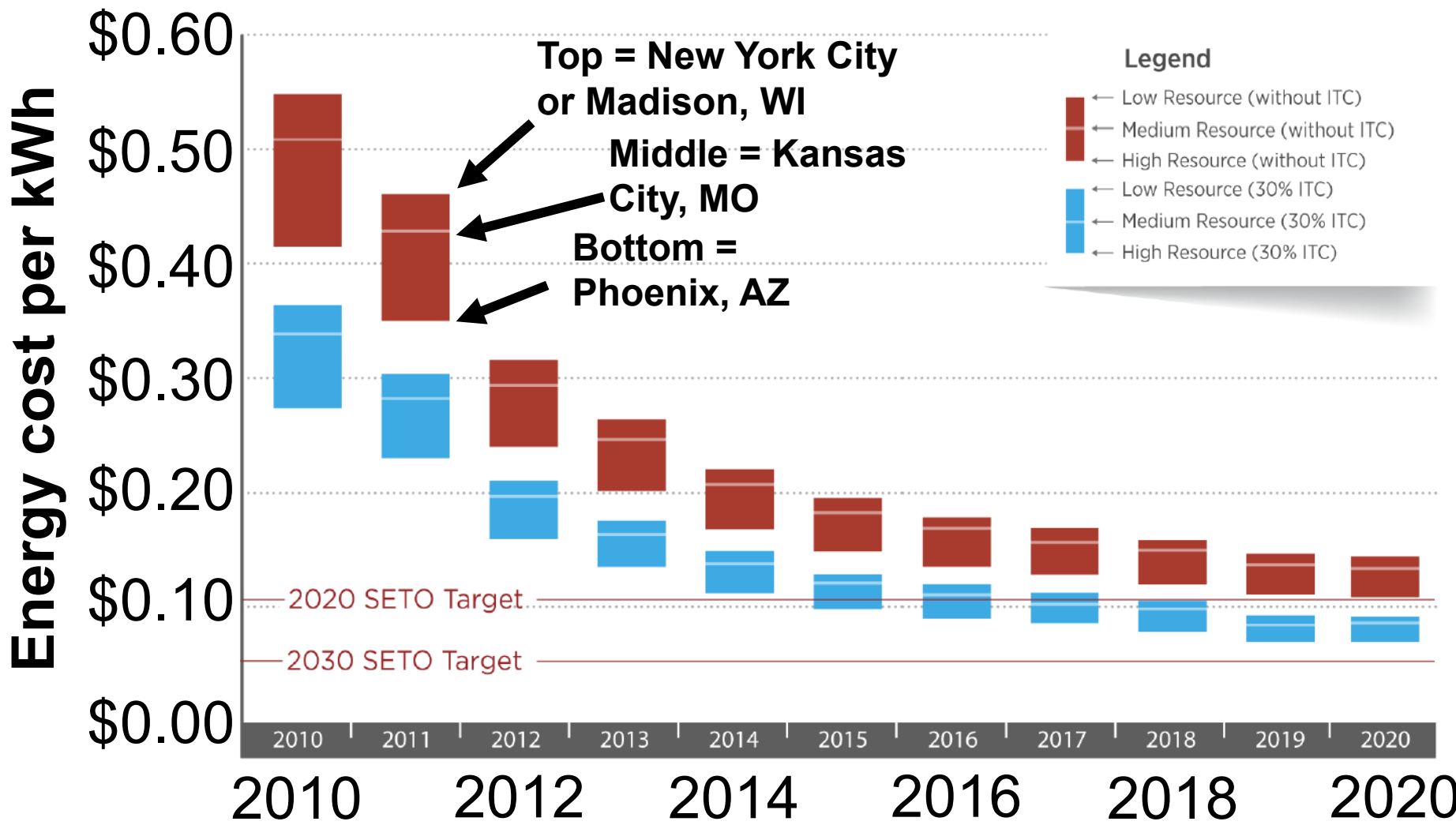


# Installed cost: *utility-scale*

Image and data from: NREL, U.S. Solar Photovoltaic System and Energy Storage Cost Benchmark: Q1 2020



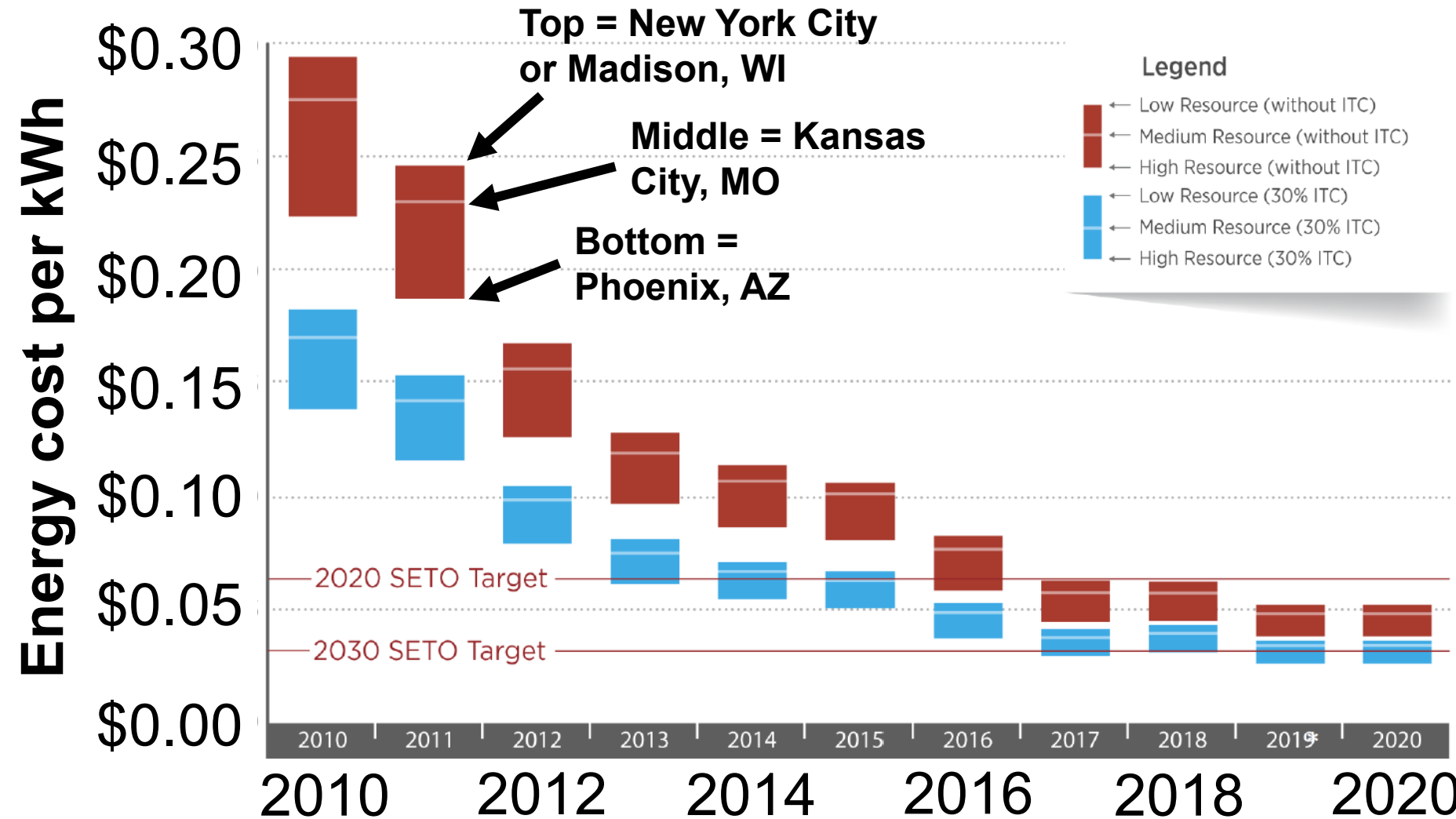
# Energy cost per kWh (30 years)



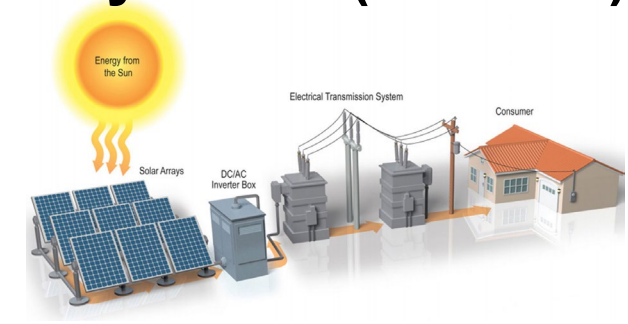
**Red = without tax credit**  
**Blue = with tax credit**

Data / image source: NREL, U.S. Solar Photovoltaic System and Energy Storage Cost Benchmark: Q1 2020

# Energy cost per kWh (30 years)



## Utility-Scale (100 MW)

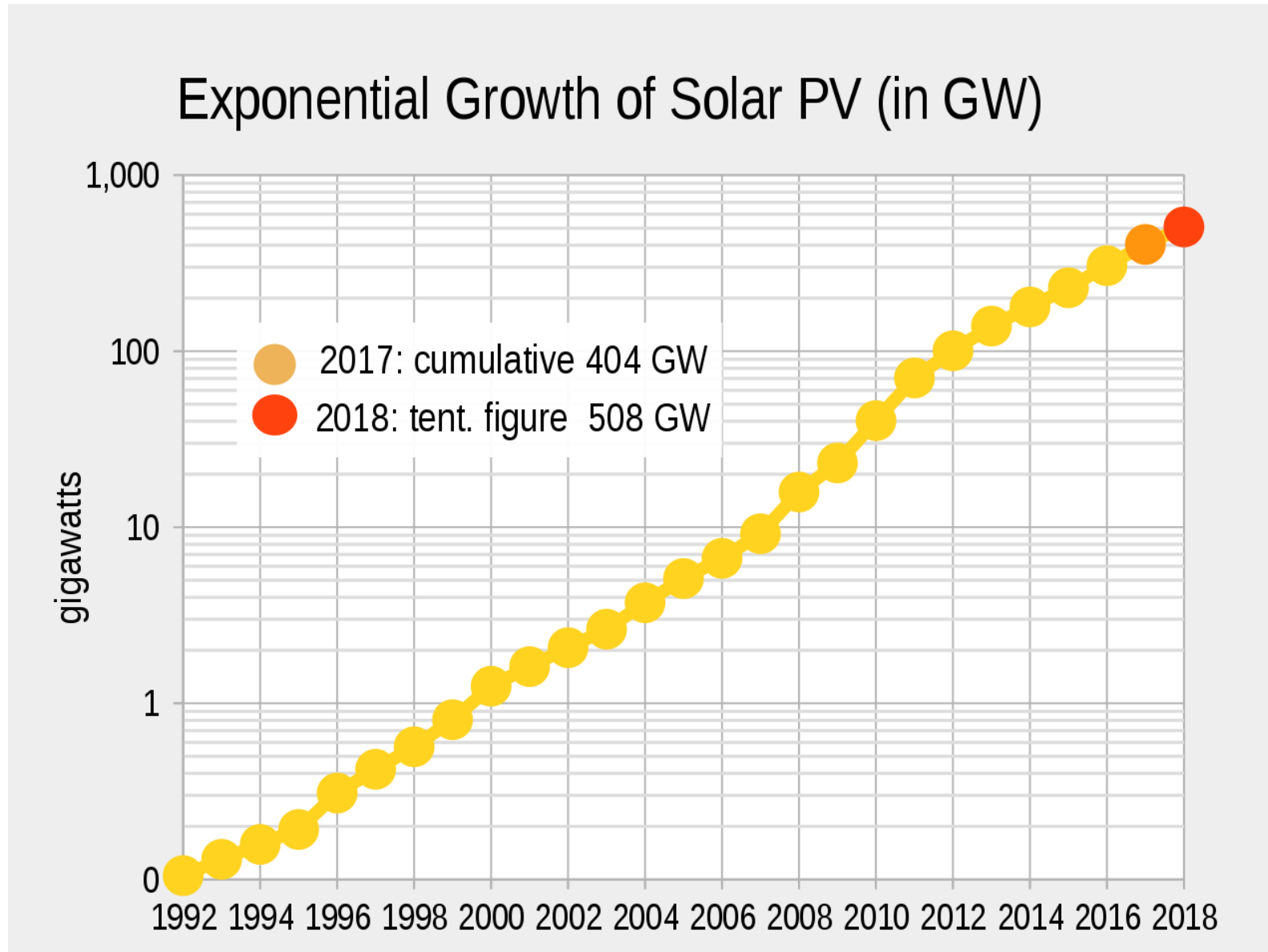


**Red = without tax credit**

**Blue = with tax credit**

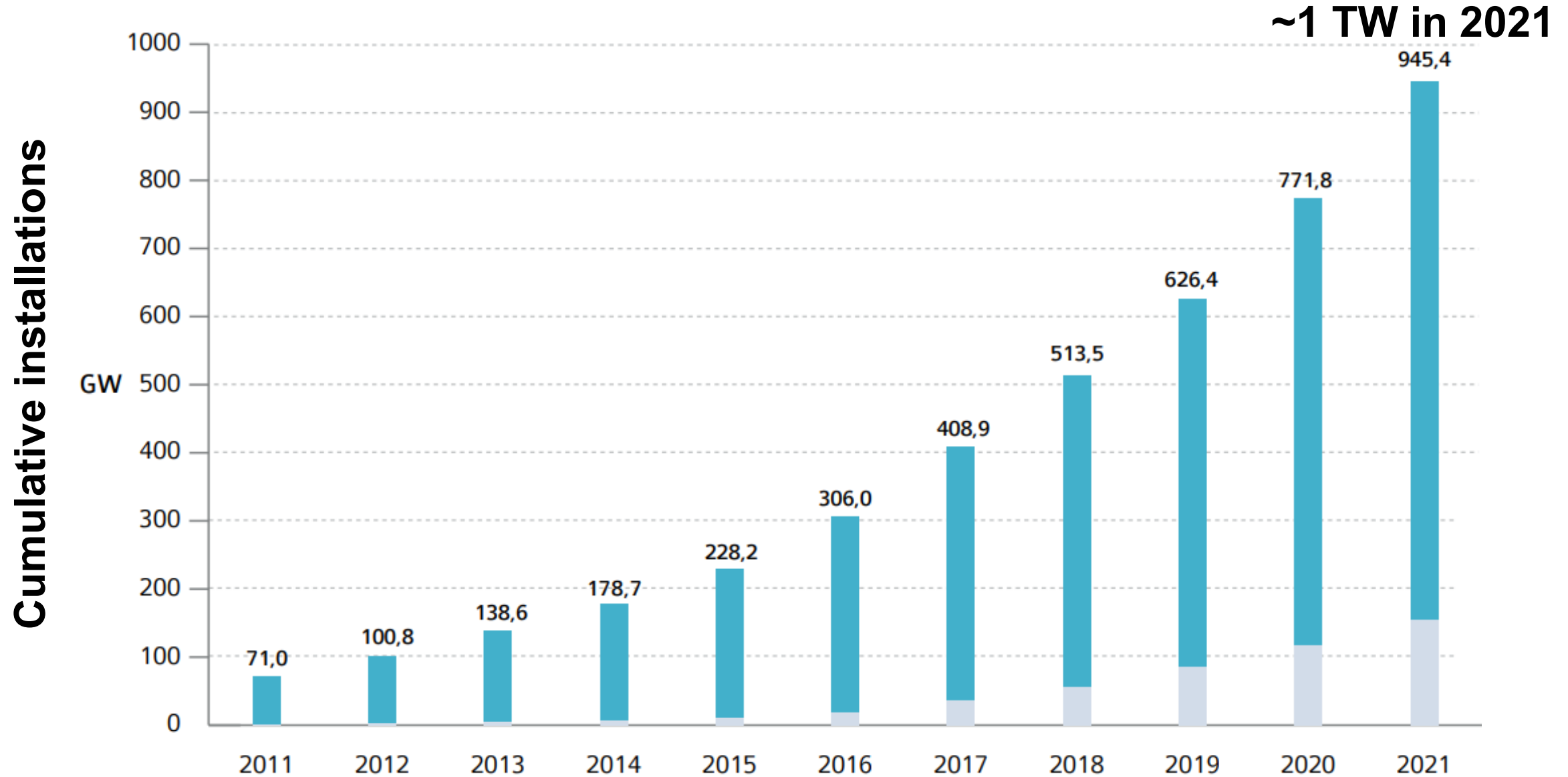
Data / image source: NREL, U.S. Solar Photovoltaic System and Energy Storage Cost Benchmark: Q1 2020

Global PV market grown exponentially in 30 years.



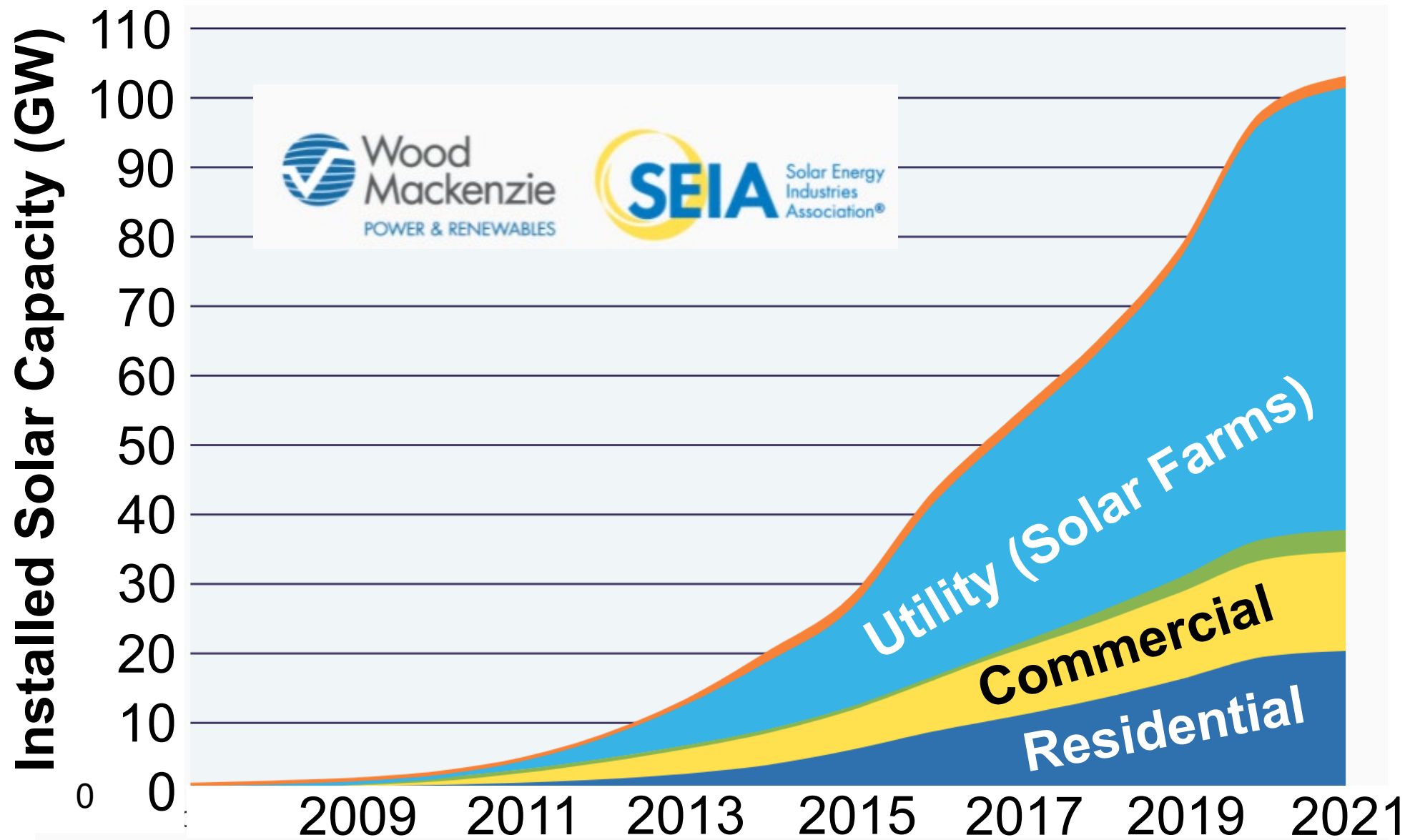
Data source: [iea-pvps.org](http://iea-pvps.org)

# Closer look at last 10 years



Data/image source: [iea-pvps.org](https://www.iea-pvps.org/)

U.S. installations have also dramatically grown.



# Reasons for dramatic growth

## ■ Capitalism and economies of scale

- Business of manufacturing and installing solar have become dramatically more efficient.



## ■ R&D

- Improved cell efficiency and decreased materials processing costs



## ■ Government policy

- Tax credits
- Inflation Reduction Act of 2022 extends ITC of 30% through 2032





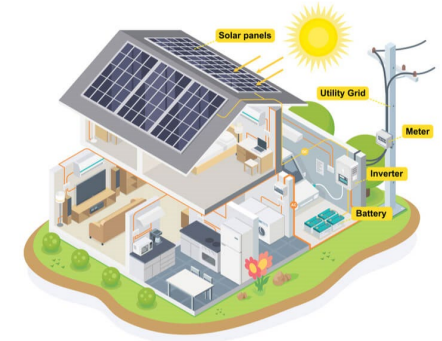
- The sun as a source of energy
- How does a photovoltaic solar cell work?
- Types of solar cells and their efficiencies
- Economics of solar cells then and now
- ➔ ■ **Solar in Wisconsin**
- The future



# Energy cost per kWh (30 years)

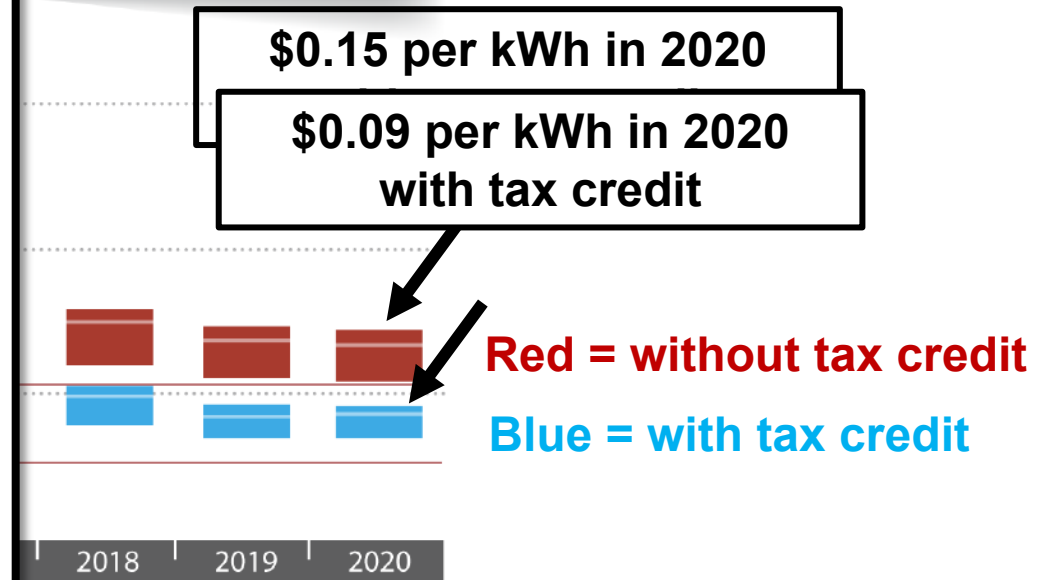
- Current MGE rate is \$0.15 per kWh.
- MGE rate will inflate with time to much greater than \$0.15 per kWh.
- Fixed cost of solar now will lock in price of \$0.09 kWh.
- Significant savings over 30 years.
- Break even point < 15 years.

## Residential



### Legend

- Low Resource (without ITC)
- Medium Resource (without ITC)
- High Resource (without ITC)
- Low Resource (30% ITC)
- Medium Resource (30% ITC)
- High Resource (30% ITC)



Data / image source: NREL, U.S. Solar Photovoltaic System and Energy Storage Cost Benchmark: Q1 2020

# Payback time of < 15 years

solarreviews.com/solar-panel-cost/wisconsin

SolarReviews Solar Calculator Reviews Your Utility Blog Learn About Solar

## Current solar panel cost near you in WI

Showing data for:

- Investment of \$10,000 now will return a "profit" of \$20,000+ in 30 years.
- Reduce pollution and save money at the same time!!!

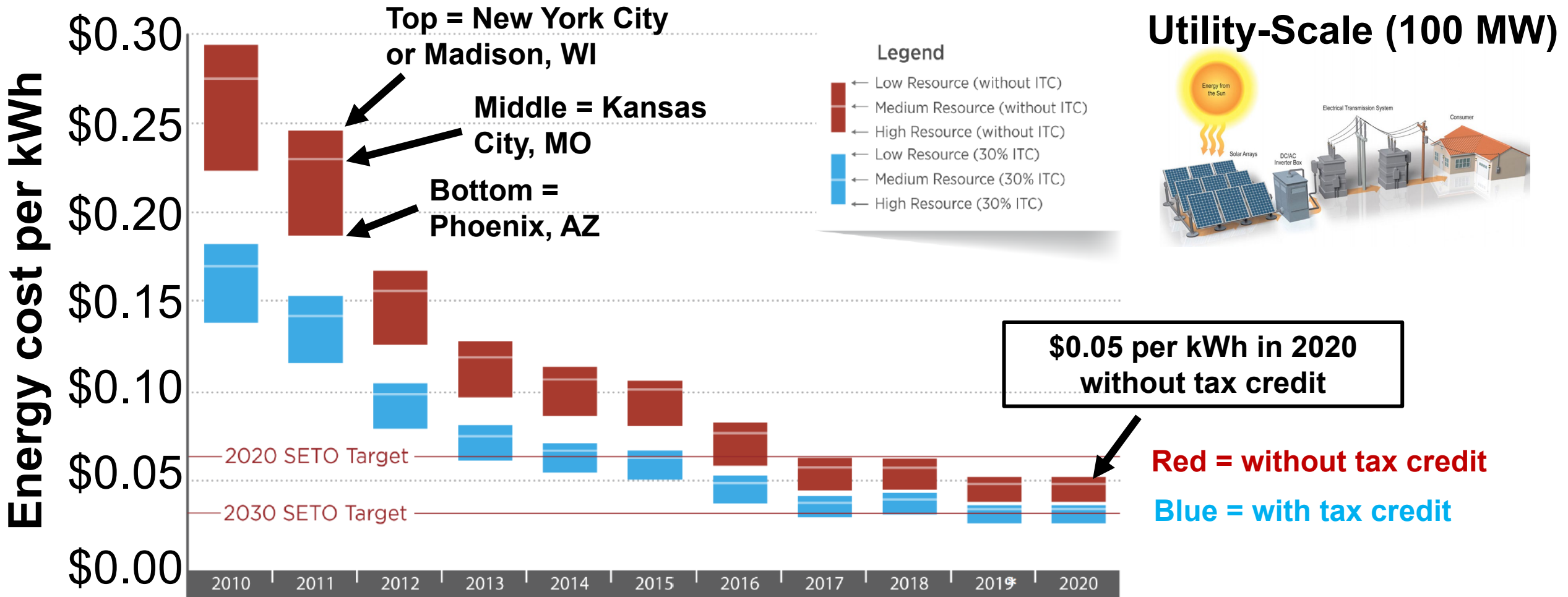
Prices based on a 5.8kW system, after 30% federal tax credit

(with tax incentives) Cost range of local prices <b>\$9,113-\$11,138</b>	Net profit (savings less system cost) <b>\$21,843-\$26,698</b>
--	---

*with tax incentives*

Payback period  
**8.8-10.8 years**

# Energy cost per kWh (30 years)



Data / image source: NREL, U.S. Solar Photovoltaic System and Energy Storage Cost Benchmark: Q1 2020

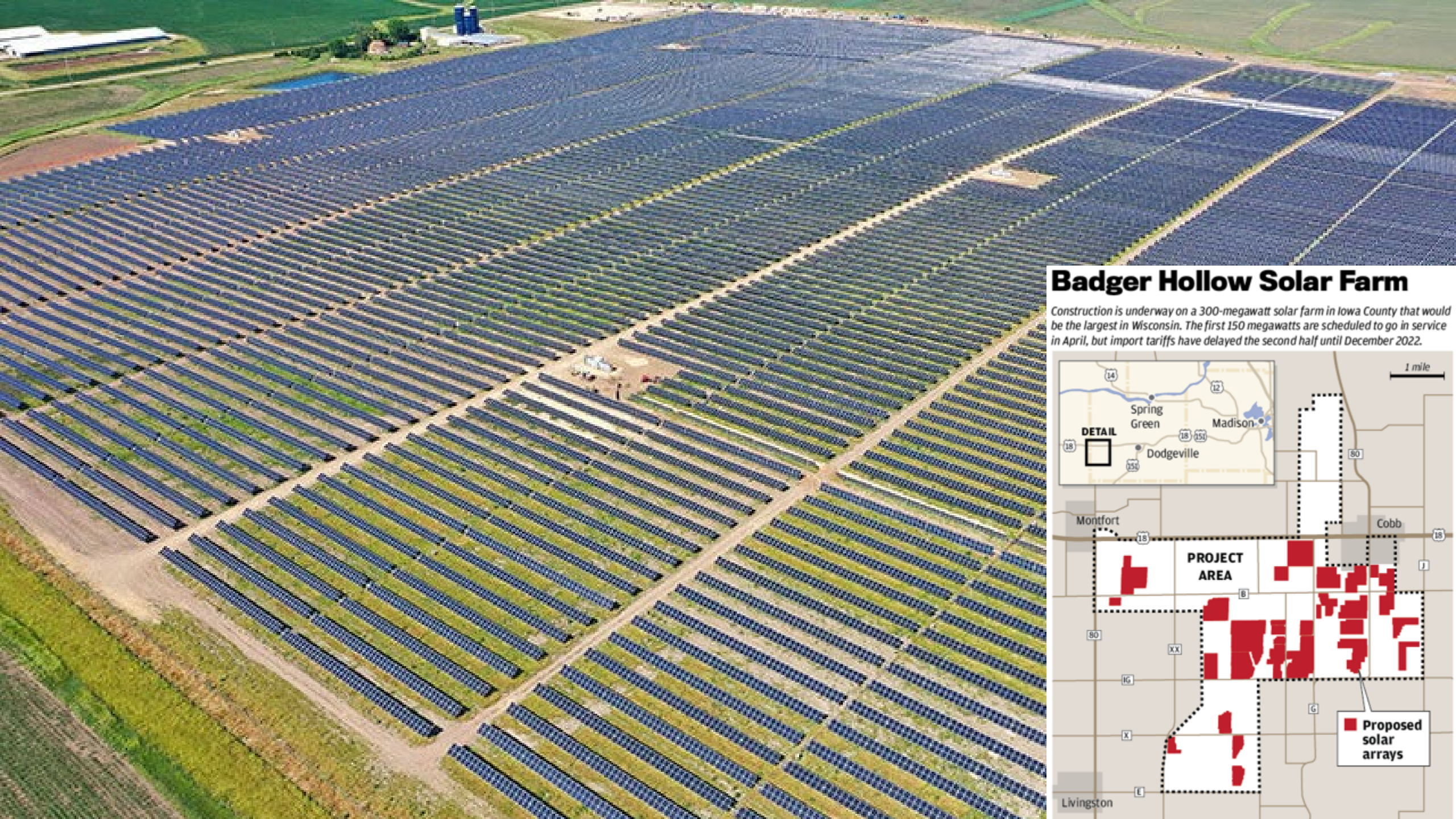
# Utility solar in development across Wisconsin

PROJECT NAME	LOCATION	SIZE OF PROJECT	PROJECT STATUS
Two Creeks Solar	Manitowoc County	150 MW	Operational November 2020
Badger Hollow Solar 1	Iowa County	150 MW	Operational December 2021
Badger Hollow Solar 2	Iowa County	150 MW	Online March 2023
Point Beach Solar	Manitowoc County	100 MW	Operational September 2021
Wood County Solar	Wood County	149 MW	Online November 2022
Bear Creek Solar	Richland County	49 MW	Online October 2022
North Rock Solar	Rock County	50 MW	Online March 2023
Paris Solar	Kenosha County	200 MW	Online June 2023

Alliant Energy has proposed 1100 MW and WEC+MGE 350 MW of solar utility projects.

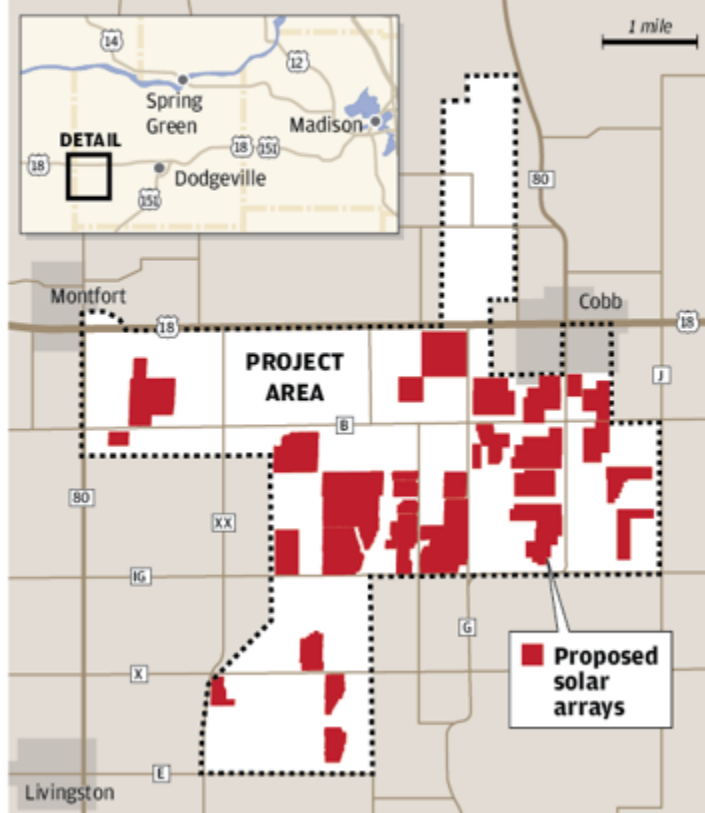






## Badger Hollow Solar Farm

Construction is underway on a 300-megawatt solar farm in Iowa County that would be the largest in Wisconsin. The first 150 megawatts are scheduled to go in service in April, but import tariffs have delayed the second half until December 2022.



- The sun as a source of energy
- How does a photovoltaic solar cell work?
- Types of solar cells and their efficiencies
- Economics of solar cells then and now
- Solar in Wisconsin
- ➔ ■ **The future**



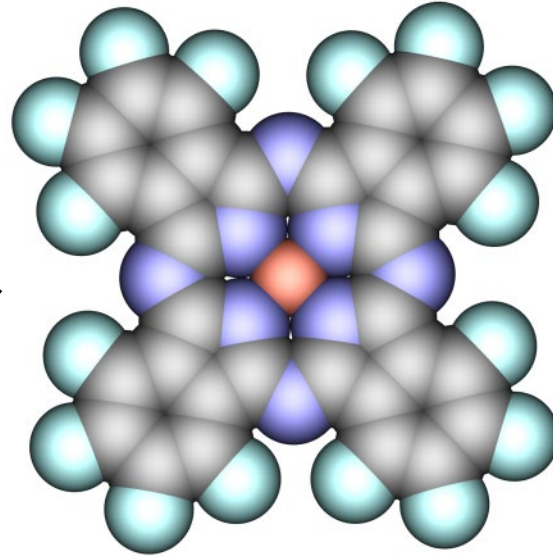


# Looking forward

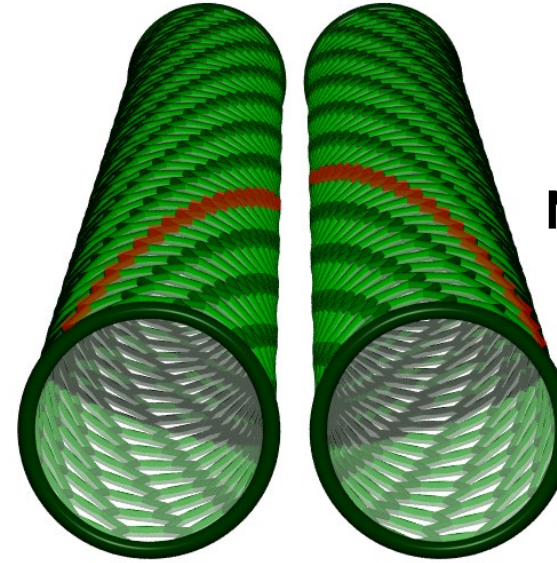
- Economies of scale will continue to push down price.
- R&D of new materials and processes will continue to improve the \$ / W ratio.
- Challenges
  - Energy storage
  - Existing fossil fuel infrastructure already in place... will take time to transition and retire existing plants
  - Land usage conflicts

# Exploratory cells: new materials

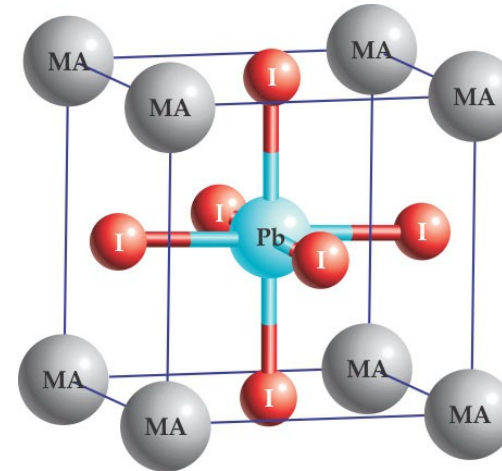
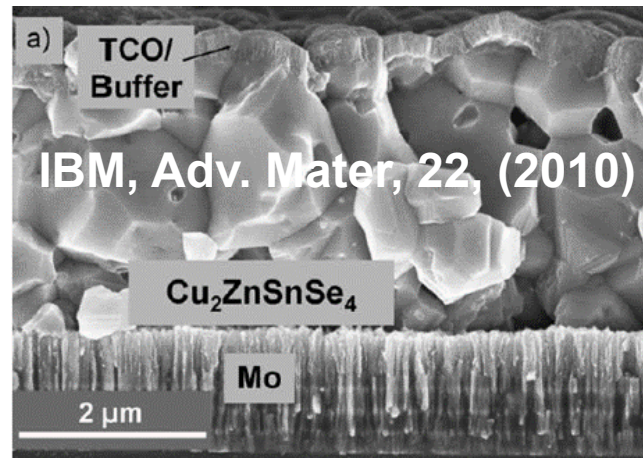
**Organic molecules & polymers**



**Nanotechnology  
(quantum dots  
and carbon  
nanotubes)**

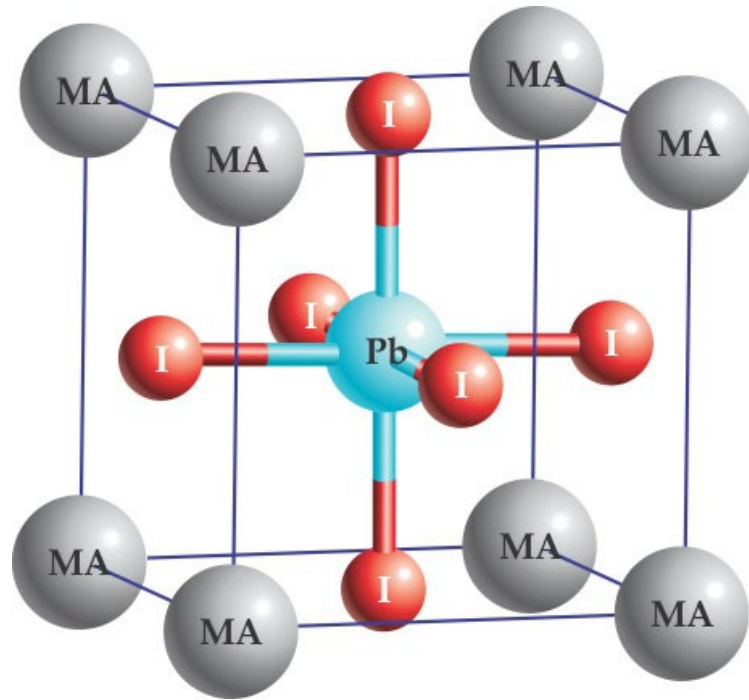


**New  
inorganic  
semi-  
conductors**



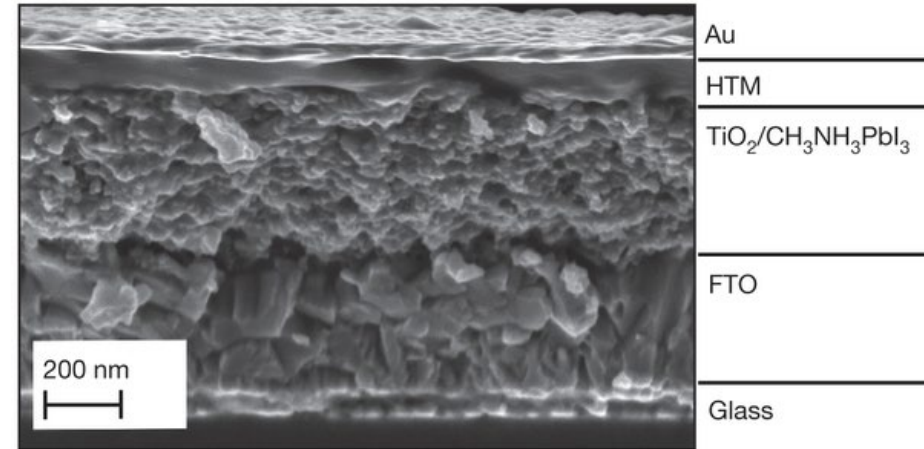
**Inorganic  
/organic  
hybrid  
perovskites**

# Organic-inorganic hybrid perovskites



MA = methylammonium

<http://scitation.aip.org/docserver/fulltext/pt.5.7058figure1.jpg>



[http://www.nature.com/nature/journal/v499/n7458/images\\_article/nature12340-f2.jpg](http://www.nature.com/nature/journal/v499/n7458/images_article/nature12340-f2.jpg)

- Discovered just a few years ago
- Solution-processable → Inexpensive
- >20% already!!!
- Stability and lifetime still an issue

# Economic conclusions

- Technology has rapidly matured.
- Scale of manufacturing has grown exponentially.
- Price has plummeted.
- Solar is now cost- competitive or superior.



**The Sun Is Ready to Make Your Electricity  
Greener and Cheaper in Wisconsin**

Thank you!

■ **Wisconsin specific resources**

- REWEW Wisconsin  
[www.renewwisconsin.org/solar-energy/](http://www.renewwisconsin.org/solar-energy/)
- Solar Energy Industries Association (SEIA)  
<https://www.seia.org/state-solar-policy/wisconsin-solar>

■ **More information on solar technologies**

- Solar Photovoltaics Technology Basics  
<https://www.energy.gov/eere/solar/solar-photovoltaic-technology-basics>
- Copy of these slides  
<https://arnold.engr.wisc.edu/outreach.html>





# Science and engineering summary

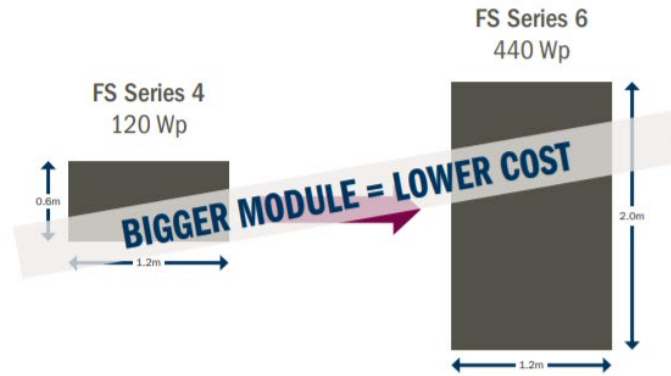
- The active materials in a photovoltaic solar cell are semiconductors.
- A junction between P-doped and N-doped semiconductors (PN junction) is used to separate the positive and negative charges generated by light → electricity.
- Single- and polycrystalline single-junction Si and polycrystalline thin film single-junction CdTe are currently the most commonly installed photovoltaic technologies.
- Efficiency is about 20%.

# First Solar Thin Film CdTe Series 4 versus Series 6... Improvements in manufacturing and efficiency

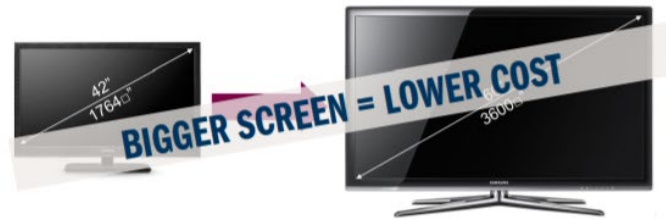
## THIN FILM SCALES

- Unit of process for CdTe is the glass; scaling benefit
- Unit of process for c-Si is the wafer; no scaling benefit
- Same manufacturing process as the previous technology
- Highest power utility-scale modules in the market

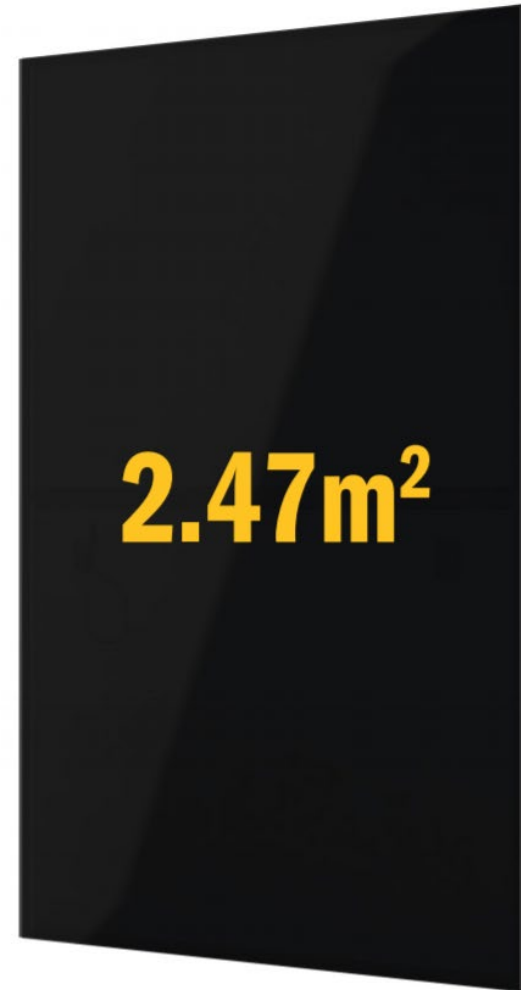
Thin film cost scales non-linearly...



...similar to LCD screens.



No cost benefit from scaling c-Si



Source: FirstSolar.com



# What makes a semiconductor a good semiconductor?

- Proper band gap
  - ◆ Determines energies of light waves absorbed and PV voltage
- Abundance of materials
- Economics of processing / scale-up / toxicity / cost of disposal
- “Speed” of charges
- Thermal and chemical stability / lifetime / reliability
- Electronic nature of atomic-scale defects and imperfections in semiconductors

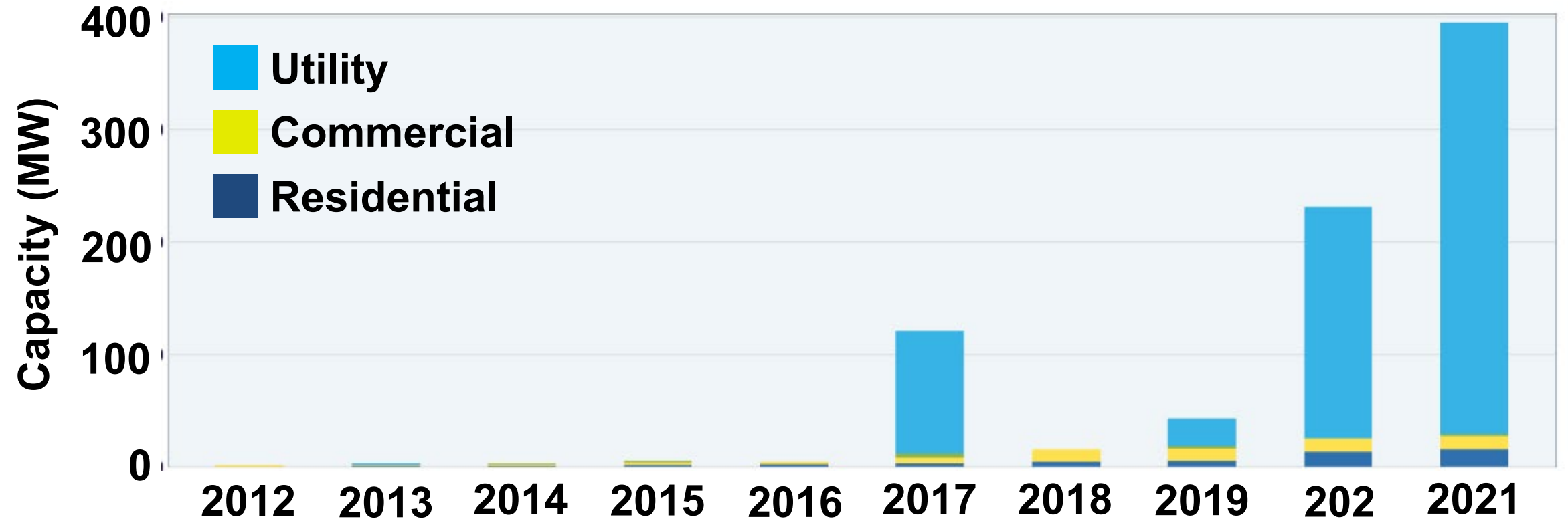
# Is there enough land? Yes.

- “Wisconsin could produce about 50% of our annual electricity needs through the use of solar panels on only 125,000 acres of land.
- For comparison, according to the USDA, as of 2021 the total land in farms for Wisconsin was approximately 14,300,000 acres.”

<https://www.renewwisconsin.org/solar-energy/>



# Result of favorable economics: Solar installations rapidly growing in WI!



# What about storage?

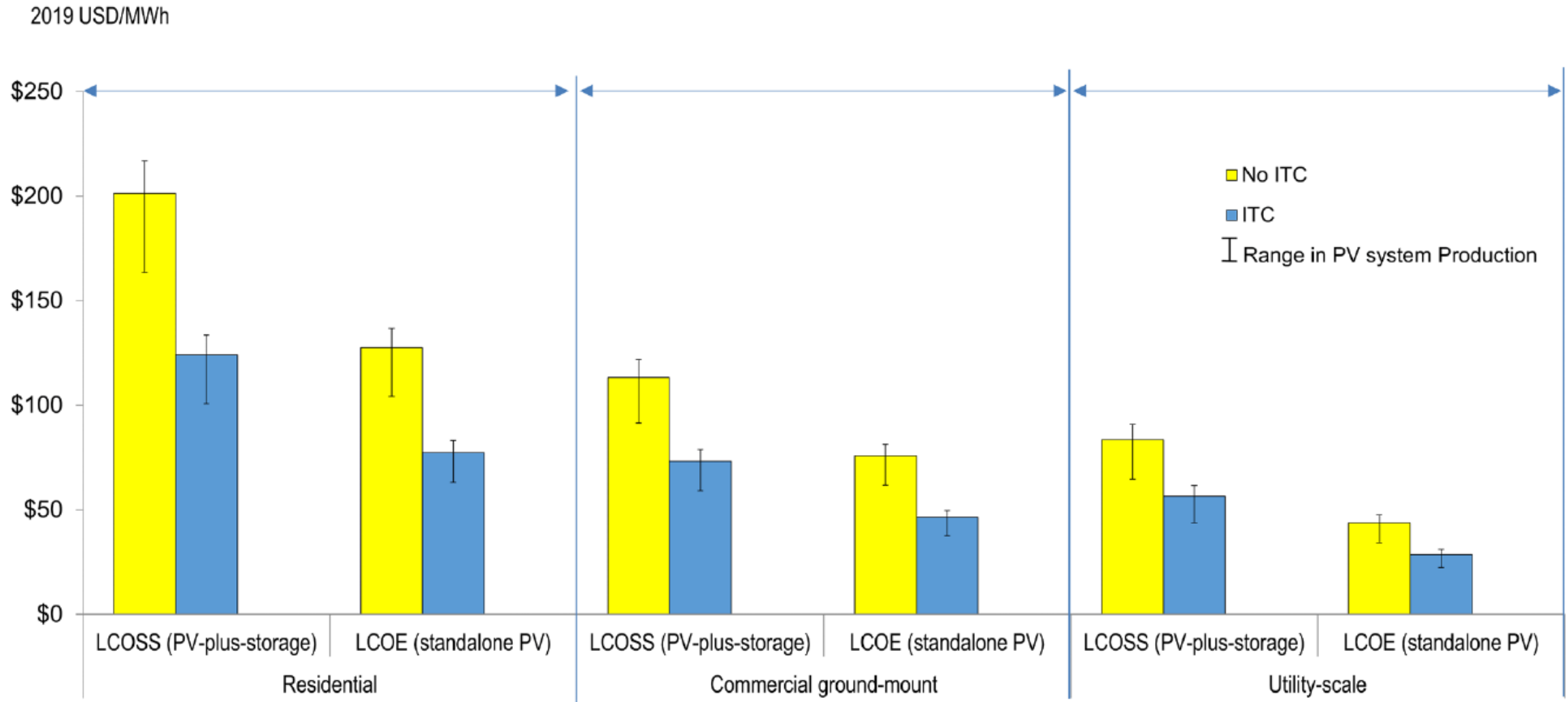
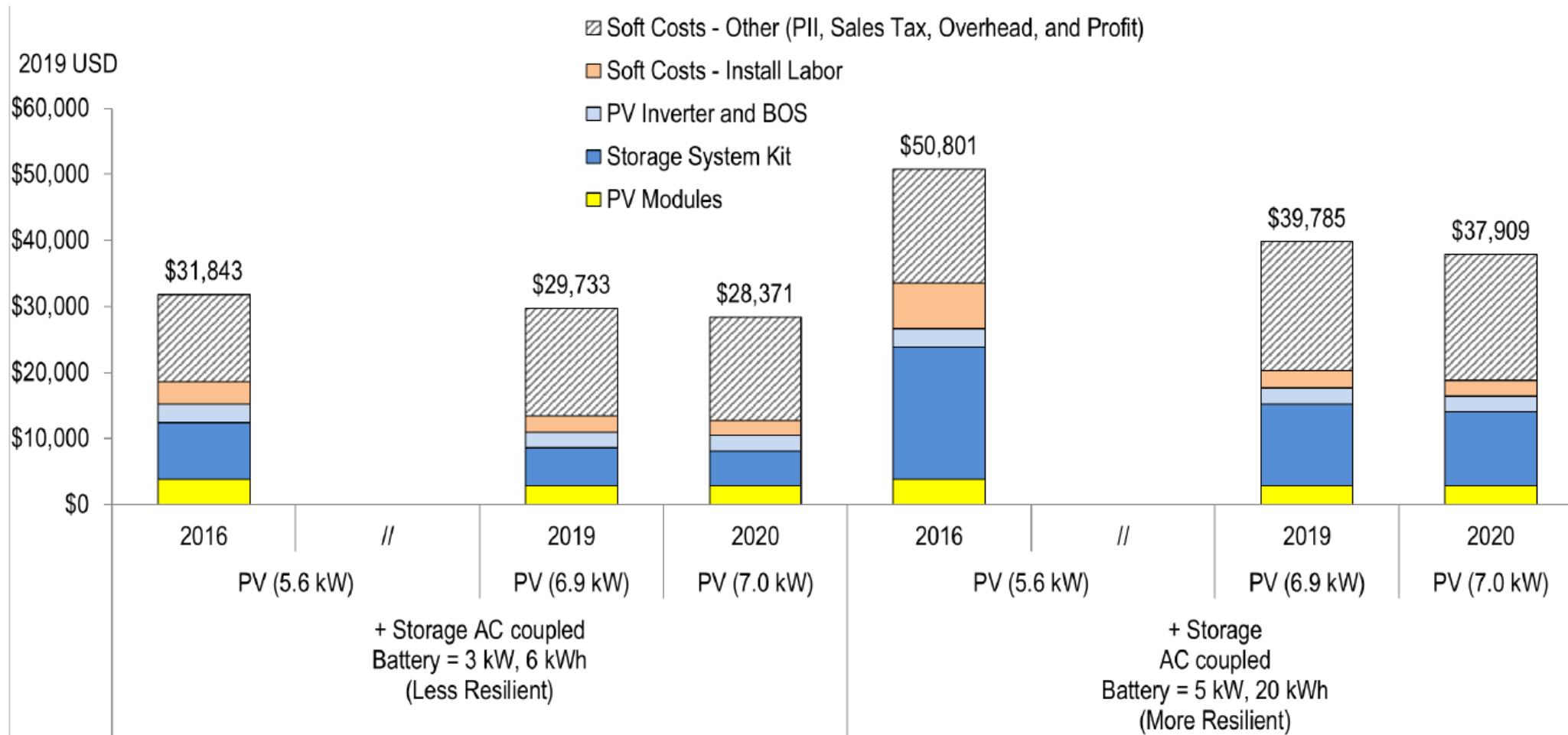


Image and data from: NREL, U.S. Solar Photovoltaic System and Energy Storage Cost Benchmark: Q1 2020

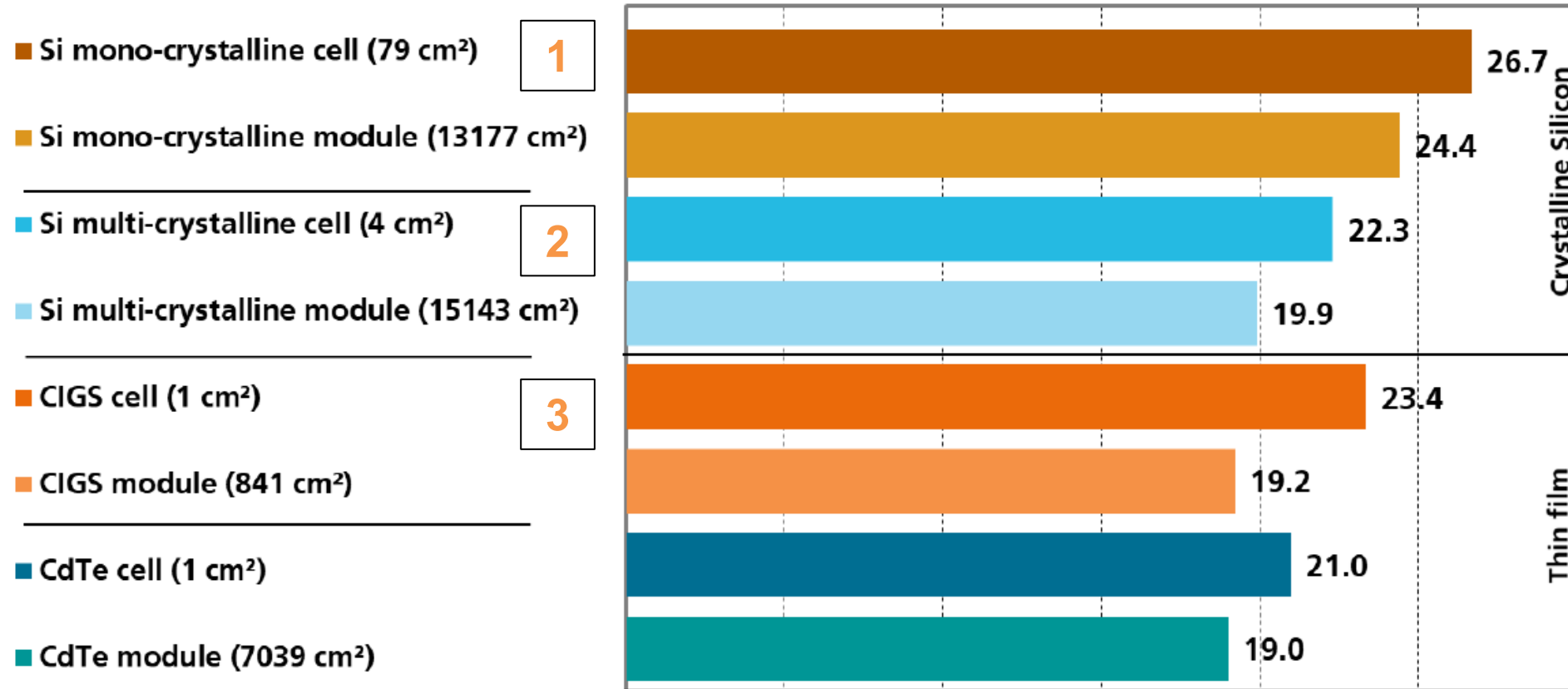
# Storage costs decreasing with time as well...



**Figure ES-5. Residential PV-plus-storage system cost benchmark summary (inflation-adjusted), 2016, 2019, and 2020**

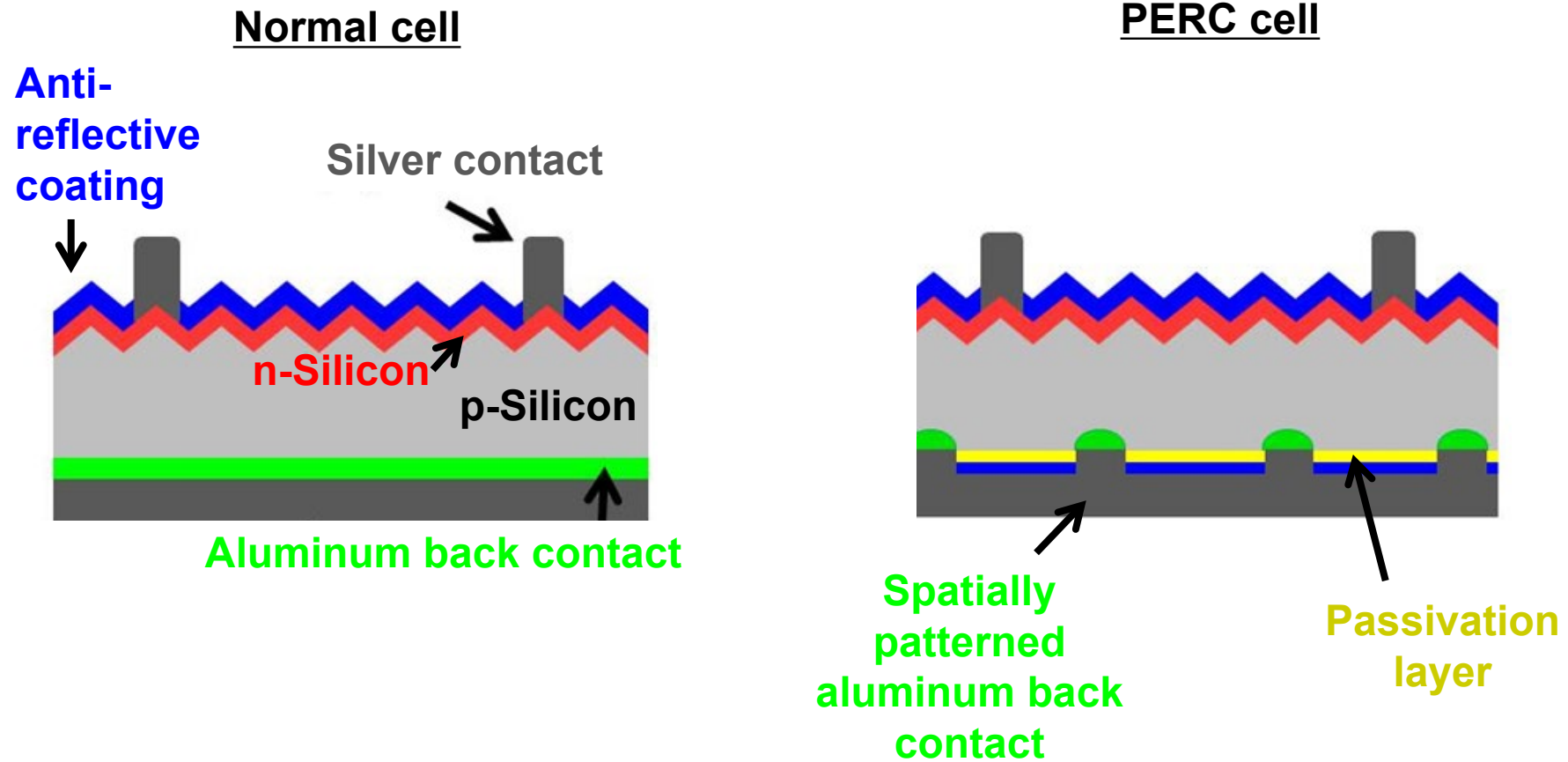
Image and data from: NREL, U.S. Solar Photovoltaic System and Energy Storage Cost Benchmark: Q1 2020

# More Detailed Comparison



<https://www.ise.fraunhofer.de/content/dam/ise/de/documents/publications/studies/Photovoltaics-Report.pdf>

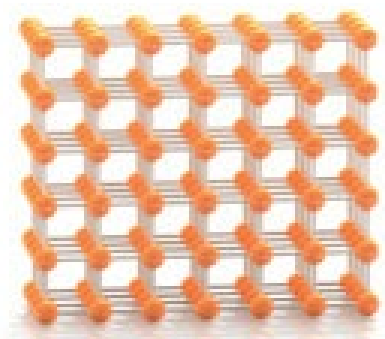
# Passivated Emitter Rear Contact (PERC) Single Crystal Silicon Cells



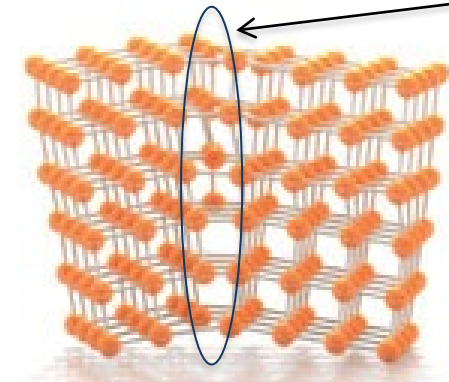
Source: Institute for Solar Energy Research Hamelin (ISFH)

# Crystallinity

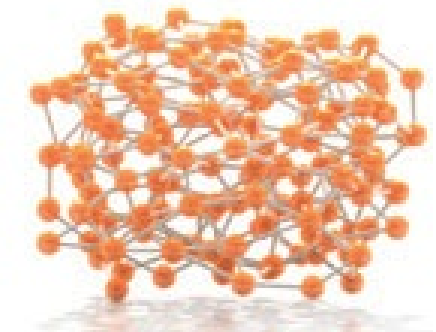
Defects slow charges and cause losses.



*Crystalline*



*Polycrystalline*



*Amorphous*

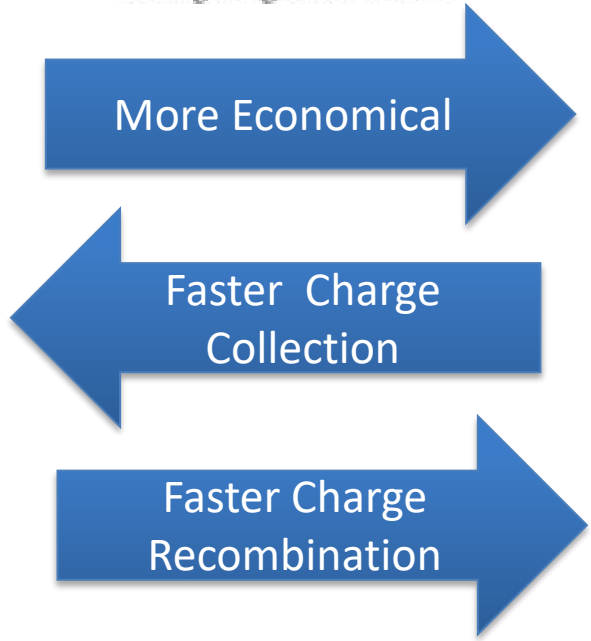
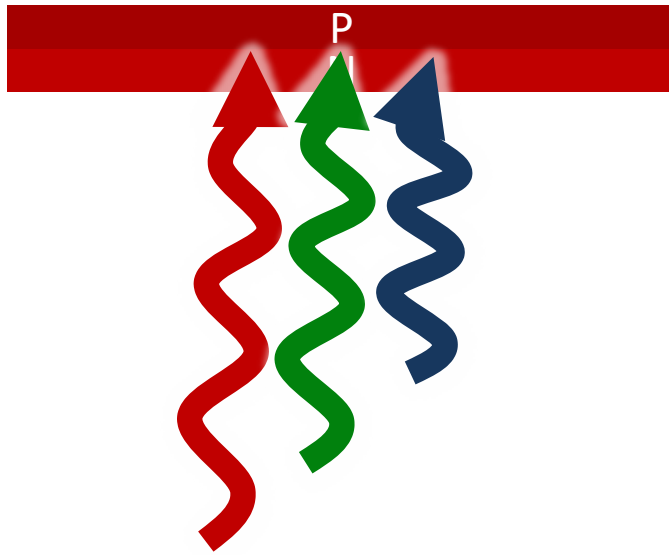


Image from: <http://spectrum.ieee.org/image/1838375>



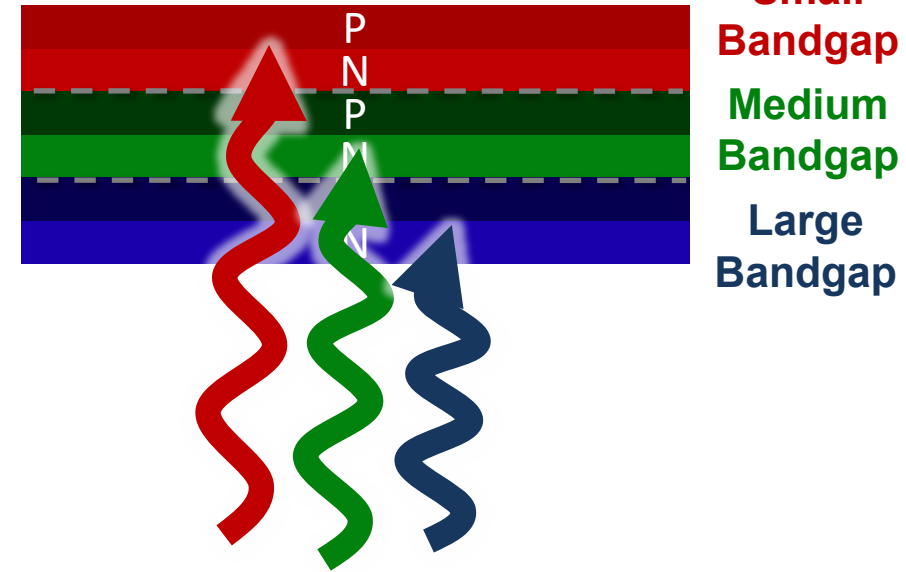
# Number of Junctions

Single-Junction:



- ❑ When the photon energy  $>$  band gap, extra energy is wasted.
- ❑ For example, a blue photon produces same energy as red photon).
- ❑ Maximum efficiency  $\sim 30\%$ .

Multi-Junction:

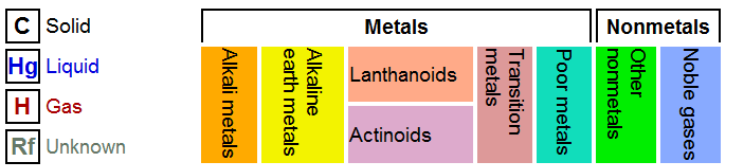


- ❑ In multi-junction cell, blue photon can produce more energy than red photon.
- ❑ Maximum efficiency  $\sim 50\%$  (for 3 cells).
- ❑ More complex fabrication  $\rightarrow$  More \$\$\$.

# Composition (Different Semiconductors)

## Periodic Table of Elements

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1	<b>H</b> Hydrogen 1.00794																	<b>He</b> Helium 4.002602	
2	<b>Li</b> Lithium 6.941	<b>Be</b> Beryllium 9.012182																	<b>Ne</b> Neon 20.1797
3	<b>Na</b> Sodium 22.98976928	<b>Mg</b> Magnesium 24.3050																	<b>Ar</b> Argon 39.948
4	<b>K</b> Potassium 39.0983	<b>Ca</b> Calcium 40.078	<b>Sc</b> Scandium 44.955912	<b>Ti</b> Titanium 47.887	<b>V</b> Vanadium 50.9415	<b>Cr</b> Chromium 51.9961	<b>Mn</b> Manganese 54.938045	<b>Fe</b> Iron 55.845	<b>Co</b> Cobalt 58.933195	<b>Ni</b> Nickel 58.6934	<b>Cu</b> Copper 63.546	<b>Zn</b> Zinc 65.38	<b>Ga</b> Gallium 69.723	<b>Ge</b> Germanium 72.64	<b>As</b> Arsenic 74.92160	<b>Se</b> Selenium 78.96	<b>Br</b> Bromine 79.904	<b>Kr</b> Krypton 83.798	
5	<b>Rb</b> Rubidium 85.4678	<b>Sr</b> Strontium 87.62	<b>Y</b> Yttrium 88.90585	<b>Zr</b> Zirconium 91.224	<b>Nb</b> Niobium 92.90638	<b>Mo</b> Molybdenum 95.96	<b>Tc</b> Technetium (97.9072)	<b>Ru</b> Ruthenium 101.07	<b>Rh</b> Rhodium 102.90550	<b>Pd</b> Palladium 106.42	<b>Ag</b> Silver 107.8682	<b>Cd</b> Cadmium 112.411	<b>In</b> Indium 114.818	<b>Sn</b> Tin 118.710	<b>Sb</b> Antimony 121.760	<b>Te</b> Tellurium 127.60	<b>I</b> Iodine 126.90447	<b>Xe</b> Xenon 131.293	
6	<b>Cs</b> Cesium 132.9054519	<b>Ba</b> Barium 137.327	57-71	<b>Hf</b> Hafnium 178.49	<b>Ta</b> Tantalum 180.94788	<b>W</b> Tungsten 183.84	<b>Re</b> Rhenium 186.207	<b>Os</b> Osmium 190.23	<b>Ir</b> Iridium 192.217	<b>Pt</b> Platinum 195.084	<b>Au</b> Gold 196.966569	<b>Hg</b> Mercury 200.59	<b>Tl</b> Thallium 204.3833	<b>Pb</b> Lead 207.2	<b>Bi</b> Bismuth 208.98040	<b>Po</b> Polonium (209)	<b>At</b> Astatine (209, 8671)	<b>Rn</b> Radon (222, 0176)	
7	<b>Fr</b> Francium (223)	<b>Ra</b> Radium (226)	89-103	<b>Rf</b> Rutherfordium (261)	<b>Db</b> Dubnium (262)	<b>Sg</b> Seaborgium (266)	<b>Bh</b> Bohrium (264)	<b>Hs</b> Hassium (277)	<b>Mt</b> Meitnerium (268)	<b>Ds</b> Darmstadtium (271)	<b>Rg</b> Roentgenium (272)	<b>Uub</b> Ununbium (285)	<b>Uut</b> Ununtrium (284)	<b>Uuq</b> Ununquadium (289)	<b>Uup</b> Ununpentium (288)	<b>Uuh</b> Ununhexium (282)	<b>Uus</b> Ununseptium	<b>Uuo</b> Ununoctium (284)	



For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com) http://www.ptable.com/

<b>57 La</b> Lanthanum 138.90547	<b>58 Ce</b> Cerium 140.116	<b>59 Pr</b> Praseodymium 140.90765	<b>60 Nd</b> Neodymium 144.242	<b>61 Pm</b> Promethium (145)	<b>62 Sm</b> Samarium 150.36	<b>63 Eu</b> Europium 151.964	<b>64 Gd</b> Gadolinium 157.25	<b>65 Tb</b> Terbium 158.92535	<b>66 Dy</b> Dysprosium 162.500	<b>67 Ho</b> Holmium 164.93032	<b>68 Er</b> Erbium 167.259	<b>69 Tm</b> Thulium 168.93421	<b>70 Yb</b> Ytterbium 173.054	<b>71 Lu</b> Lutetium 174.9668
<b>89 Ac</b> Actinium (227)	<b>90 Th</b> Thorium 232.03806	<b>91 Pa</b> Protactinium 231.03688	<b>92 U</b> Uranium 238.02891	<b>93 Np</b> Neptunium (237)	<b>94 Pu</b> Plutonium (244)	<b>95 Am</b> Americium (243)	<b>96 Cm</b> Curium (247)	<b>97 Bk</b> Berkelium (247)	<b>98 Cf</b> Californium (251)	<b>99 Es</b> Einsteinium (252)	<b>100 Fm</b> Fermium (257)	<b>101 Md</b> Mendelevium (258)	<b>102 No</b> Nobelium (259)	<b>103 Lr</b> Lawrencium (262)



# Thick Crystal or Thin Film

*Thin-Film*



~ 5  $\mu\text{m}$  (~0.2 mils)

- Very thin films possible if semiconductor absorbs light strongly (depends on composition).
- Uses less material.
- Requires supporting substrate.
- Typically polycrystalline.

*Thick Crystal*



~ 500  $\mu\text{m}$  (~20 mils)

- ◆ Self-supporting.
- ◆ Single crystalline.

# Multi-Junction Solar Cells

## Crystallinity

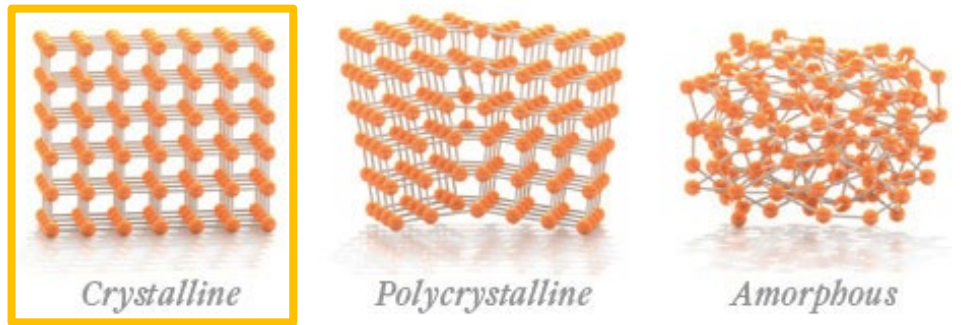


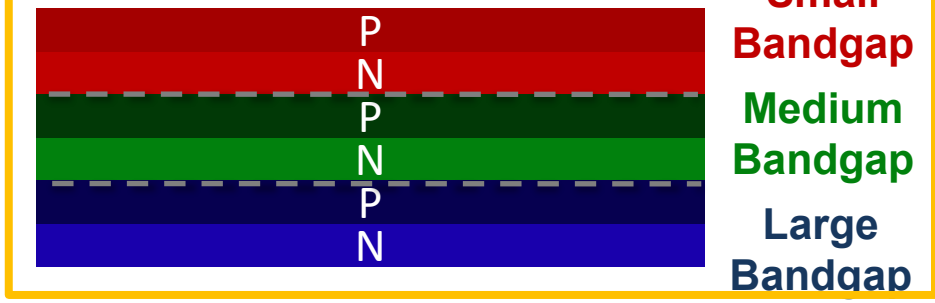
Image from: <http://spectrum.ieee.org/image/1838375>

## Number of Junctions

Single-Junction:



Multi-Junction:



**Small Bandgap**  
**Medium Bandgap**  
**Large Bandgap**

## Thick Crystal or Thin Film

Thin-Film



Thick Crystal



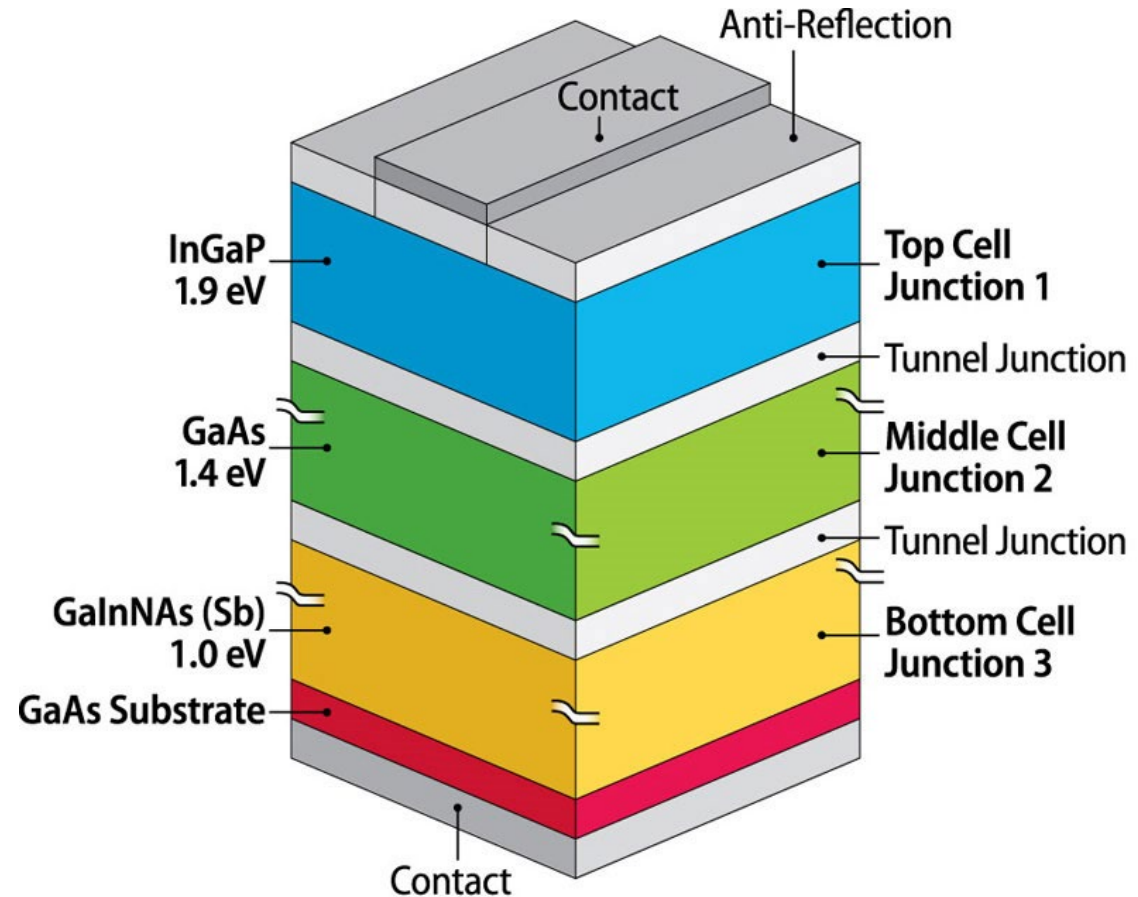
## Composition

**Periodic Table of Elements**

Design and Interface Copyright © 1997 Michael Dayah (michael@dayah.com). <http://www.ptable.com/>

# Example: Near-World Record Multi-Junction Cell

- ❑ Solar Junctions
- ❑ 43.5% (2012)
- ❑ Recent advances in efficiency due to ability to grow new unique combinations of materials on top of one-another.



from: <http://www.nrel.gov/continuum/spectrum/awards.cfm>

# Disadvantage: Very Costly

Many layers of P- and N- doped III,V semiconductors

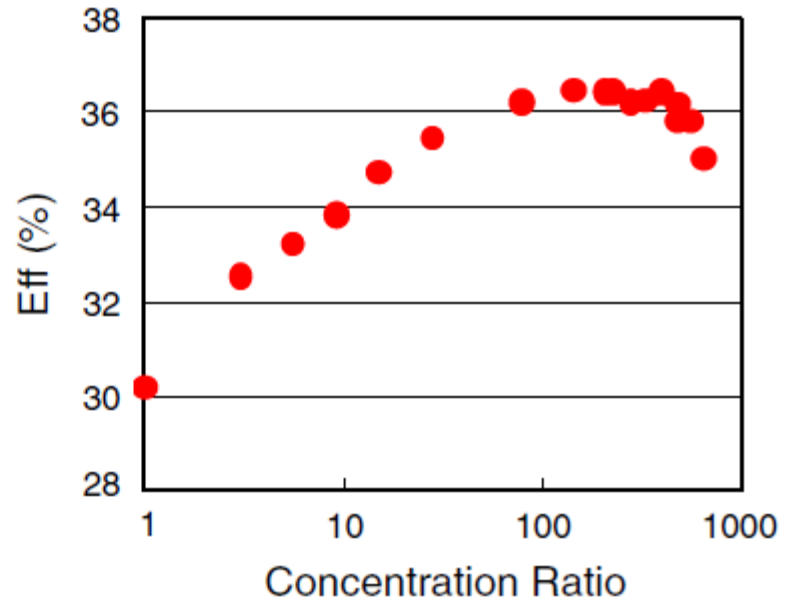
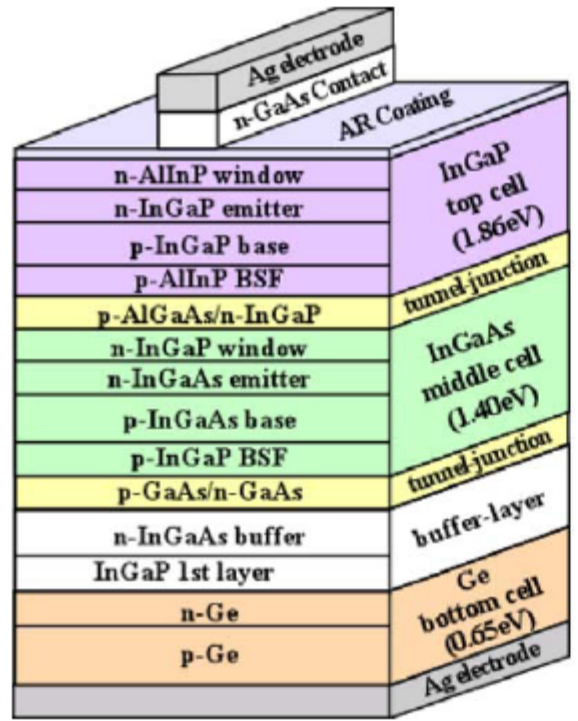


Fig. 5. A schematic cross section of a high-efficiency InGaP/InGaAs/Ge 3-junction solar cell and efficiency of a concentrator cell as a function of concentration ratio.

■ Higher efficiency but much higher cost!

Available online at [www.sciencedirect.com](http://www.sciencedirect.com)  
 ScienceDirect  
 Solar Energy Materials & Solar Cells 91 (2006) 361–377  
 www.elsevier.com/locate/solmat

Super high-efficiency multi-junction and concentrator solar cells  
 Masafumi Yamaguchi<sup>a,\*</sup>, Tatsuya Takamoto<sup>b</sup>, Kenji Araki<sup>b,c</sup>  
<sup>a</sup>Toyota Technological Institute, 2-12-1 Hirakata, Temaku, Nagoya 461-8511, Japan  
<sup>b</sup>Solar Energy Conversion, 2-1-1 Higashi, Sendai, Japan 981-1201, Japan  
<sup>c</sup>Daikin World Corporation, 2-19-1 Daikoku, Maibara, Nagoya 471-8542, Japan  
 Available online 24 July 2006

# Possible Solution: Concentrators

- Reduce cost by using lenses to focus light from large areas into smaller solar cells

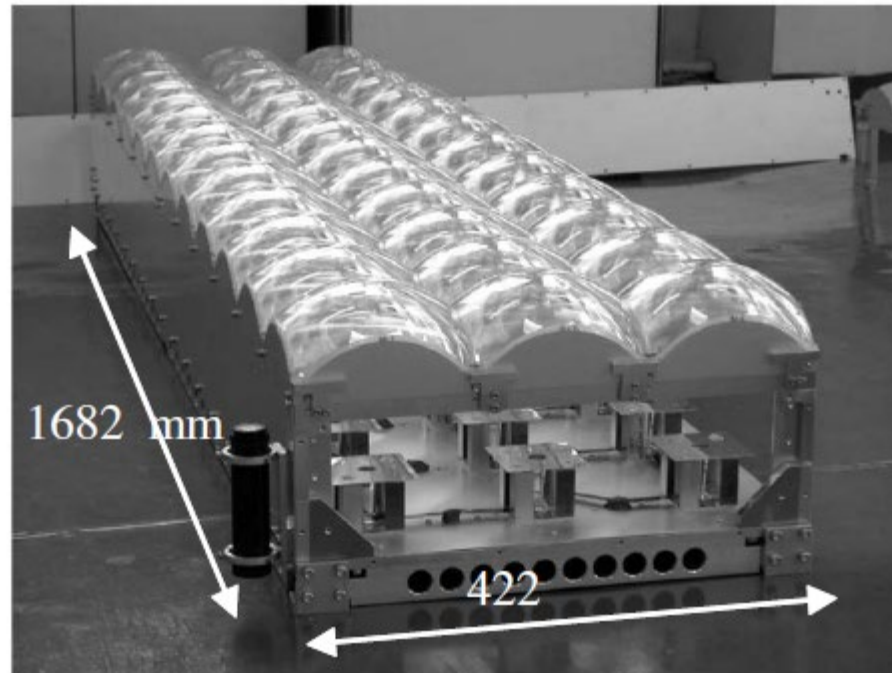
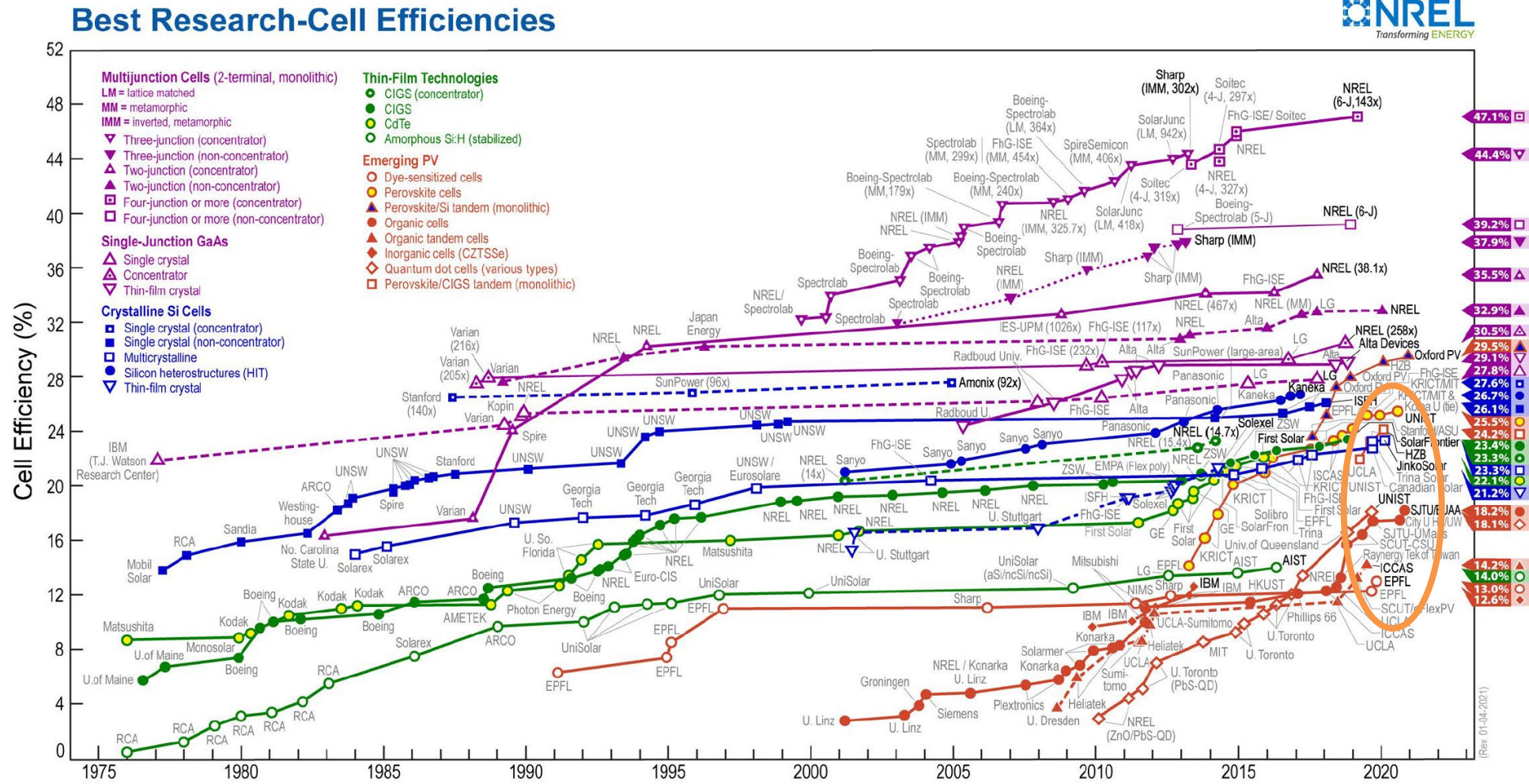


Fig. 6.  $7000\text{cm}^2$  and 400X concentrator module with the 36 receivers connected in series and dome-shaped Fresnel lens made by injection mold.

- Must track the sun to enable focusing → expensive.

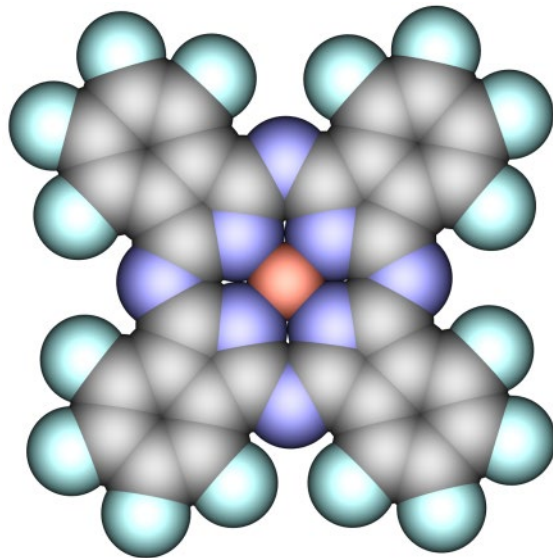
# Highly Exploratory Cells



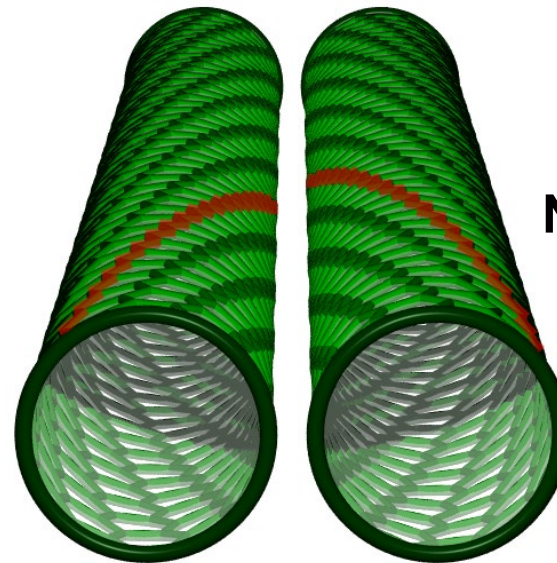


# Highly Exploratory Cells: New Materials

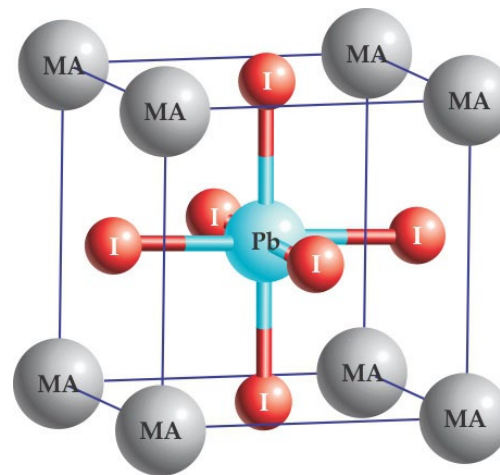
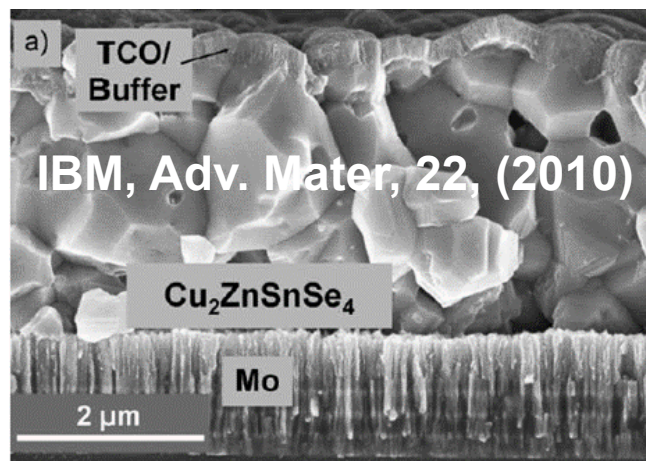
**Organic molecules & polymers**



**Nanotechnology  
(quantum dots  
and carbon  
nanotubes)**

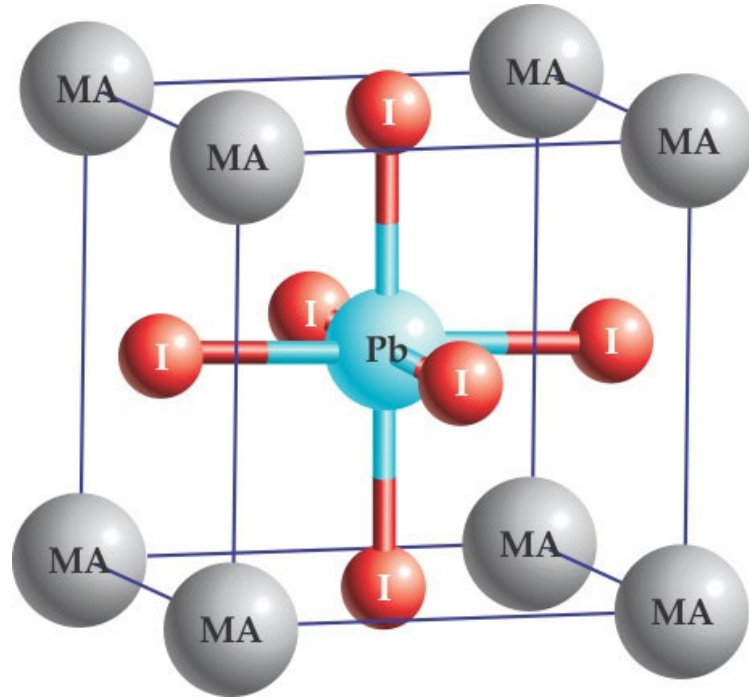


**New  
inorganic  
semi-  
conductors**



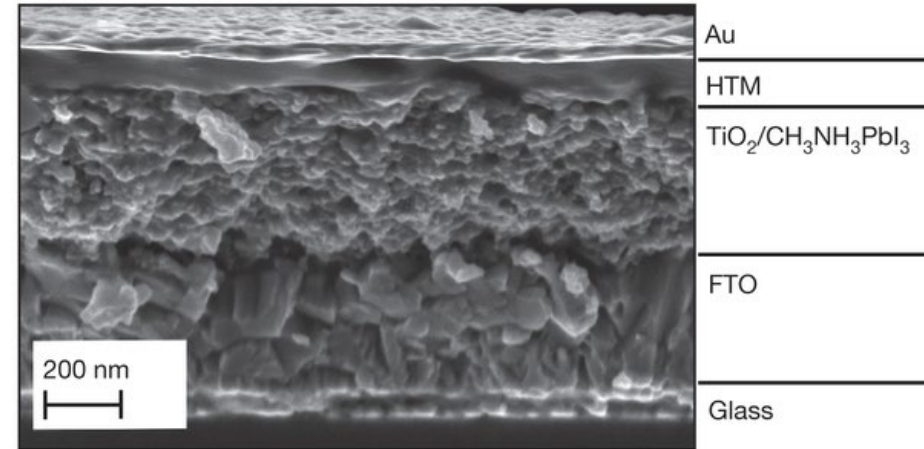
**Inorganic/or  
ganic hybrid  
perovskites**

# Organic-Inorganic Hybrid Perovskites



MA = methylammonium

<http://scitation.aip.org/docserver/fulltext/pt.5.7058figure1.jpg>



[http://www.nature.com/nature/journal/v499/n7458/images\\_article/nature12340-f2.jpg](http://www.nature.com/nature/journal/v499/n7458/images_article/nature12340-f2.jpg)

- Discovered just a few years ago
- Solution-processable → Inexpensive
- >20% already!!!
- Stability and lifetime still an issue